

1 Rheology of cellulose nanofibrils and silver nanowires
2 for the development of screen-printed antibacterial
3 surfaces

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19 **Abstract:** TEMPO (2,2,6,6-tetramethylpiperidine-N-oxyl)-oxidized cellulose nanofibrils (T-
20 CNF) and silver nanowires (Ag NWs) were formulated as active inks. Their rheological
21 properties were investigated to design optimal conditions for processing by the screen-printing
22 process, with the aim of preparing antibacterial patterns. Rheological experiments mimicking
23 the screen-printing process was applied to different ink formulations to investigate their
24 thixotropic and viscosity properties. The experiments conducted at 1% wt total mass content
25 and different ratios of T-CNF/Ag NWs, showed that the recovery (%), the recovery time and
26 the viscosity are formulation dependent. A ratio 2:1 (T-CNF:Ag NWs) and total mass content
27 of 2.5% wt was then selected to prepare an ink suitable for screen printing. Printing defects
28 were corrected by addition of water-soluble polymer hydroxypropyl methylcellulose (HPMC).
29 The selected formulation printed on flexible polyethylene terephthalate (PET) substrate
30 displayed a 67.4% antibacterial activity against E. Coli in a standard contact active test, with a
31 transparency superior to 70%, proving the promising features of the developed solution for
32 active packaging applications.

33 **Keywords:** cellulose nanofibrils; silver nanowires; rheology; screen-printing; antibacterial
34 activity.

35 **1. Introduction**

36 The packaging industry is facing new challenges in today's world and new innovations in
37 packaging solutions have emerged such as active and intelligent packaging. Active packaging
38 is usually defined as a technology able to respond to a stimulus either from the internal and
39 external atmosphere of the packaging, by either releasing or adsorbing active substances.
40 Several categories of active packaging can be distinguished such as gas scavengers,
41 antioxidants or antibacterial packaging [1]. Antibacterial packaging is a common way to fight
42 against bacterial contamination and so an interesting technique to enhance the shelf-life of
43 packaged product especially for food applications.

44 Different active agents have been used for antibacterial packaging applications such as essential
45 oils, organic acids, natural extract or nanoparticles. Silver nanoparticles (Ag NPs) demonstrate
46 a strong antibacterial activity: its mode of action and the influence of different parameters on
47 its activity have been thoroughly reviewed in the scientific literature [2]. Examples of Ag NPs
48 activity against bacterial contamination in real food products have also been reported, against
49 for instance cooked cereal or fruit [3, 4]. The use of Ag NPs or silver microparticles have been
50 studied deeply in the field of printed electronics due to their high electronic and thermal
51 conductivity properties. Notably the properties of silver nanoparticles with a high aspect ratio
52 and rod-like aspect called silver nanowires have been studied because of their capability to form
53 percolated networks [5]. Silver nanowires have also been studied for antibacterial applications
54 adaptable to packaging applications and especially with polymeric substrate or composite such
55 as polyethylene naphthalate (PEN) [6], polyvinyl alcohol (PVA) [7] or
56 polypropylene/polytetrafluoroethylene systems (PP/ePTFE) [8], but also bio-based solution
57 such as konjac glucomannan [9], chitosan [10] or PLA materials [11].

58 More recently, silver nanoparticles have been used in composites with cellulose and
59 nanocellulose materials [12]. Composites with cellulose nanofibrils (CNF) present several

60 advantages such as to entrap the silver nanoparticles within the CNF network, to stabilize the
61 Ag NPs and to control its dispersion as the produced systems are mainly prepared by in-situ Ag
62 NPs synthesis [13, 14].

63 Because of the inherent activity of antibacterial packaging, active materials are mainly directly
64 inserted inside or at the surface of the packaging film. For instance, using a coating or printing
65 process to functionalize the surface of polymeric materials is an interesting way to produce
66 antibacterial packaging by taking advantages of the versatility and flexibility of the deposition
67 processes. Printing processes enable the deposition of a specific pattern when compared to
68 surface deposition such as coatings. This localized deposition implies economy of materials
69 and so the development of cost-effective solutions. Printing processes also enable the
70 production of tailored solutions to a specific problem, and have been used in the field of active
71 and intelligent packaging for the development of smart detectors, such as gas and humidity
72 indicators/sensors and microbial sensors/indicators [15–18].

73 Only few references have reported the use of hybrid inks made of CNF and active materials for
74 deposition by printing techniques. For instance, Koga et al. (2013) reported inkjet printing of
75 TEMPO-oxidised CNF and carbon nanotubes for printed electronics applications [19]. Choi et
76 al. (2016) also inkjet-printed CNF as a pre-layer on paper to improve the deposition of
77 supercapacitors made of carbon nanotubes and a solid-state electrolyte. Another cellulosic
78 substrate was used by Nechyporchuk et al. (2017), who printed regular pigment and silver
79 nanoparticle conductive inks on cotton fabric [20]. Screen-printing was also used by El Baradai
80 et al. (2015) to print a mixture of microfibrillated cellulose, graphite and carboxymethyl
81 cellulose for the production of an electrode in lithium ion batteries [21]. Hoeng et al. (2017)
82 reported the use of TEMPO-oxidised cellulose nanofibrils (T-CNF) and silver nanowires (Ag
83 NWs) for printed electronics application [22]. The authors prepared inks fit for screen-printing
84 requirement using the thickening characteristic and the shear-thinning properties of the T-CNF.

85 The ink optimization was based solely on the ink viscosity and the authors showed the need to
86 add a water-soluble polymer named hydroxypropyl methylcellulose to modify the thixotropic
87 and viscoelastic properties of the ink. No in-depths characterisation of the impact of the T-
88 CNF/Ag NWs ratios and ink composition on the thixotropic behaviour of the inks were
89 performed, and the study was mainly a proof of concept focused only on electrical properties.
90 Cellulose nanofibrils in ink formulation are interesting due to their good properties as a
91 dispersing agent, rheology modifier or film forming capability. Other examples using printing
92 processes with cellulose nanofibrils mainly include 3D printing [23] or the use of nanocellulose
93 materials as a substrate [24].

94 Printing processes include a wide range of techniques, the most common used for packaging
95 applications being flexography, offset, gravure, screen and inkjet. Each process requires a
96 specific ink behaviour, that needs to be tailored to fit the process requirement. Screen-printing
97 is one of the most versatile printing techniques and consists of a squeegee pushing ink through
98 a patterned screen mesh with open areas. Screen-printing is usually considered as easy to use,
99 low cost and accepts a wide range of ink viscosity (around 1 to 200 Pa.s⁻¹), but does require
100 shear-thinning behaviour and precise viscoelastic properties. CNF materials then present
101 interesting rheological properties for such a process as they present strong shear thinning
102 behaviour attributed to alignment of particles under shear and to the destruction of local
103 entanglements (particle aggregates) [25]. This entangled network also promotes the thixotropic
104 behaviour of CNF materials and appropriate viscoelastic properties [26]. TEMPO-oxidised
105 CNF has met a strong interest since their development because of major energy savings during
106 the mechanical fibrillated process and the low dimensions and highly charged particles that are
107 produced [27]. The rheological properties of such materials are also deeply dependent on factors
108 such as pH, electrolytes, surfactants, co-solvents, temperature or concentration [28–32]

109 The main challenges linked to antibacterial materials made of nanocellulose and silver
110 nanoparticles are the materials preparation. In a large-volume and high standard application
111 such as in the packaging field, versatility, control and up-scaling of the preparation processes
112 are indeed key issues to address by the scientific community. Finally, the quality (homogeneity,
113 transparency, etc) and antibacterial efficiency of the active packaging materials are of course
114 major targets to achieve. This work is then dedicated on the investigation of the relation between
115 the properties of nanocellulose/silver nanoparticles inks (and especially TEMPO-oxidised
116 cellulose nanofibrils and silver nanowires) and their processability into antibacterial layers by
117 the screen-printing process. The screen-printing process was selected because of its high
118 versatility and flexibility. Several examples in the literature have reported the production of
119 composite made of nanocellulose and silver nanoparticles, but to our knowledge, this work is
120 the first example of printing a combination of cellulose nanofibrils and silver nanowires for the
121 development of antibacterial surfaces. Only Martins et al. (2012) used a coating process to
122 deposit a starch-based mixture of cellulose nanofibrils and silver nanoparticles on paper to
123 develop an antibacterial substrate [33]. An investigation of the rheological properties of inks
124 made of cellulose nanofibrils and silver nanoparticles have been reported by Hoeng et al. (2017)
125 who did not fully investigate the impact of the component's ratios and mass content on the
126 thixotropic properties of the suspensions [22].

127 In this work, the rheology properties of different formulations of TEMPO-oxidised cellulose
128 nanofibrils and silver nanowires were quantified and evaluated with respect to the potential for
129 use as ink in screen-printing. Different inks were first prepared at 1% wt total mass content and
130 different ratios of T-CNF/Ag NWs. The thixotropic and viscosity properties of the inks were
131 measured using a shear-induced rheological model, which mimics the screen-printing process.
132 Higher mass content inks, as well as the influence of the hydroxypropyl methylcellulose
133 (HPMC) additive, were also investigated. This optimization step led to the preparation of an

134 ink fit for screen-printing. Based on this, the influence of HPMC additives was assessed with
135 respect to the topography of the print, print quality and the antibacterial activity of the produced
136 pattern.

137 **2. Materials and Methods**

138 **2.1. Materials**

139 The materials comprise active inks as well as the substrate used for the preparation of the
140 antibacterial layers, but also the different chemicals used in the characterization tests.

141 The active inks were prepared using TEMPO oxidised cellulose nanofibrils (T-CNF) and silver
142 nanowires (Ag NWs). The aqueous suspension of T-CNF was provided by Betulium (Finland,
143 CNF-A13, 5% wt, $1.63 \pm 0.1 \text{ mmol.g}^{-1}$ measured charge density). The aqueous suspension of
144 silver nanowires (Ag NWs) was purchased from NanoGap (USA) referenced under the
145 tradename NGAP NF Ag-3170 (2.6% wt). The polyethylene terephthalate (PET) substrate used
146 for the printing test was provided by Dupont (France) (Melinex® ST726 – 175 μm). Deionized
147 water (DI water) was used for all experiments. Hydroxypropyl methylcellulose (HPMC) was
148 also used in some inks formulation and was purchased from ThermoFisher Scientific (US) with
149 the following specifications: 27.0-30.0% wt methoxy content and 7.0-12.0% wt
150 hydroxypropoxy content. Hydrochloric acid (>99%) and sodium hydroxide (>99%) were
151 acquired from Sigma Aldrich (US).

152 The following chemicals were used for the antibacterial tests: Escherichia Coli ATCC 8739
153 (Microbiologics, USA), L α -phosphatidyl choline (Sigma Aldrich, France, >99%), L-histidine
154 (Roth, Germany, >98.5%), Tween®80 (Roth, Germany), potassium dihydrogen phosphate
155 (Roth, Germany, >99%), sodium thiosulphate (Roth, Germany, >99%). The nutrient broth (NB,
156 Roth, Germany) presents the following composition: 15 g.l^{-1} peptone, 6 g.l^{-1} sodium chloride,
157 3 g.l^{-1} beef extract, 1 g.l^{-1} glucose. The plate counting agar (PCA, BD Difco, USA) presents the
158 following composition: 15 g.l^{-1} agar, 5 g.l^{-1} peptone and 3 g.l^{-1} beef extract. The materials were
159 used as received. Isotonic solution (IS) was prepared by dissolving calcium chloride in DI water

160 at 8.5 g.l⁻¹. The neutralising solution was prepared by dissolving the following chemicals in 1 l
161 of DI water: 3 g of L α -phosphatidyl choline, 1 g of L-histidine, 30 g of Tween®80, 10 g of
162 potassium dihydrogen phosphate, 4.10⁻⁴ g of sodium thiosulphate and the pH was controlled
163 between 7 and 7.2.

164 **2.2. Raw material characterization**

165 The work reported by Hoeng et al. (2017) showed how ultimately the T-CNF film provided a
166 matrix within which the Ag NW's were dispersed and suspended [22]. Furthermore, viscosity
167 of a particle laden fluid (such as T-CNF and Ag NWs suspensions) is influenced by the size
168 and shape of any particle content [34, 35] and consequently determination of T-CNF and Ag
169 NWs dimension is an important consideration.

170 To measure their morphological and dimensional parameters, the T-CNF were imaged by
171 Transmission Electronic Microscope (TEM) on a CM 200 equipment from FEI/Phillips (Japan)
172 with a TemCam F216 camera from TVIPS (Germany)[36]. To obtain distinct fibres, prior to
173 imaging, the T-CNF was diluted, drop-casted on a copper grid and dyed with uranyl acetate.
174 The acceleration voltage was set at 200 kV. The diameter and length of the T-CNF were
175 measured by analysing the TEM images using the FIJI software [36–38]. Due to coloration and
176 entanglement, measurements were complex to perform, however errors were managed through
177 a suitable sample size of nanofibrils: 10 nanofibrils were measured on each image and the
178 measurements were conducted on 10 different images. A representative example of the TEM
179 images is provided in **Fig. 1**. The average and associated standard deviation are presented. The
180 measured parameters for the T-CNF and the ones provided by the manufacturer for the Ag NWs
181 are summarized in **Table 1**.

182 **Table 1** T-CNF and Ag NWs morphological parameters either measured or supplied by the
 183 manufacturer

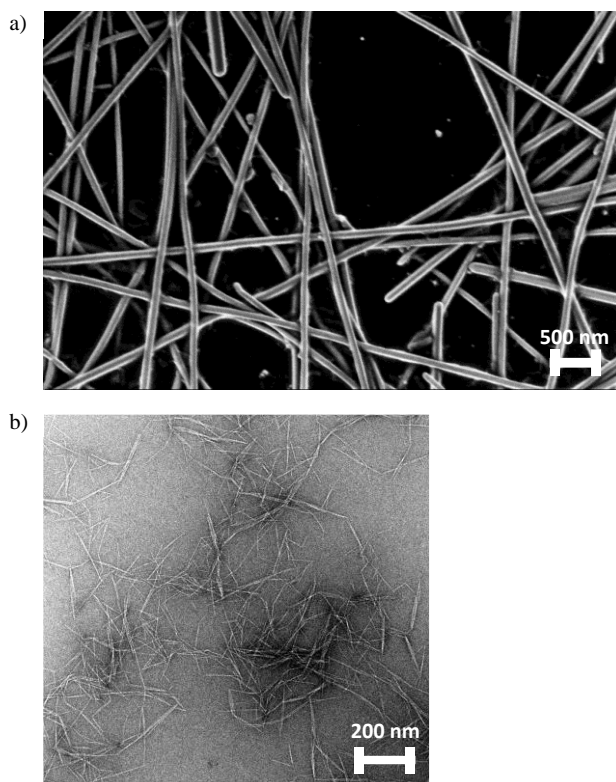
Materials	Length (nm)	Diameter (nm)	Average calculated aspect ratio	Oxidation charge ($\mu\text{mol.g}^{-1}$)
T-CNF	297 ± 133	8 ± 4	37	1630 ± 93
Ag NWs	$28\ 000 \pm 23\ 000$	72 ± 21	389	-

184 As the oxidation charge of T-CNF materials has a strong impact on the rheological parameters
 185 [39, 40], the charge density (mmol.g^{-1}) of the T-CNF was measured following a standard
 186 procedure. A suspension at approximately $250\ \text{mg.ml}^{-1}$ was prepared and the pH was set to
 187 $\text{pH} < 3.5$ with hydrochloric acid ($0.1\ \text{mol.l}^{-1}$). After 15 min of magnetic stirring, the mixture was
 188 titrated with sodium hydroxide ($0.01\ \text{mol.l}^{-1}$) by recording conductivity. Equivalent volumes
 189 were calculated by intersection of the linearly correlated curves: the first volume corresponds
 190 to the titration of the hydrochloric acid whereas the second corresponds to the titration of the
 191 T-CNF carboxylic acid function. Charge density (X) was calculated using the following
 192 equation (1):

$$193 \quad X (\text{mmol.g}^{-1}) = \frac{C_{\text{NaOH}} \times V_{2\text{eq}}}{m} \quad (1)$$

194 Where X (mmol.g^{-1}) is the charge density, C_{NaOH} (mmol.l^{-1}) is the sodium hydroxide
 195 concentration, $V_{2\text{eq}}$ (l) is the second equivalent volume corresponding to the carboxylic acid
 196 function titration volume and m (g) the dry mass of T-CNF titrated. Three titrations were
 197 conducted and the average is presented in **Table 1**.

198 The Ag NWs were imaged using a Field-Emission Gun Scanning Electron Microscope (FEG-
 199 SEM) apparatus (Ultra 55) from Zeiss (France). A diluted suspension of Ag NWs was drop-
 200 casted on a SEM stub and left to dry overnight. The Ag NWs were then imaged after
 201 Gold/Palladium coating (2 nm) and using a 3kV accelerating voltage and 5.4 mm working
 202 distance. A representative example of the images is provided in **Fig. 1**.



203

204 **Fig. 1** Electron microscopy of the raw materials with a) SEM-FEG of Ag NWs and b) TEM of
205 T-CNF

206 In order to achieve individual nanoparticles on a smooth surface, diluted suspensions of T-CNF
207 and Ag NWs were also drop-coated on Mica, left to dry overnight and then imaged by Atomic
208 Force Microscopy (AFM) using a Veeco NanoScope-V (Canada) microscope equipped with a
209 OTESPA Bruker (USA) silicon cantilever. The images are available in the supplementary
210 material (**Fig SI 1**).

211 To assess homogeneity, the turbidity of the T-CNF suspension (0.1% wt) was measured with a
212 portable turbidimeter. During the measurement, the scattered light at 90° angle is compared to
213 the incident light and the experiments were performed 6 times.

214 **2.3. Inks preparation**

215 All active inks prepared throughout this work were redispersed in water at the desired mass
216 content and ratio between T-CNF and Ag NWs using an Ultra Turrax high shear disperser for
217 30 s at 10 000 rotation per minute (RPM), that is based on laboratory good practice experience
218 (proper dispersion but no physical damage). Throughout this work, the ink concentration is
219 mass concentration (% wt) and the solvent was DI water. Several inks with T-CNF mass content
220 ranging from 0 to 1.66% wt and Ag NWs mass content ranging from 0 to 1% wt were prepared.

221 The inks based on T-CNF/Ag NWs (1.66% wt/0.83% wt) and HPMC additives (0, 3, 5 and 8%
222 wt) were prepared with the following protocol: the desired mass of T-CNF and Ag NWs were
223 weighed with the desired amount of water, and the desired mass of HPMC was added slowly at
224 60°C under several steps of high shear dispersion (30 s, 10 000 RPM) until the HPMC was
225 uniformly dissolved. To simplify the notation in the following section, the inks will be named
226 only after their HPMC content (0% HPMC, 3% HPMC, 5% HPMC or 8% HPMC).

227 **2.4. Rheology and inks properties**

228 All rheological tests were conducted on a MCR 302 rotational rheometer from Anton Paar
229 (Germany). All the tests were performed at 20°C with a set gap of 0.108 mm and using a cone
230 on plate geometry with the following characteristics: diameter of 50 mm, truncate of 55 μm and
231 angle of 1.013°. Rheological events occurring during the screen-printing process was simulated
232 using a protocol adapted from Neidert et al (2008) [41], where shear rates that are typical of
233 those experienced during different stages of the process are applied. During a deposition process
234 such as screen-printing, high shear is usually applied and the ink structure can thus be altered
235 (network modification, change in the solid-liquid interaction, etc), and this is particularly true
236 for inks composed of high aspect ratio nanoparticles. The thixotropic rheological component of

237 these inks needs to be determined to obtain the desired behaviour after deposition on the
238 substrate. The first step simulates the ink at rest on the screen by applying a 2 s^{-1} shear rate for
239 150 s (step 1). The second step simulates the squeegee passing over the screen and transferring
240 the ink to the substrate by applying a 1000 s^{-1} shear rate during 15 s (step 2). The last step
241 simulates the ink at rest on the substrate after printing by applying a 2 s^{-1} shear rate for 150 s
242 (step 3). The three-step rheological experiment was performed three times for the formulation
243 at 1% wt total mass and was found to be highly reproducible. For economy of materials, the
244 other formulations were then tested once.

245 Surface tension in conjunction with substrate specific surface energy is important in the
246 consideration of substrate wetting and hence the surface tension of the prepared inks was
247 evaluated using the Du Noüy ring method on a Sigma 700 tensiometer from Attension (UK)
248 with at least 20 measurements on each suspension.

249 **2.5. Screen-printing**

250 The screen-printing experiments were conducted on a Dek Horizon 03i printer (ASM Pacific
251 Technology, Singapore) with the following optimized parameters: 1 mm print gap, 120 mm.s^{-1}
252 forward speed, 80 mm.s^{-1} reverse speed, and 7 kg squeegee load. The polyester screen mesh
253 used presents the following specifications: $120 \text{ thread.cm}^{-1}$ mesh count, $34 \mu\text{m}$ thread diameter,
254 $12 \mu\text{m}$ emulsion thickness and 45° screen angle. The resulting prints were dried at 120°C for 1
255 min in an oven and each formulation was printed at least 3 times to assess the reproducibility
256 of the process. The quality of the print was quantified by scanning them in 8-bit grey scale on
257 a black background and performing image analysis using the FIJI software. The grey level
258 histogram distribution and the corresponding standard deviation could then be extracted using
259 the software. For a straightforward representation, the histogram distributions were smoothed
260 and drawn as curves.

261 **2.6. Print surface characterisation**

262 As parameters linked to the quality control of the deposition, the thickness (nm) and surface
263 roughness (μm) of the deposited layer was measured on a Veeco Wyco NT9300 white light
264 interferometer with the following parameters: 1.2×0.93 mm measured area, 736×480 pixels
265 resolution, $\times 5$ magnification. To measure the thickness, a specific focus was made on the edge
266 area of the printed layer. The height difference between the substrate and the ink layer was
267 measured by excluding the printing edge which can present irregularities. The average surface
268 roughness (S_a) was measured on the whole scanned area. At least 5 measurements were taken
269 on 3 different printed squares for each formulation.

270 The transparency of the printed pattern was investigated by measuring the transmittance at 550
271 nm wavelength on a UV-spectrophotometer (Shimadzu Manufacturing Inc., USA). At least 5
272 measurements were conducted on 3 different printed square for each sample.

273 **2.7. Antibacterial properties**

274 Prior to testing, the desired solutions, glassware and tools were sterilized in an autoclave for 20
275 min at $120^\circ\text{C}/1.034$ bar. The antibacterial properties of the screen-printed layers toward *E. Coli*
276 bacteria were evaluated by performing the AATCC TM100-1998 standard [42] test. To
277 summarize the technique, a bacterial suspension with a known concentration was deposited on
278 the surface of the samples and the concentration was measured again after 24 hours of contact
279 to assess the activity of the prints. For each sample, a $2\text{ cm} \times 2\text{ cm}$ surface of the prints was cut
280 and dry-sterilized at 60°C for 16 h in an oven prior to the experiments. *E. Coli* bacterial
281 suspension was diluted at 5.10^5 CFU. ml^{-1} in 20% NB (1 volume NB, 4 volume IS) and $200\ \mu\text{l}$
282 was drop-casted in microdroplets form on the surface of the sample. The systems were then
283 incubated at 37°C for 24 h. The samples were then recovered, washed with 50 ml of the

284 neutralizing solution and the remaining bacterial concentration was measured by a plate
285 counting method using plate counting agar. Antibacterial activity (AA, %) was calculated with
286 the following equation (2):

$$287 \quad AA (\%) = \frac{BC_{PET\ ref} - BC_{print}}{BC_{PET\ ref}} \times 100 \quad (2)$$

288 Where $BC_{PET\ ref}$ (log CFU) is the bacterial concentration on PET substrate after 24 h of
289 incubation, BC_{print} (Log CFU) the equivalent on the printed sample. 3 different samples were
290 tested for each formulation and the average is presented with associated standard deviation.

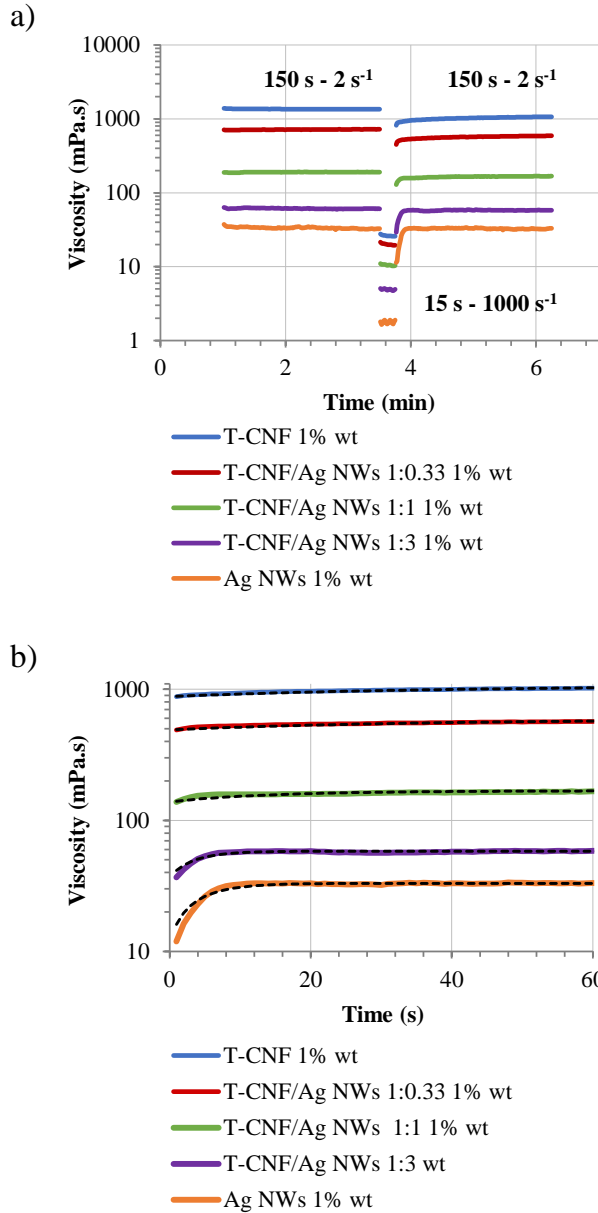
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292 **3. Results and discussions**

293 **3.1. Rheological properties of raw materials**

294 The morphological parameters of the supplied T-CNF were assessed by TEM microscopy.
295 Their length was measured at 297 ± 133 nm and their diameter at 8 ± 4 nm, which is relatively
296 small but similar to previous studies for T-CNF with high surface charge [35, 43]. The T-CNF
297 suspension is highly homogeneous as indicated by its turbidity measured at 18.9 ± 1.8 NTU.
298 The length and diameter of the Ag NWs are significantly higher, respectively 28 ± 23 μ m and
299 72 ± 21 nm, as supplied by the manufacturer. This difference in dimensions can be seen on the
300 TEM, SEM-FEG and AFM images (**Fig. 1** and **Fig. SI 1**). Different aqueous inks were prepared
301 using the raw materials and their rheological parameters were then investigated.

302 For these primary experiments, the total mass content was fixed at 1% wt and ratios between
303 the two components was varied between 0 and 1. The results are displayed in **Fig. 2** with the
304 overview of the three steps (**Fig. 2a**) and a zoom on the step 3 (**Fig. 2b**).). The latter includes
305 an associated curve that will be used to capture the recovery time consistently between samples.
306 The individual step 3 and associated fitting for each formulation can be found in the
307 supplementary material (**Fig. SI 2**).



308

309 **Fig. 2** 3-step rheological model of shear-induced deposition process for T-CNF/Ag NWs inks
 310 at 1% wt and ratios varying from 0 and 1 with a) overview of the viscosity vs time response and
 311 b) step 3 built-up curve and associate stretch exponential model (dotted lines)

312 Different parameters (recovery time, recovery, viscosity) were extracted from this experimental
 313 data and are summarized in **Table 2**. The viscosity (step 1, step 2 or step 3) is the viscosity
 314 measured at the end of the corresponding step. The recovery (%) is the ratio of the viscosity at
 315 the end of step 1 and at the end of step 3 as calculated from equation 3:

316
$$Recovery (\%) = 100 \times \frac{\eta_{end\ STEP\ 3}}{\eta_{end\ STEP\ 1}} \quad (3)$$

317 Where $\eta_{end\ STEP\ 1}$ (Pa.s) and $\eta_{end\ STEP\ 3}$ (Pa.s) are the viscosity at the end of steps 1 and 3
 318 respectively. The recovery time (s) was estimated by fitting the viscosity recovery in step 3 to
 319 the first order stretched exponential model from equation 4, as classically used to characterise
 320 the build-up recovery structure of thixotropic fluid, using the assumption that the recovery
 321 process is due to the re-orientation of the system during the transition steps [34, 44, 45].

322
$$\eta (Pa.s) = \eta_{\infty} + (\eta_0 - \eta_{\infty})(1 - e^{-\left(\frac{t}{\tau}\right)^r}) \quad (4)$$

323 Where η (Pa.s) is the viscosity, η_{∞} (Pa.s) the viscosity for an infinite shearing time, η_0 (Pa.s) the
 324 viscosity at the beginning of the shear step, t (s) is the time, τ (s) a time constant and r a constant
 325 (dimensionless). In the specific case of our study, the equation 4 was adapted to yield equation
 326 5:

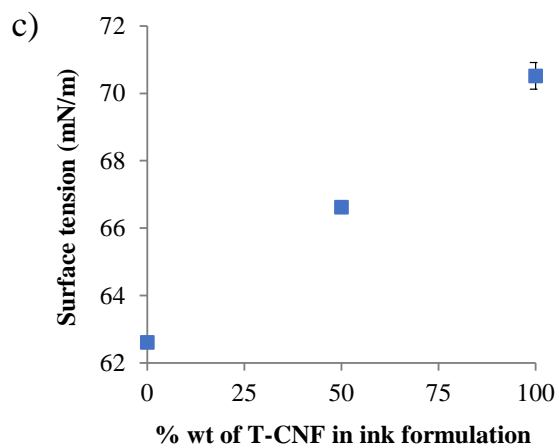
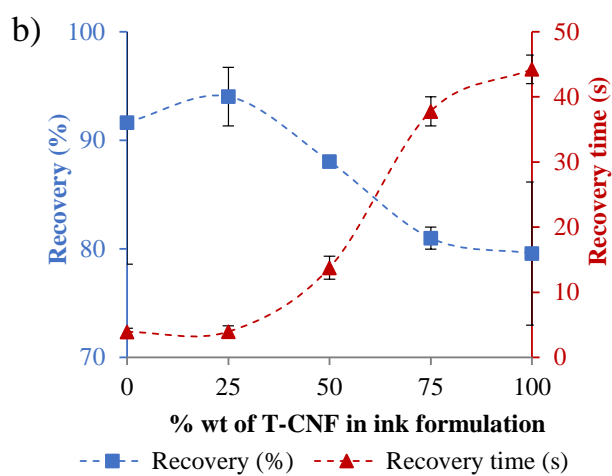
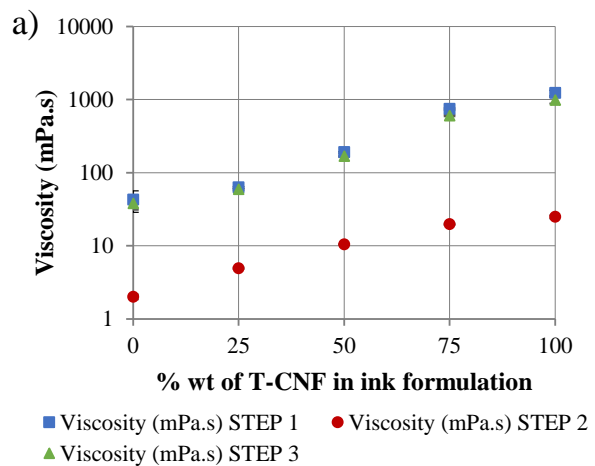
327
$$\eta (Pa.s) = \eta_{end\ STEP\ 3} + (\eta_{start\ STEP\ 3} - \eta_{end\ STEP\ 3})(1 - e^{-\left(\frac{t}{\tau}\right)}) \quad (5)$$

328 Where η (Pa.s) is the viscosity, $\eta_{end\ STEP\ 3}$ (Pa.s) the viscosity at the end of the step 3, η_{start}
 329 STEP 3 (Pa.s) the viscosity at the start of the step 3 (taken 1 s after the switch from step 2 to
 330 step 3, to limit inertia interferences), t (s) is the time and τ (s) is the recovery time. The r constant
 331 was assumed to be equal to 1.

332 **Table 2** Recovery time (s), recovery (%) and viscosity (mPa.s) extracted from the 3-step model,
 333 compared to the T-CNF content

	Ag NWs 1% wt	T-CNF/Ag NWs 1% wt 1:3	T-CNF/Ag NWs 1% wt 1:1	T-CNF/Ag NWs 1% wt 1:0.33	T-CNF 1% wt
% T-CNF	0	25	50	75	100
Recovery time (s)	3.9 ± 0.5	4.0 ± 0.9	13.8 ± 1.8	37.8 ± 2.2	44.2 ± 2.2
Recovery (%)	91.7 ± 13.1	94.0 ± 2.7	88.1 ± 0.4	81.0 ± 1.0	79.6 ± 6.6
Viscosity (mPa.s) step 1	43 ± 14	63 ± 2	190 ± 5	746 ± 20	1235 ± 103
Viscosity (mPa.s) step 2	2 ± 1	5 ± 1	10 ± 1	20 ± 1	25 ± 1
Viscosity (mPa.s) step 3	38 ± 7	59 ± 2	167 ± 5	604 ± 13	981 ± 99

334 In previous work, it was established that the rheological behaviour is predominantly dictated
 335 by the T-CNF in the T-CNF/Ag NWs inks [22, 46]. For this reason, the different parameters
 336 extracted were drawn against the proportion of T-CNF in the ink (**Fig. 3**).



337

338 **Fig. 3** Rheological parameters extracted from the 3-step model compared to the T-CNF content
 339 of the inks with a) the viscosity (mPa.s) at the end of each step, b) the recovery (%) and the
 340 recovery time (s) and other ink properties with c) measured surface tension ($\text{mN}\cdot\text{m}^{-1}$)

341 Firstly, and as expected, all the different inks prepared exhibit strong shear thinning behaviour
342 [25, 47] (**Fig. 3a**). This behaviour is essential for the formulated ink to properly flow ahead of
343 the squeegee and then through the mesh opening to cover the desired surface.

344 The recovery (%) is dependent on the T-CNF concentration: the higher the T-CNF
345 concentration the lower the recovery (**Fig. 3b**), which may be due to a permanent disruption of
346 the more entangled network. The recovery time was also impacted by the ink composition (**Fig.**
347 **3b**). The recovery time indeed increase with an increasing T-CNF content. It is below 10 s from
348 0 to 50% wt of T-CNF in the ink, and reaches 44.2 s for T-CNF 1%. The same plausible
349 explanation applies for both parameters: the higher amount of T-CNF, the more entangled and
350 denser the network, and so the permanent the disruption of the network is by the high shear
351 step. This proposed mechanism is represented schematically in **Fig. 4**. For the formulation at
352 low CNF content, the high shear step orients the T-CNF and Ag NWs yet does not significantly
353 impact the network structure because of a low entanglement factor. On the other hand, regarding
354 the formulation at high T-CNF content, not only does the high shear step align the nanoparticles
355 but could also break the pre-formed entangled network and disrupt any uncomplete percolation
356 pathway.

357 In an attempt to analyse this, the entanglement of the system could also be assessed by
358 comparing the concentration of raw materials in the different ink formulation to the estimated
359 critical percolation threshold where particulate contact occurs. By assuming a cellulose density
360 of 1.5 g.cm⁻³ and silver density of 10.5 g.cm⁻³ and using the estimated dimensions of the T-CNF
361 and Ag NWs the critical percolation threshold can be calculated using equation 6 [48, 49]:

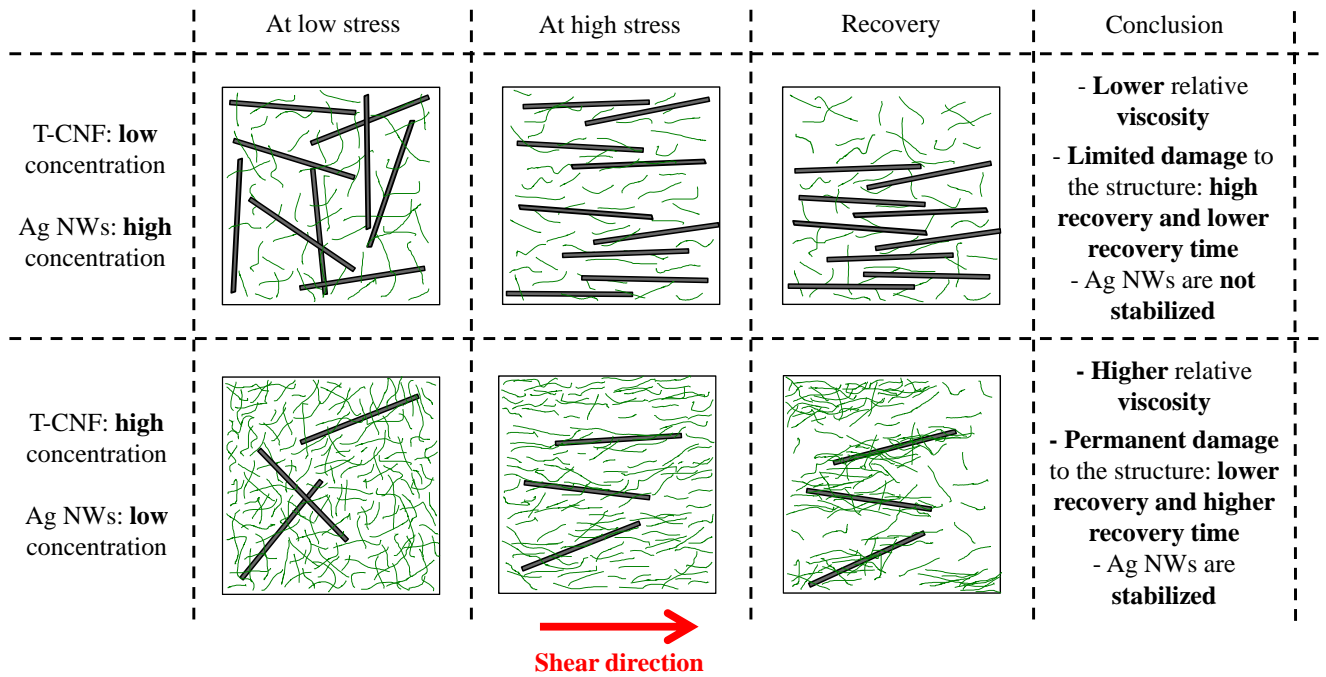
362
$$\varphi_c = 0.7 \frac{d}{L} \quad (6)$$

363 Where φ_c is the estimated critical percolation threshold, d and L are respectively the diameter
364 and length of the raw materials. The estimated critical percolation threshold was then compared

365 to the volume concentration for each of the raw materials in this ink (**Table 3**). This shows the
 366 fact that the systems are not fully percolated at the concentration and so consistent with the
 367 previous proposed mechanism. Furthermore, it is important to keep in mind that this estimation
 368 of the entanglement and crowdedness does not consider the effect of electrostatic repulsion,
 369 which is a critical parameter when working with highly charged T-CNF materials. Indeed, if
 370 highly charged CNF such as T-CNF are not expected to display major flocculation behaviour,
 371 the impact of high shear rate on the T-CNF network is still important [50]. High shear rate is
 372 expected to break down the gel structure and lead to the fibres alignment, deformation and
 373 gradient of fibres concentration at the microscale level [47, 51]. Beyond the T-CNF, the impact
 374 of the Ag NWs and their equilibrium with silver ions is unclear, but could also play a role in
 375 the charged structure of the network [25, 52]. In order to fully understand the complex hybrid
 376 systems made of T-CNF and Ag NWs, an interesting path to explore in the future could be to
 377 work on better modelling the networks formed, by considering the difference in
 378 flexibility/rigidity of the raw materials, as well as adding a rotational component to their
 379 orientation under shear [53, 54].

380 **Table 3** Estimation of the critical percolation threshold compared to the volume concentration
 381 (% vol) of T-CNF and Ag NWs in the ink formulations

	% wt	% vol	ϕ_c (% vol)
T-CNF	0.25	0.17	1.89
	0.5	0.33	
	0.75	0.50	
	1	0.67	
Ag NWs	0.25	0.02	0.18
	0.5	0.05	
	0.75	0.07	
	1	0.10	



382

383 **Fig. 4** Schematic representation (cross-section) of two separate formulation (high or low ratio
 384 of T-CNF/Ag NWs) under different situation (low or high shear stress, recovery) and impact on
 385 its network

386 For screen printing, the optimum target for the recovery would be close to 100% to allow a
 387 maximum print resolution or stabilisation of a coated film. The time necessary for the ink to
 388 recover its viscosity is also an important parameter. The ideal situation would be to have a
 389 recovery fast enough to achieve a homogenous levelling (> ca. 10 s) but not too long to avoid
 390 extensive slumping (< ca. 200 s) [22, 55, 56]. All the different formulations display a high
 391 enough recovery (%) for printing with high resolution. However, in the 0 to 50% of T-CNF
 392 range, the recovery times are small (< 10 s) and these characteristic times could be too small
 393 for an even levelling of the ink on the substrate to provide a smooth surface [22, 55]. This
 394 suggests that the adequate rheological conditions are a higher mass content ink with a T-CNF
 395 content higher than 50% wt. The increasing recovery times with the increasing T-CNF content
 396 is promising because a higher mass content ink is usually used in screen-printing applications.
 397 Classically, in the field of functional printing and coating, another parameter to consider for an

398 optimal deposition is the substrate/ink interaction. It was found out that the inks prepared (**Fig.**
399 **3c**) presents a relatively high surface tension (ranging from 62 to 71 mN.m⁻¹) compared to the
400 surface energy of the PET substrate (*ca.* 40 mN.m⁻¹). Although this study is aimed at the
401 rheological behaviour of the active inks composed of T-CNF and Ag NWs, this energy
402 difference between inks and substrate could lead to unsatisfactory substrate/ink interaction and
403 especially poor wetting and spreading of the ink on the substrate, and so potential dewetting
404 printing defects.

405 As discussed above, T-CNF has a strong impact on viscosity, and in order to fit more closely
406 with the screen-printing process requirements, the concentration of T-CNF was fixed at 1.66%
407 wt, and a total mass of 2.5 %wt with a 2:1 ratio was selected. Previous work on similar systems
408 has shown the positive impact of the addition of water-soluble polymer such as hydroxypropyl
409 methylcellulose (HPMC) [22], mainly on the recovery time and the viscoelastic properties of
410 the formulated inks. So, different amount (0, 3, 5 and 8% wt) were added in the selected T-
411 CNF/Ag NWs blend.

412 The same three step rheological model was applied to the HPMC formulations, and the same
413 parameters linked to viscosity and recovery were extracted (**Table 4**). The viscosity *versus* time
414 graphs can be found in the electronic supplementary material (**Fig. SI 3** and **Fig. SI 4**). The
415 reference 0% HPMC (=T-CNF/Ag NWs 2.5% wt 2:1) displays a recovery (72.4%) slightly
416 higher than the T-CNF 1.66% wt formulation (64.7%) showing that the addition of Ag NWs
417 also impacts the recovery (%), whereas the recovery time was relatively similar for both
418 formulations (respectively 39.6 s for 0% HPMC and 37.3 s for T-CNF 1.66% wt).

419 The viscosity drastically increases with the concentration of HPMC, going from 8.2 Pa.s for
420 HPMC 0% to 57.4 Pa.s for HPMC 8%, due to the thickening effect of the water-soluble
421 cellulose derivative [22]. With an increasing amount of HPMC, the recovery (%) drops from

422 72.4% for HPMC 0% to 59.2% for HPMC 3%. It is not too much impacted for HPMC 5% and
 423 HPMC 8% (respectively 59.1 and 65.5%). The recovery time however decreases with an
 424 increasing content of HPMC, reaching 13.6 s for 8% HPMC compared to 29.2 for 5% HPMC
 425 or 33.6 for 3% HPMC.

426 The recovery parameters (time and %) are promising for 0% HPMC but its viscosity is in the
 427 low range of what is classically expected for screen-printing inks. The 5% HPMC formulation
 428 is selected to be a good compromise between recovery and viscosity parameters.

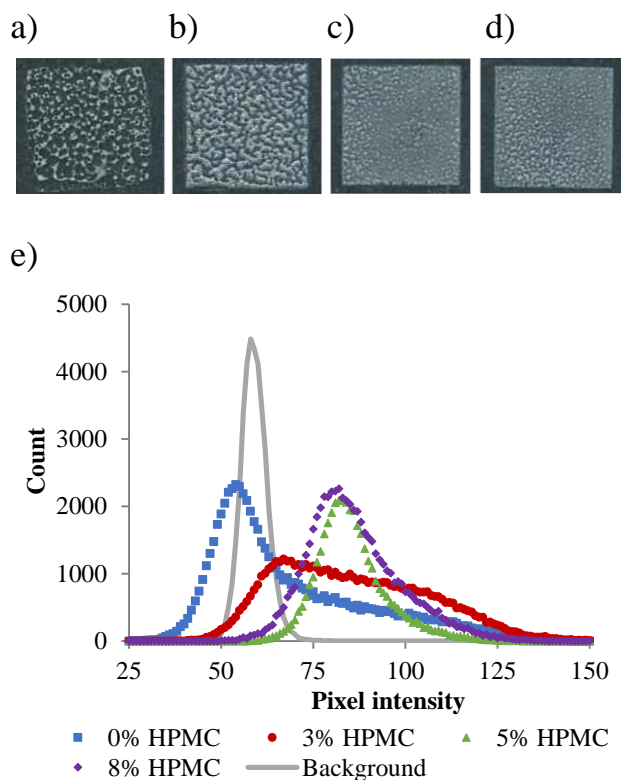
429 **Table 4** Values of recovery time (s), recovery (%) and viscosity (mPa.s) extracted from the 3-
 430 step model for the formulations with different HPMC content compared to T-CNF 1.66% wt

	T-CNF 1.66%	0% HPMC	3% HPMC	5% HPMC	8% HPMC
Recovery time (s)	40.4	39.6	33.6	29.2	13.6
Recovery (%)	64.7	72.4	59.2	59.1	65.5
Viscosity (mPa.s) step 1	8691	8209	17229	29341	57426
Viscosity (mPa.s) step 2	10	72	293	795	1057
Viscosity (mPa.s) step 3	5627	5945	10193	17344	37607

431 **3.2. Additive use and impact on print quality**

432 The formulation with different content of HPMC were printed and the resulting patterns can be
 433 found in **Fig. 5a, b, c** and **d**. The print quality is greatly enhanced visually with the addition of
 434 5% wt or 8% wt of HPMC. This is mostly consistent with the previous rheological parameters
 435 extracted from the experiments but not entirely, showing that the usual rheological test can only
 436 give a general guidance for provision of the print quality.

437 For a quantitative comparison, the quality of the print was estimated by image analysis. Adding
438 HPMC reduces the statistical dispersion of the grey histogram and the graphs are shifted from
439 the black background to a higher intensity meaning a better coverage of the surface and less
440 mottling defect that makes the substrate appear under the print (**Fig. 5e**). There is not much
441 difference between the prints at 5 and 8% of HPMC and this was confirmed by the standard
442 deviation of the grey histogram distribution (**Table 5**).



443
444 **Fig. 5** Quality assessment of the printed square on PET substrate with T-CNF/Ag NWs 2.5%
445 wt 2:1 and different amount of HPMC with visual pictures for a) 0% wt, b) 3% wt, c) 5% wt
446 and d) 8% wt and grey histogram distribution for the different formulation printed and
447 comparison with the black background

448 The topography of the print was also assessed by white light interferometry by measuring the
449 thickness and surface roughness of the prints (**Table 5**). For the reference formulation without
450 any HPMC (0% HPMC), the quality of the print was not sufficient to measure reliable

451 topological parameters. This could be due to poor wetting dynamics which seems to be
452 enhanced by the addition of HPMC. For all other samples, the quality was deemed to be
453 sufficient for reliable measurements. However, the issue of homogeneous spreading and the
454 presence of defects even for high content of HPMC is reflected in the high standard deviation
455 obtained for thickness and roughness measurements. For 3% HPMC, 5% HPMC and 8%
456 HPMC, the measured dry thickness is less than 1 μm which might be surprising for screen
457 printing but consistent with the very low mass content of the ink compared to conventional
458 screen-printing inks. The measured thicknesses are increasing with an increasing content of
459 HPMC and range from 269 to 811 nm. For all samples the surface roughness is relatively similar
460 for each sample. Due to the print quality, the transparency measurements were not reliable for
461 the 0% HPMC. The pattern transmittance was measured at 75.2, 73.3 and 68.0%, respectively
462 for 3% HPMC, 5% HPMC and 8% HPMC. There is a significant decrease in the transparency
463 after printing, when compared to the PET substrate value, measured at $89.8 \pm 1.4\%$. However,
464 the final values reached are still relatively high and not very far from packaging targets, where
465 a transparency value of 80% is an accepted guideline. The materials produced in this study are
466 then considered as transparent enough for the developed solution to be used in active packaging
467 application.

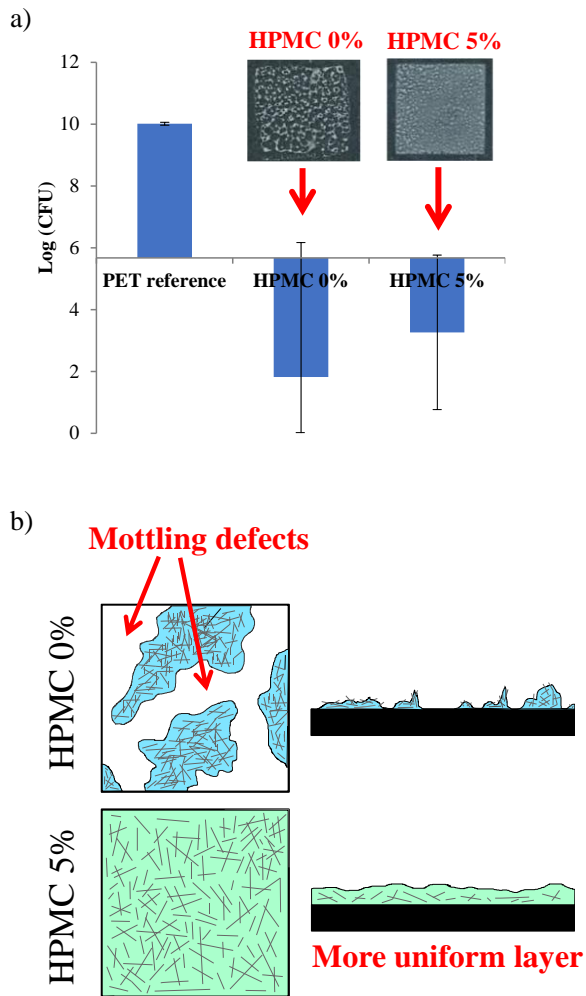
468 **Table 5** Characterization of the prints as function of the HPMC content in the ink with grey
 469 histogram standard deviation, thickness (nm), surface roughness (nm) and transparency (%)
 470 (N/A= not applicable)

HPMC content (%)	0	3	5	8
Grey histogram standard deviation	20.8	20.2	9.6	12.7
Thickness (nm)	N/A	269 ± 138	622 ± 140	811 ± 276
Surface Roughness (<i>Sa</i> , nm)	N/A	450 ± 106	482 ± 134	549 ± 137
Transparency (T550nm, %)	N/A	75.2 ± 1.1	73.3 ± 0.9	68.0 ± 1.9

471 Finally, the HPMC 0% and HPMC 8% prints were tested against E. Coli bacterial
 472 contamination using the AATCC TM100-1998 standard designed for active surfaces (**Fig. 6a**).
 473 This test is adapted for contact active surface and is performed by drop-casting a bacterial
 474 suspension (know volume and known concentration) on top of the tested surface. After 24 hours
 475 of contact in an incubator (37°C, 100 RPM), the bacterial suspension is recovered, numbered
 476 and compared with the starting concentration. On the **Fig. 6a**, the horizontal line represents the
 477 starting concentration and the histograms represent the final concentration of the bacterial
 478 suspension after contact with either the PET reference film (untreated) and printed samples
 479 HPMC 0% and HPMC 5%. The antibacterial activity was calculated comparing the bacterial
 480 concentration at the end of the test, between the PET reference and the different samples.

481 Despite the low thickness of the deposited layers, the prints present a bactericidal activity
 482 toward E. Coli with a calculated antibacterial activity of 81.8% and 67.4% for HPMC 0% and
 483 HPMC 5% respectively. As previously described, the print quality is greatly enhanced from
 484 HPMC 0 to HPMC 5% and only a small loss in the antibacterial activity is registered between
 485 the two formulations, which probably corresponds to a deeper entrapment of the Ag NWs due
 486 to the HPMC addition and consequent lower accessibility and lower silver ions release (**Fig.**
 487 **6b**). This is a significant result that proves the versatility of the formulation that can be adapted

488 to fit the process requirements, even at a large concentration of additives (5% wt) compared to
 489 the concentration of the antibacterial silver nanowires (0.84% wt). The high standard deviation
 490 of **Fig. 6a** can be attributed to the heterogenous surface due to the printing defects.



491

492 **Fig. 6 a)** Qualitative analysis of the antibacterial activity of PET reference, HPMC 0% and
 493 HPMC 5% samples, using AATCC TM100-1998 standard vs *E. Coli* with results expressed in
 494 log of colony forming units (log CFU) and b) schematic representation of HPMC 0% and
 495 HPMC 5% and the benefits of adding HPMC into the ink formulation

496 **4. Conclusion**

497 The rheological properties of ink formulations based on TEMPO-oxidised cellulose nanofibrils
498 and silver nanowires were investigated in a systematic step-by-step manner. The viscosity and
499 thixotropic behaviour of low-solid content inks were found to be highly dependent on the
500 formulation and higher content screen-printing inks were then prepared based on the optimal
501 formulation. Hydroxypropyl methylcellulose was included as an additive which improved the
502 print quality and the rheological properties of the mixture were investigated in a similar manner.
503 The quality of the print (image analysis, surface roughness and thickness) were correlated to
504 the rheological findings and the antibacterial activity of the prints was established, which proves
505 the potential application to use such a formulation for patterned antibacterial packaging
506 applications. The possibility to introduce localized antibacterial pattern is interesting for all
507 sorts of applications (food or medical packaging lids and seals for example) and promotes
508 materials economy as well as easily tailored antibacterial solution when compared to full
509 surface deposition.

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515 imaging.

516

517 **Conflicts of interest**

518 There are no conflicts to declare.

519

520 **Electronic Supplementary Information**

521 The AFM images, as well as the details of the 3-step rheological model (viscosity *versus* time
522 graphs) applied to the different formulation can be found in the Electronic Supplementary
523 Information.

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