

1 **Diffusive CH<sub>4</sub> fluxes from aquaculture ponds using floating**  
2 **chambers and thin boundary layer equations**

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32 **HIGHLIGHTS**

33 ● Aquaculture ponds emit CH<sub>4</sub>

34 ● Large variations in diffusive CH<sub>4</sub> fluxes are estimated by different thin boundary  
35 layer (TBL) models

36 ● Methane fluxes measured by chambers and match those estimated by only some  
37 TBL models

38 **ABSTRACT**

39 Static floating chambers (*FCs*) are the conventional method to measure CH<sub>4</sub> fluxes  
40 across the water-air interface in ponds, while thin boundary layer (*TBL*) modelling is  
41 increasingly used to estimate CH<sub>4</sub> fluxes. In this study, both *FCs* measurements and  
42 *TBL* models of gas transfer velocity were used to determine CH<sub>4</sub> evasion from  
43 aquaculture ponds in southeastern China. The surface water CH<sub>4</sub> concentrations  
44 ranged from 0.4 to 9.1 μmol L<sup>-1</sup> with an average of 4.8±0.8 μmol L<sup>-1</sup>. CH<sub>4</sub> flux was  
45 always positive, indicating the ponds as a persistent CH<sub>4</sub> source to air. Mean CH<sub>4</sub> flux  
46 based on different *TBL* models showed large variations, ranging between 19 and 316  
47 μmol m<sup>-2</sup> h<sup>-1</sup>. Compared against the direct measurement *FCs*, three *TBL* models  
48 developed for the open sea, flowing estuarine system and lentic ecosystem (*TBL*<sub>W92a</sub>,  
49 *TBL*<sub>RC01</sub>, and *TBL*<sub>CL98</sub>, respectively) overestimated CH<sub>4</sub> emission by 40-200%, while  
50 the wind tunnel-based *TBL* model (*TBL*<sub>LM86</sub>) underestimated CH<sub>4</sub> emission. Two *TBL*  
51 models developed for lakes (*TBL*<sub>W92b</sub> and *TBL*<sub>CW03</sub>) gave estimates similar to *FCs*.

52 **Keywords:** Methane fluxes; Thin boundary layer models; Floating chambers;  
53 Water-air interface; Shallow aquaculture pond; Subtropical estuary

## 54 **1. Introduction**

55 Methane (CH<sub>4</sub>) emissions from inland and coastal aquatic systems are potentially  
56 significant sources of atmospheric CH<sub>4</sub> (Bastviken et al., 2011; Musenze et al., 2014;  
57 Yang et al., 2011). CH<sub>4</sub> release from open water can be via diffusion and/or ebullition  
58 (bubbling) (Bastviken et al., 2004). Diffusive fluxes across the water-air interface are  
59 usually determined by using static floating chambers (*FCs*) or thin boundary layer  
60 (*TBL*) models. The *FCs* approach determines CH<sub>4</sub> fluxes based on the change in CH<sub>4</sub>  
61 concentrations in the chamber headspace over time. The *TBL* approach calculates the  
62 CH<sub>4</sub> flux from piston velocity and gas concentration in the water (Natchimuthu et al.,  
63 2017; Zhao et al., 2019). Previous studies have used either one of the two approaches  
64 to quantify CH<sub>4</sub> fluxes from aquatic ecosystems (e.g., Musenze et al., 2014;  
65 Natchimuthu et al., 2016; Wang et al., 2017; Welti et al., 2017). However, detailed  
66 comparison of the two methods is rare (e.g., Duchemin et al., 1999; Matthews et al.,  
67 2003), particularly for small pond ecosystems.

68 Recent studies have shown that very small ponds (area <0.001 km<sup>2</sup>) are hotspots  
69 for CH<sub>4</sub> emission (Holgerson, 2015; Holgerson and Raymond, 2016; Wik et al., 2016;  
70 Yuan et al., 2019). However, the scalability of these measurements are largely  
71 constrained by the lack of rigorous quantifications of the area, number, and spatial  
72 distribution of small ponds globally (Jonsson et al., 2008; Zhao et al., 2019) and the  
73 different flux measurement methods between studies. In particular, the lack of  
74 consensus on gas flux measurement methods remains a major source of uncertainty in  
75 greenhouse gas assessment. For instance, the *TBL*<sub>LM86</sub>, *TBL*<sub>Wan92a</sub> and *TBL*<sub>Wan92b</sub>,

76 *TBL*<sub>RC01</sub>, *TBL*<sub>CL98</sub>, and *TBL*<sub>CW03</sub> models, which were developed by [Liss and Merlivatt](#)  
77 (1986), [Wanninkhof \(1992\)](#), [Raymond and Cole \(2001\)](#), [Cole and Caraco \(1998\)](#), and  
78 [Crusius and Wanninkhof \(2003\)](#), respectively, are widely adopted wind-based models  
79 to estimate CH<sub>4</sub> transfer velocities and fluxes. Among these *TBL* models, the  
80 *TBL*<sub>LM86</sub>, *TBL*<sub>Wan92a</sub>, and *TBL*<sub>RC01</sub> models were developed for wind tunnels, open sea,  
81 and flowing estuarine systems, respectively, while *TBL*<sub>Wan92b</sub>, *TBL*<sub>CL98</sub> and *TBL*<sub>CW03</sub>  
82 models were developed for the lentic ecosystem (e.g., lake). It is unclear to what  
83 extent these different models are transferable to other aquatic ecosystems ([Musenze et](#)  
84 [al., 2014](#)), and there is also a paucity of study comparing CH<sub>4</sub> fluxes by the different  
85 approaches.

86 Aquaculture ponds are an important component of the global inland aquatic  
87 habitats ([FAO, 2017](#)), and the total surface area of freshwater and brackish  
88 aquaculture ponds is estimated to be around 110,000 km<sup>2</sup> ([Verdegem and Bosma,](#)  
89 [2009](#)). Despite the importance of aquaculture ponds for CH<sub>4</sub> emission ([Hu et al., 2016;](#)  
90 [Wu et al., 2018;](#) [Yang et al., 2015, 2019a;](#) [Yuan et al., 2019](#)), relevant CH<sub>4</sub> flux data  
91 are disproportionately scarce, and the published results were predominantly  
92 determined by *FCs* rather than *TBL* modelling ([Hu et al., 2016;](#) [Wu et al., 2018;](#) [Yang](#)  
93 [et al., 2015, 2019a](#)). In this study, *FCs* and *TBL* models were used to compare CH<sub>4</sub>  
94 fluxes in aquaculture ponds in southeastern China. The aims were: (1) to evaluate the  
95 performances of different wind-based *TBL* models for estimating CH<sub>4</sub> fluxes; (2) to  
96 compare the diffusive CH<sub>4</sub> emissions from aquaculture ponds derived from *FCs*  
97 measurements and *TBL* modellings; and (3) to assess which *TBL* model(s) can be used

98 to replace *FCs* for estimating CH<sub>4</sub> fluxes from ponds, with acceptable validity.

## 99 **2. Materials and Methods**

### 100 *2.1. Study area*

101 Our study sites are located at the central-western Shanyutan Wetlands in the Min  
102 River Estuary (MRE) in southeastern China (Figure S1, 26°00'36"–26°03'42" N,  
103 119°34'12"–119°40'40" E). This area is characterized by a subtropical monsoon  
104 climate, with a multi-year annual average temperature and precipitation of 19.6 °C  
105 and 1,350 mm, respectively (Tong et al., 2010). The wetlands are dominated by a  
106 semidiurnal tide with a large tidal range (2.5-6 m) that follows a spring-neap-spring  
107 tidal cycle (Luo et al., 2014; Tong et al., 2010). The dominant vegetation are the  
108 native *Cyperus malaccensis* and *Phragmites australis*, and the invasive *Spartina*  
109 *alterniflora*. Over the past 10 years, much of area has been converted to aquacultural  
110 ponds (Yang et al., 2017a).

### 111 *2.2. Aquaculture pond management*

112 Small and shallow aquaculture earthen ponds (area of 0.8–2.5 ha and depth of  
113 1.1–1.8 m) are a key feature in the MRE, covering a total area of around 234 ha in the  
114 Shanyutan Wetland (Yang et al., 2017b). Semi-intensive production is concentrated  
115 between June and November, which yields a single annual crop of shrimps. The ponds  
116 are filled with brackish water (average salinity of 2.0– 8.5‰) from the MRE. The  
117 shrimps are fed twice daily (at 07:00 and 16:00 hr) with commercial aquatic feed  
118 pellets containing 42% protein. Three to five paddlewheel aerators operate four times

119 a day (07:00–09:00, 12:00–14:00, 18:00–20:00, and 00:00–03:00 hr) to provide  
120 oxygen. For this study, three ponds separated by <10 m (see [Table S1](#) for basic  
121 characteristics) ([Zhang et al., 2019](#)) were selected for the measurements. Additional  
122 details about the shrimp pond system and management can be found in [Yang et al.](#)  
123 ([2017b](#)).

### 124 *2.3. Determination of dissolved CH<sub>4</sub> concentration*

125 Field campaigns were carried out at the three ponds between June and November  
126 2017 following the main aquaculture operation. In each pond, water and gas samples  
127 were collected at three sites along a foot-bridge that extended ~10 m from the  
128 embankment to the pond center. Samples were collected two or three times each  
129 month in each pond for a total of 15 times. The total number of samples was 3 ponds  
130 × 3 sites × 15 times = 135. To measure the dissolved CH<sub>4</sub> concentrations, surface  
131 water (at a depth of ~20 cm) was collected by a homemade water sampler and  
132 transferred into 55-mL gas-tight glass serum bottles that had been flushed with pond  
133 water 2-3 times. A 0.2 mL aliquot of saturated HgCl<sub>2</sub> solution was added to each  
134 bottle to inhibit bacterial activity of water sample ([Borges et al., 2018](#); [Hu et al., 2018](#)),  
135 and the bottle was immediately sealed with a butyl rubber stopper and an aluminum  
136 screw cap to exclude air bubbles. Sample bottles were transported back to the  
137 laboratory in an ice-packed cooler. Dissolved CH<sub>4</sub> concentrations were measured  
138 within 2 d of collection using the headspace equilibration method: Approximately 25  
139 mL of water in each bottle was displaced by N<sub>2</sub> gas (>99.999% purity) to create  
140 headspace. The bottle was then shaken vigorously for 20 min and left at room



141 temperature for 30 min to attain equilibrium between the air and the water phases  
142 (Cotovicz et al., 2016). Afterward, approximately 10 mL of the headspace was  
143 extracted and injected into a gas chromatograph (GC-2010, Shimadzu, Kyoto, Japan)  
144 equipped with a flame ionization detector (FID) to determine the CH<sub>4</sub> concentration.  
145 Standard CH<sub>4</sub> gas (at 2, 8, 500, 1000, and 10,000 ppm) was used to calibrate the FID.  
146 Dissolved CH<sub>4</sub> concentration was calculated based on the volume of water, headspace  
147 air and gas solubility coefficient for the specific water temperature and salinity (Farias  
148 et al., 2017; Wanninkhof, 1992; Xiao et al., 2017).

#### 149 *2.4. Determination of diffusive CH<sub>4</sub> flux across the water-air interface*

##### 150 *2.4.1. Measurement using floating chambers*

151 This study used a modified chamber placed on a floating buoy (Figure S2). The  
152 opaque floating chamber was made from inverted plastic basin  
153 (polyethylene/plexiglas®) with a volume and area of 5.2 L and 0.1 m<sup>2</sup>, respectively.  
154 The chamber was covered with aluminum tape to minimize internal heating by  
155 sunlight (Natchimuthu et al., 2016; Yang et al., 2019). A thin gauze (pore diameter  
156 0.001 mm) covering the opening minimized the entry of bubbles into the chamber  
157 (Figure S2). A fan inside the chamber mixed the headspace air during the sampling. In  
158 order to quantify the potential contribution of CH<sub>4</sub> ebullition flux from the ponds,  
159 total CH<sub>4</sub> fluxes were also determined using floating chamber without gauze.

160 The chamber was deployed for a period of 45 min and headspace air samples  
161 being extracted at 15-min intervals (0, 15, 30, 45 min) using 60-mL syringes equipped

162 with three-way stopcocks. The gas samples were immediately transferred into  
163 pre-evacuated airtight gas sampling bags (Dalian Delin Gas Packing Co., Ltd., China),  
164 transported to the laboratory, and analyzed within 48 h using a gas chromatograph  
165 (GC-2010, Shimadzu, Kyoto, Japan) equipped with a FID, following the method of  
166 [Tong et al. \(2010\)](#). The detection limit for CH<sub>4</sub> was 0.3 ppm, and the relative standard  
167 deviations of the measurements were  $\leq 2.0\%$  in 24 h.

168 CH<sub>4</sub> emission flux ( $F_{\text{CH}_4}$ ,  $\mu\text{mol m}^{-2} \text{hr}^{-1}$ ) was calculated from the slope of the  
169 regression between headspace CH<sub>4</sub> concentration and time ([Yang et al., 2019](#)).  
170 Generally, if  $r^2$  of the regression is  $> 0.90$ , the CH<sub>4</sub> emission is considered as diffusion  
171 only ([Bastviken et al., 2010](#); [Zhu et al., 2016](#)). If  $r^2$  is  $< 0.90$ , the emission is  
172 considered as a combination of ebullition and diffusion. The floating chambers with  
173 gauze (FCs-G) and without gauze (FCs-NG) showed distinctly linear ( $r^2 > 0.9$ ) and  
174 nonlinear ( $r^2 < 0.9$ ) increases in methane concentration, respectively; therefore, the  
175 contribution of ebullition could be calculated as the difference between the FCs-G and  
176 the FCs-NG measurements.

#### 177 2.4.2. Estimation using thin boundary layer models

178 Saturation ( $S$ ) of CH<sub>4</sub> in pond water was calculated as ([Hu et al., 2018](#)):

$$179 S = C_{\text{water}}/C_{W_s} = C_{\text{water}}/(\alpha \times C_{\text{air}}) \times 100\% \quad (\text{Eq. 1})$$

180 where  $C_{\text{water}}$  is dissolved CH<sub>4</sub> concentration in pond water;  $C_{W_s}$  is the saturated CH<sub>4</sub>  
181 concentration ( $\mu\text{mol L}^{-1}$ );  $C_{\text{air}}$  is the atmospheric CH<sub>4</sub> concentration ( $\mu\text{mol mol}^{-1}$ ) at  
182 the sampling site; and  $\alpha$  is the Bunsen coefficient ([Wanninkhof, 1992](#)).

183 Diffusive flux of CH<sub>4</sub> ( $F$ ,  $\mu\text{mol m}^{-2} \text{hr}^{-1}$ ) across the water-air interface can be

184 described by a theoretical diffusion model ([Musenze et al., 2014](#)):

$$185 \quad F = k \times (C_{\text{water}} - C_{\text{eq}}) \quad (\text{Eq. 2})$$

186 where  $C_{\text{water}}$  ( $\mu\text{mol L}^{-1}$ ) is the measured dissolved  $\text{CH}_4$  concentration in surface water,

187  $C_{\text{eq}}$  ( $\mu\text{mol L}^{-1}$ ) is the dissolved  $\text{CH}_4$  concentration in equilibrium with the air above,

188 and  $k$  is the gas transfer velocity ( $\text{cm h}^{-1}$ ). The  $k$  value was parameterized as a function

189 of wind speed and normalized for surface water temperature ( $T$ ,  $^{\circ}\text{C}$ ) using a Schmidt

190 number ( $Sc$ ) derived from Eq. 3 ([Wanninkhof, 1992](#)):

$$191 \quad Sc = 2039.20 - 120.31T + 3.4209T^2 - 0.040437T^3 \quad (\text{Eq. 3})$$

192 This study evaluated the variations in  $\text{CH}_4$  fluxes estimated by eight widely used

193 wind-based models developed for different environments, including wind tunnels,

194 open sea, estuarine systems and lakes, as follows:

195 LM86 ([Liss and Merlivatt 1986](#))

$$196 \quad F_{LM86} = 0.17U_{10} (Sc / 600)^{-2/3} (C_{\text{water}} - C_{\text{eq}}) \quad 0 < U_{10} \leq 3.6 \quad (\text{Eq. 4})$$

$$197 \quad F_{LM86} = (2.85U_{10} - 9.65)(Sc / 600)^{-1/2} (C_{\text{water}} - C_{\text{eq}}) \quad 3.6 < U_{10} \leq 13 \quad (\text{Eq. 5})$$

198 W92a ([Wanninkhof, 1992](#))

$$199 \quad F_{W92a} = 0.31U_{10}^2 (Sc / 660)^{-1/2} (C_{\text{water}} - C_{\text{eq}}) \quad (\text{Eq. 6})$$

200 RC01 ([Raymond and Cole, 2001](#))

$$201 \quad F_{RC01} = 1.91 \exp(0.35U_{10})(Sc / 600)^{-1/2} (C_{\text{water}} - C_{\text{eq}}) \quad (\text{Eq. 7})$$

202 CL98 ([Cole & Caraco, 1998](#))

203  $F_{CL98} = [2.07 + (0.215 \times U_{10}^{1.7})](Sc / 600)^{-2/3} (C_{water} - C_{eq})$  (Eq. 8)

204 W92b (Wanninkhof, 1992)

205  $F_{W92b} = 0.45U_{10}^{1.64} (Sc / 600)^{-1/2} (C_{water} - C_{eq})$  (Eq. 9)

206 CW03 (Crusius & Wanninkhof, 2003)

207  $F_{CW03} = 0.72U_{10} (Sc / 600)^{-2/3} (C_{water} - C_{eq}) \quad U_{10} < 3.7$  (Eq. 10)

208  $F_{CW03} = (4.33U_{10} - 13.3)(Sc / 600)^{-1/2} (C_{water} - C_{eq}) \quad U_{10} \geq 3.7$  (Eq. 11)

209 In the above equations,  $U_{10}$  was determined according to the logarithmic wind profile  
 210 relationship (Crusius and Wanninkhof, 2003):

211  $U_{10} = U_z [1 + \frac{(C_{d10})^{1/2}}{K} \ln(\frac{10}{z})]$  (Eq. 12)

212 where  $U_z$  is the wind speed ( $m s^{-1}$ ) at height  $z$  above the water surface (2.5 m in this  
 213 study),  $C_{d10}$  is the drag coefficient at 10 m above the water surface ( $0.0013 m s^{-1}$ ), and  
 214  $K$  is the von Karman constant (0.41). Generally, the calculation of  $U_{10}$  is sensitive of  
 215 the stability of the atmosphere. If the atmosphere over the aquatic systems is unstable,  
 216 and the equation used to calculate  $U_{10}$  needs to be adjusted. The air-water temperature  
 217 difference can be used to determine the atmospheric stability; if the air-water  
 218 temperature difference is positive, the atmosphere is considered stable. During the  
 219 study period, the air temperature was 0.1-3.8 °C higher than water temperature,  
 220 indicating that the atmosphere over the ponds was largely in the stability regime.  
 221 Therefore, no adjustment was needed for  $U_{10}$ , and Eq. 12 was appropriate for  
 222 calculating  $U_{10}$ . Some recent studies have applied surface renewal models that take

223 into account both wind speed and buoyancy to determine the  $k$  values (e.g., [Czikowsky](#)  
224 [et al., 2018](#); [MacIntyre et al., 2010](#); [MacIntyre et al., 2018](#)).

## 225 *2.5. Measurement of meteorological and environmental variables*

226 Meteorological variables including air temperature ( $A_T$ ), air pressure ( $A_P$ ) and  
227 precipitation were recorded using an automatic meteorological station (Vantage Pro 2,  
228 China) installed at the MRE weather station in the China Wetland Ecosystem  
229 Research Network. The distance between the automatic meteorological station and  
230 sampling ponds is about 75 m. The precision for air temperature, atmospheric  
231 pressure, and precipitation were  $\pm 0.2$  °C,  $\pm 1.5$  hPa, and  $\pm 0.4$  mm  $\text{min}^{-1}$ , respectively  
232 (Yang et al., 2020). Wind speed ( $W_S$ ) was measured at 1 Hz at a resolution of 0.4 m  
233  $\text{s}^{-1}$ .

234 Water temperature, electrical conductivity (EC), pH, dissolved oxygen (DO),  
235 total organic carbon (TOC) and total dissolved nitrogen (TDN) content of surface  
236 water (~20 cm below the water surface) were measured at each sampling site in all  
237 sampling campaigns. Water temperature and pH were measured by a portable  
238 pH/mV/Temperature meter (IQ150, IQ Scientific Instruments, USA), and EC and DO  
239 by an electrical conductivity meter (2265FS EC, Spectrum Technologies, USA) and a  
240 multiparameter water quality probe (550A YSI, USA), respectively. The relative  
241 standard deviations of EC, pH, and DO measurements were  $\square 1.0\%$ ,  $\square 1.0\%$  and  
242  $\square 2.0\%$ , respectively.

243 Water samples for TOC and TDN analyses were collected using a 5-L plexiglass

244 water sampler, transferred to a 150-mL polyethylene bottle, and then transported to  
245 the laboratory in an ice-packed cooler. The water samples were filtered through a  
246 0.45- $\mu\text{m}$  cellulose acetate filter (Biotrans nylon membranes); the filtrates were then  
247 analysed by a TOC analyzer (TOC-V<sub>CPH/CPN</sub>, Shimadzu, Kyoto, Japan) for TOC and a  
248 flow injection analyzer (Skalar Analytical SAN<sup>++</sup>, The Netherlands) for NO<sub>3</sub><sup>-</sup>-N. The  
249 detection limits for NO<sub>3</sub><sup>-</sup>-N and TOC were 6  $\mu\text{g L}^{-1}$  and 4  $\mu\text{g L}^{-1}$ , respectively. The  
250 relative standard deviations of NO<sub>3</sub><sup>-</sup>-N and TOC measurements were  $\approx$ 3.0% and  
251  $\approx$ 1.0%, respectively.

## 252 2.6. Statistical analysis

253 Repeated-measures analysis of variance (RM-ANOVA) was conducted to test the  
254 differences in diffusive CH<sub>4</sub> flux between the different approaches over the study  
255 period. Pearson correlation analysis was used to examine the relationships between (1)  
256 dissolved CH<sub>4</sub> concentration or CH<sub>4</sub> fluxes and environmental variables, and (2)  
257 diffusive CH<sub>4</sub> fluxes measured using *FCs* and those estimated using the *TBL* models.  
258 The coefficient of variation (CV) for CH<sub>4</sub> fluxes on each sampling campaign was  
259 determined by dividing the standard deviation by the mean value. Statistical analyses  
260 were conducted using the software SPSS (v. 17.0, SPSS Inc., USA) and the  
261 significance level was set at  $p < 0.05$ . Data are presented as mean  $\pm$  1 standard error.

262 Generalized linear modelling was conducted to compare the variables that  
263 influenced CH<sub>4</sub> emission flux from the different methods (i.e. *FCs* + 8 *TBL* models).  
264 The “gls” function from the “nlme” R package (Pinheiro et al., 2018) with a saturated

265 model was conducted for all variables (dissolved CH<sub>4</sub>,  $U_{10}$ , water temperature,  
266 dissolved oxygen, total dissolved carbon and dissolved nitrate). This model was run  
267 using the stepAIC function in R “MASS” package that follows the Akaike  
268 Information Criterion (AIC) (Venables and Ripley, 2002). It was used to identify the  
269 best model (lowest AIC value) in each case.

### 270 **3. Results**

#### 271 *3.1. Meteorological and environmental variables*

272 The average air temperature ( $A_T$ ) and air pressure ( $A_P$ ) during the study were  
273  $28.7 \pm 0.4$  °C (range: 18.6–35.6 °C) and  $1010.0 \pm 0.5$  hPa (range: 985–1025 hPa),  
274 respectively. Notably, the maximal  $A_T$  appeared in July and the minimal  $A_P$  happened  
275 in August, different from the other months.  $W_S$  ranged from 0.2 to 18.8 m s<sup>-1</sup>, and it  
276 varied between seasons, with a peak in July (Figure S3a). Approximately 92% of  $W_S$   
277 fell within the range of 0.2–4.0 m s<sup>-1</sup> (Figure S3b).

278 There were temporal variations in surface water characteristics during the study  
279 period. The mean water temperature ranged from 18.1 °C (November) to 34.4 °C  
280 (August) (Figure S4a), while the mean DO concentration varied between 9.4 mg L<sup>-1</sup>  
281 (August) and 19.9 mg L<sup>-1</sup> (November) (Figure S4). The mean TOC concentration  
282 varied between 9.9 mg L<sup>-1</sup> (July) and 57.3 mg L<sup>-1</sup> (November) (Figure S3), while  
283 NO<sub>3</sub><sup>-</sup>-N concentrations ranged from 504 µg N L<sup>-1</sup> (June) to 10.7 µg N L<sup>-1</sup> (November)  
284 (Figure S4).

#### 285 *3.2. Model estimated $k$ values and dissolved CH<sub>4</sub> concentrations*

286 The mean  $k$  value showed considerable variations between models:  $k_{RC01}$   
287 ( $6.5 \pm 0.8 \text{ cm h}^{-1}$ )  $>$   $k_{W92a}$  ( $3.5 \pm 0.7 \text{ cm h}^{-1}$ )  $>$   $k_{FCs}$  ( $3.2 \pm 0.4 \text{ cm h}^{-1}$ )  $>$   $k_{CL98}$  ( $2.9 \pm 0.3 \text{ cm}$   
288  $\text{h}^{-1}$ )  $>$   $k_{CW03}$  ( $2.5 \pm 0.5 \text{ cm h}^{-1}$ )  $>$   $k_{W92b}$  ( $2.4 \pm 0.4 \text{ cm h}^{-1}$ )  $>$   $k_{LM86}$  ( $0.6 \pm 0.1 \text{ cm h}^{-1}$ ) (Figure  
289 1).

290 Dissolved  $\text{CH}_4$  concentration varied considerably during the study period  
291 ( $0.1\text{--}31.1 \mu\text{mol L}^{-1}$ ), with a large increase between June and August, followed by a  
292 small decrease toward November (Figure 2). The water was supersaturated in  $\text{CH}_4$  in  
293 all ponds and on all sampling dates, with an overall mean of  $4.8 \pm 0.8 \mu\text{mol L}^{-1}$  ( $162.0$   
294  $\pm 18.4 \text{ ppmv}$ ), equivalent to 8700% saturation (range of  $200 - 5.9 \times 10^4 \%$  saturation).

#### 295 3.4. $\text{CH}_4$ flux estimates by using TBL models and FCs method

296 There were considerable differences in the estimated diffusive  $\text{CH}_4$  fluxes among  
297 the TBL models ( $TBL_{RC01}$ :  $215.9 \pm 39.2 \mu\text{mol m}^{-2} \text{ h}^{-1}$ ;  $TBL_{CL98}$ :  $115.0 \pm 21.9 \mu\text{mol m}^{-2}$   
298  $\text{h}^{-1}$ ;  $TBL_{W92a}$ :  $102.9 \pm 19.5 \mu\text{mol m}^{-2} \text{ h}^{-1}$ ;  $TBL_{W92b}$ :  $78.3 \pm 13.9 \mu\text{mol m}^{-2} \text{ h}^{-1}$ ;  $TBL_{CW03}$ :  
299  $74.9 \pm 13.2 \mu\text{mol m}^{-2} \text{ h}^{-1}$ ; and,  $TBL_{LM86}$ :  $19.5 \pm 3.7 \mu\text{mol m}^{-2} \text{ h}^{-1}$ ) (Table 1, Figure 3 and  
300 Figure S5). Although there were marked variations in the flux estimates among the  
301 various models, results from all models showed similar temporal patterns (Figure 3).  
302 The largest fluxes were generally recorded between August and October, while the  
303 lowest fluxes were consistently recorded in June and November (Figure 3).

304 Direct measurements using FCs with gauze (FCs-G) and without gauze  
305 (FCs-NG) methods were  $75.0 \pm 12.5$  (Figure 3) and  $2231.3 \pm 681.3 \mu\text{mol m}^{-2} \text{ h}^{-1}$   
306 (Figure S6; Yang et al., unpublished data), showing significant difference between the



307 two methods (Independent Samples T-Test,  $F = 118.190$ ,  $p < 0.001$ ). On average,  
308 ebullitive  $\text{CH}_4$  flux accounted for 33%-99% of the total  $\text{CH}_4$  emissions during the  
309 study period.

### 310 *3.5. Environmental influences on dissolved $\text{CH}_4$ concentrations and fluxes*

311 Pearson correlation analysis showed that dissolved  $\text{CH}_4$  concentration was  
312 positively correlated with air temperature and TOC ( $p < 0.01$ ), and negatively with  
313  $\text{NO}_3^-$ -N and EC ( $p < 0.01$ ) (Table 2).  $\text{CH}_4$  flux was positively correlated with air  
314 temperature ( $p < 0.05$ ), TOC and dissolved  $\text{CH}_4$  concentration ( $p < 0.01$ ), and negatively  
315 with  $\text{NO}_3^-$ -N ( $p < 0.01$ ) and EC ( $p < 0.05$ ) (Table 2 and Table S3). Environmental  
316 variables explained a larger proportion of variability in  $\text{CH}_4$  flux derived from the  
317 TBL models ( $R^2 = 0.46$ - $0.54$ ) than those from direct FCs measurements ( $R^2 = 0.35$ )  
318 (Table S2).

## 319 **4. Discussion**

### 320 *4.1. $\text{CH}_4$ supersaturation and degassing from aquaculture ponds*

321 There are very few studies on  $\text{CH}_4$  in small ponds, particularly, those created for  
322 aquaculture purposes. In this study, the dissolved  $\text{CH}_4$  concentration in surface water  
323 of the aquaculture ponds ranged from 0.1 to 31.1  $\mu\text{mol L}^{-1}$  during the study period,  
324 which was higher than that observed in many small ponds in Florida ( $\sim 2.2 \mu\text{mol L}^{-1}$ ;  
325 Barber et al., 1988), Colorado ( $\sim 1.0 \mu\text{mol L}^{-1}$ ; Bastviken et al., 2004), Wisconsin and  
326 Minnesota (0.3–2.3  $\mu\text{mol L}^{-1}$ ; Smith and Lewis, 1992) in the USA, in Sweden ( $\sim 1.3$   
327  $\mu\text{mol L}^{-1}$ ; Natchimuthu et al., 2014), Canada (0.5–6.7  $\mu\text{mol L}^{-1}$ ; Pelletier et al., 2014),

328 and Siberia ( $\sim 2.6 \mu\text{mol L}^{-1}$ ; [Repo et al., 2007](#)). In addition,  $\text{CH}_4$  concentration in our  
329 aquaculture ponds was generally larger than that in some nutrient-enriched rivers in  
330 China, i.e. Lixiahe River ( $0.2\text{--}0.8 \mu\text{mol L}^{-1}$ ; [Wu et al., 2019](#)), Beitang Drainage River  
331 and Dagu Drainage River ( $0.3\text{--}1.7 \mu\text{mol L}^{-1}$ ; [Hu et al., 2018](#)). Similar to other inland  
332 aquatic systems, such as lakes (e.g., [Wen et al., 2016](#); [Wik et al., 2016](#); [Yan et al.,](#)  
333 [2018](#)), reservoirs (e.g., [Deemer et al., 2016](#); [Musenze et al., 2014](#); [Wang et al., 2017](#)),  
334 rivers (e.g., [Barbosa et al., 2016](#); [Striegl et al., 2012](#)), floodplains ([Barbosa et al., 2020](#))  
335 and small ponds (e.g., [Holgerson and Raymond, 2016](#); [Wik et al., 2016](#)), our  
336 aquaculture ponds were vastly supersaturated in  $\text{CH}_4$  relative to air ( $2.71\text{--}599.81$   
337 times the equilibrium concentration ) ([Figure 2b](#)). The small temperate ponds in the  
338 Yale Myers Forest in Connecticut, the USA, have some of the highest concentrations  
339 of  $\text{CH}_4$  ( $21.0\text{--}58.9 \mu\text{mol L}^{-1}$ , equivalent to  $119\text{--}2907$  times the equilibrium  
340 concentration) ([Holgerson, 2015](#)). The  $\text{CH}_4$  concentrations and supersaturation levels  
341 in our aquaculture ponds fall well within the range reported by [Holgerson \(2015\)](#),  
342 showing that aquaculture ponds in the subtropical estuaries are also hotspots for  $\text{CH}_4$   
343 production and emission.

344 In inland aquatic ecosystems, the strong  $\text{CH}_4$  release is likely a result of large  
345 organic matter inputs from the catchment, algae and aquatic plants that sustain high  
346 methanogenesis rates ([Finlay et al., 2009](#); [Lundin et al., 2013](#); [Venkiteswaran et al.,](#)  
347 [2013](#); [Yan et al., 2018](#)), as indicated by the significant relationship between dissolved  
348  $\text{CH}_4$  and nutrient level ([Huttunen et al., 2003](#); [Kortelainen et al., 2001](#); [Wen et al.,](#)  
349 [2016](#)). The shrimp ponds in this study are semi-artificial ecosystems that are

350 maintained through a daily feed supply for the production of aquatic animals.  
351 However, only a small portion of the feed input is converted into shrimp biomass,  
352 with the feed utilization efficiency of ~4.0–27.4% (Chen et al., 2016; Molnar et al.,  
353 2013; Yang et al., 2017b). Surface sediments in the aquaculture systems typically  
354 retain a large amount of organic matter from feces and residual feeds (Chen et al.,  
355 2016; Yang et al., 2017b) that can support high levels of CH<sub>4</sub> production and its  
356 subsequent release to atmosphere. Although organic matter content was not quantified  
357 in this study, our results confirmed the significantly correlation between dissolved  
358 CH<sub>4</sub> and TOC concentration ( $p < 0.01$ ; Table 2), which lends support to the notion that  
359 CH<sub>4</sub> supersaturation in the aquaculture ponds was related to the large input of organic  
360 matter.

#### 361 4.2. Comparison of different TBL modelled CH<sub>4</sub> fluxes

362 Although previous studies have compared the performance of different TBL  
363 models in estimating diffusive CH<sub>4</sub> flux in inland waters (Amouroux et al., 2002; Li et  
364 al., 2015; Musenze et al., 2014; Xiao et al., 2017; Zappa et al., 2007), such  
365 comparison is scarce for shallow ponds, particularly aquaculture ponds. To the best of  
366 our knowledge, this study is the first attempt to compare the estimates of diffusive  
367 CH<sub>4</sub> flux using different TBL models over the whole aquaculture period in  
368 aquaculture ponds. Interestingly, although the patterns of temporal variations in  
369 diffusive CH<sub>4</sub> flux were largely consistent among the TBL models (Figure 3), there  
370 were clear differences in the magnitude of flux estimated from the different models  
371 (Table 1).

372 Notably, the mean flux estimated by the  $TBL_{RC01}$  model ( $215.6 \mu\text{mol m}^{-2} \text{h}^{-1}$ ) was  
373 an order of magnitude greater than that derived from the  $TBL_{LM86}$  model ( $19.4 \mu\text{mol}$   
374  $\text{m}^{-2} \text{h}^{-1}$ , [Figure 3](#)). Moreover,  $\text{CH}_4$  flux estimated by the  $TBL_{RC01}$  model was 2 - 3  
375 times larger than that by the  $TBL_{W92a}$ ,  $TBL_{CL98}$ ,  $TBL_{W92b}$  and  $TBL_{CW03}$  models ([Table 1](#)  
376 and [Figure S5](#)). However, there was no significant difference between the  $TBL_{W92a}$   
377 and  $TBL_{CL98}$  models ( $p>0.05$ ; [Table 1](#) and [Figure S5](#)) or between the  $TBL_{W92b}$  and  
378  $TBL_{CW03}$  models ( $p>0.05$ ; [Table 1](#) and [Figure S5](#)). In other inland waters (river and  
379 reservoirs), [Gao et al. \(2014\)](#) and [Musenze et al. \(2014\)](#) also found that the estimated  
380 diffusive  $\text{CH}_4$  fluxes derived from the  $TBL_{RC01}$  model were substantially greater than  
381 those from other  $TBL$  models.

382 The differences in the estimated  $\text{CH}_4$  flux between different  $TBL$  models were  
383 likely a result of different weighting of wind as a driver of gas transfer velocity  
384 ([Musenze et al., 2014, Figure 1](#)). Because these wind-based models were originally  
385 developed for specific environments under different conditions ([Gao et al., 2014](#);  
386 [Musenze et al., 2014](#)), their suitability for other situations could be questioned ([Bade,](#)  
387 [2009](#); [Musenze et al., 2014](#); [Schilder et al., 2013](#)). The  $TBL_{CL98}$  and  $TBL_{CW03}$  models  
388 were developed for lentic ecosystems under a range of wind speed, which most  
389 closely resemble aquaculture pond conditions. One may therefore argue that these two  
390 models would be most applicable to aquaculture ponds, although more *in situ*  
391 measurement will be needed to further increase the accuracy of the estimate.

#### 392 4.3. Comparison of $\text{CH}_4$ fluxes derived from FCs measurement and $TBL$ models

393 Previous studies have shown that  $\text{CH}_4$  fluxes estimated by  $TBL$  models tend to be

394 lower than those measured by *FCs* (Chuang et al., 2017; Duchemin et al., 1999; Li et  
395 al., 2015; Matthews et al., 2003). This study also compared CH<sub>4</sub> fluxes measured by  
396 *FCs* and those estimated by *TBL* models over the aquaculture season (Table 1 and  
397 Figure S5). Although there were significant correlations between *TBL* model estimates  
398 and *FCs* measurements ( $p < 0.05$  in all cases), the agreement between the two  
399 methods varied considerably between models (Figure 4). The *TBL*<sub>W92b</sub> and *TBL*<sub>CW03</sub>  
400 models gave the largest  $r^2$  values (0.82 and 0.83, respectively) and good agreements  
401 with *FCs* measurements (slope = 0.92 and 0.89, respectively), whereas *TBL*<sub>CL98</sub>  
402 yielded mean estimates virtually identical to *FCs* measurements (slope = 1) but with  
403 larger variability around the mean ( $r^2 = 0.53$ ) (Figures 4d-f). In contrast, *TBL*<sub>LM86</sub>  
404 vastly underestimated *FCs* fluxes whereas *TBL*<sub>RC01</sub> grossly overestimated *FCs* fluxes  
405 (Figures 4a,b). Approximately 80% of the diffusive CH<sub>4</sub> fluxes estimated by the  
406 models fell within the range measured by the *FC* method.

407 Balancing the consideration of overall agreement (regression slope) and estimate  
408 variability (regression  $r^2$ ), the *TBL*<sub>W92b</sub> and *TBL*<sub>CW03</sub> models appeared to give the best  
409 approximations of *FCs* measurements. While previous studies showed that *FCs* were  
410 more appropriate for determining greenhouse gas fluxes in heterogeneous  
411 environments such as lakes and reservoirs (Cole et al., 2010; Duchemin et al., 1999;  
412 Murray et al., 2015; Vachon et al., 2010; Wu et al., 2018), our results suggest that  
413 *TBL*<sub>W92b</sub> and *TBL*<sub>CW03</sub> models are reliable alternatives for estimating CH<sub>4</sub> diffusive  
414 flux in shallow aquaculture ponds.

415 In addition to diffusive flux from the water column, bottom sediment could also

416 contribute to CH<sub>4</sub> emission via ebullition, especially in eutrophic, shallow aquaculture  
417 ponds. This is illustrated by the differences in the measured CH<sub>4</sub> flux using *FCs* with  
418 and without gauze in our aquaculture ponds (Figure S6). The CH<sub>4</sub> flux measured by  
419 *FCs* without gauze ( $2231.3 \pm 681.3 \mu\text{mol m}^{-2} \text{h}^{-1}$ ) were one to two orders of magnitude  
420 higher than that by *FCs* with gauze ( $75.0 \pm 12.5 \mu\text{mol m}^{-2} \text{h}^{-1}$ ) (Figure S6); from this  
421 ebullition was estimate to contribute 96.6% to the total CH<sub>4</sub> emissions. Overall, our  
422 results showed that ebullition was the primary path of CH<sub>4</sub> emission in aquaculture  
423 ponds, and that ebullitive flux vs. diffusive flux could be easily resolved with a simple  
424 design of *FCs* with a detachable gauze.

#### 425 4.4. Implications of the comparison between different methods

426 The *FCs* method is the popular technique for measuring CH<sub>4</sub> emissions due to its  
427 ability to detect low fluxes and the simplicity of its operating principle (Bastviken et  
428 al., 2015; Lorke et al., 2015; Musenze et al., 2014; Podgrajsek et al., 2014). However,  
429 the *FCs* method requires time-consuming manual operation, which limits the  
430 frequency of measurements and can be difficult to deploy in remote areas (Acosta et  
431 al., 2017; Morin et al., 2017). Improvement of the global CH<sub>4</sub> budget would require  
432 high-resolution emission data covering large time and spatial scales, which obviously  
433 is difficult to achieve with the *FCs* method.

434 Large-scale estimates of aquatic CH<sub>4</sub> emissions using *TBL* models has been  
435 gaining popularity (Holgerson and Raymond, 2016; Martinez-Cruz et al., 2016;  
436 Musenze et al., 2014; Wang et al., 2017) due to their simplicity, practicality and low  
437 cost. There are, however, different *TBL* models to choose from, and the large

438 differences in the model performances (Figure 4) mean that selecting the appropriate  
439 model(s) would be critical, or otherwise large errors would occur when upscaling the  
440 results from small ponds to the regional/ global scale. Our results suggest that  
441 *TBL<sub>W92b</sub>* and *TBL<sub>CW03</sub>* models could be used as effective and convenient alternatives to  
442 *FCs* in shallow aquaculture ponds.

#### 443 4.5. Limitation and future research

444 The *FCs* method is a common method to measure CH<sub>4</sub> fluxes from aquatic  
445 ecosystems. However, *FCs* may create microenvironments that affect the boundary  
446 layer conditions through, for instance, blockage of wind, change of atmospheric  
447 pressure at the measurement point, and change in the gas transfer rate through  
448 pressure build-up (Duchemin et al., 1999; Matthews et al., 2003; Musenze et al.,  
449 2014). For example, the turbulence resulted from the chamber walls can enhance the  
450 efficiency of gas exchange and increase gas fluxes during low wind conditions  
451 (Matthews et al., 2003; Xiao et al., 2016).

452 *TBL* models rely on the gas transfer velocity coefficient ( $k_x$ ), which itself is  
453 estimated from some empirical wind-based models. Effects of artificial aeration,  
454 which is commonly done in aquaculture ponds, on  $k_x$  are unknown. More importantly,  
455 the *TBL* models ignore the effect of buoyancy fluxes near the air-water interface on  $k_x$ .  
456 An alternative is the surface renewal model (SRM), which considers both wind speed  
457 and buoyancy (e.g., Czikowsky et al., 2018; MacIntyre et al., 2010; MacIntyre et al.,  
458 2018).

459 The use of eddy covariance (EC) technique is increasingly popular as it can

460 provide a better characterization of the variation in CH<sub>4</sub> fluxes through  
461 quasi-continuous measurements (Acosta et al., 2017; Morin et al., 2017; Xiao et al.,  
462 2014; Zhao et al., 2019). However, its application in small water bodies (e.g., ponds)  
463 is limited by footprint contamination (Zhao et al., 2019). Developing a practical and  
464 effective way to reduce the flux footprint and the contamination from gaseous sources  
465 outside the water body will allow broader application of EC method in the future.

466 Different methods have their own limitations; careful comparison and cross  
467 calibration would be needed to increase the overall accuracy of these methods and to  
468 improve the global CH<sub>4</sub> budget.

## 469 **5. Conclusions**

470 Despite the large CH<sub>4</sub> emission potential from small ponds, there are few studies  
471 comparing the different methods to estimate CH<sub>4</sub> fluxes across the water-air interface.  
472 In this study, *FCs* and *TBL* models were used to estimate CH<sub>4</sub> fluxes from aquaculture  
473 ponds. Our results indicate that dissolved CH<sub>4</sub> concentrations in the subtropical  
474 shallow aquaculture ponds were on average ~87 times oversaturated relative to the  
475 ambient air, and thus the ponds acted as strong atmospheric CH<sub>4</sub> sources. The high  
476 organic matter loading contributed to CH<sub>4</sub> supersaturation in the ponds. This study for  
477 the first time compared the CH<sub>4</sub> fluxes measured directly by floating chambers (*FCs*)  
478 and those estimated by thin boundary layer (*TBL*) models (*TBL<sub>LM86</sub>*, *TBL<sub>W92a</sub>*, *TBL<sub>RC01</sub>*,  
479 *TBL<sub>CL98</sub>*, *TBL<sub>W92b</sub>*, and *TBL<sub>CW03</sub>*). The model estimates of diffusive CH<sub>4</sub> fluxes were  
480 highly variable, and were overall 27 - 300% larger than those measured by *FCs*. The  
481 *TBL<sub>W92b</sub>* and *TBL<sub>CW03</sub>* models provided a robust and simple alternative to *FCs* in



482 estimating diffusive CH<sub>4</sub> fluxes. Our results suggest that the comparison of different  
483 methods and selection of the most appropriate method(s) should be a high research  
484 priority to improve the accuracy of greenhouse gas fluxes from aquaculture ponds and  
485 other aquatic ecosystems.

#### 486 **Declaration of competing interest**

487 The authors declare that they have no known competing financial interests or  
488 personal relationships that could have appeared to influence the work reported in this  
489 paper.

#### 490 **Acknowledgments**

491 This research was financially supported by the National Science Foundation of China (grant  
492 numbers 41801070 and 41671088), National Key Research & Development Plan "Strategic  
493 International Scientific and Technological Innovation Cooperation" (2016YFE0202100), Council  
494 of the Hong Kong Special Administrative Region, China (CUHK458913 and CUHK Direct Grant  
495 SS15481), Spanish Government (grant CGL2016-79835), Catalan Government (grant SGR  
496 2017-1005), European Research Council (Synergy grant ERC-SyG-2013-610028), Open Research  
497 Fund Program of Jiangsu Key Laboratory of Atmospheric Environment Monitoring & Pollution  
498 Control (grant KHK1806), Priority Academic Program Development of Jiangsu Higher Education  
499 Institutions (PAPD), and Minjiang Scholar Programme. We thank Qianqian Guo, Guanghui Zhao,  
500 and Ling Li of the School of Geographical Sciences, Fujian Normal University, for assistance in  
501 the field.

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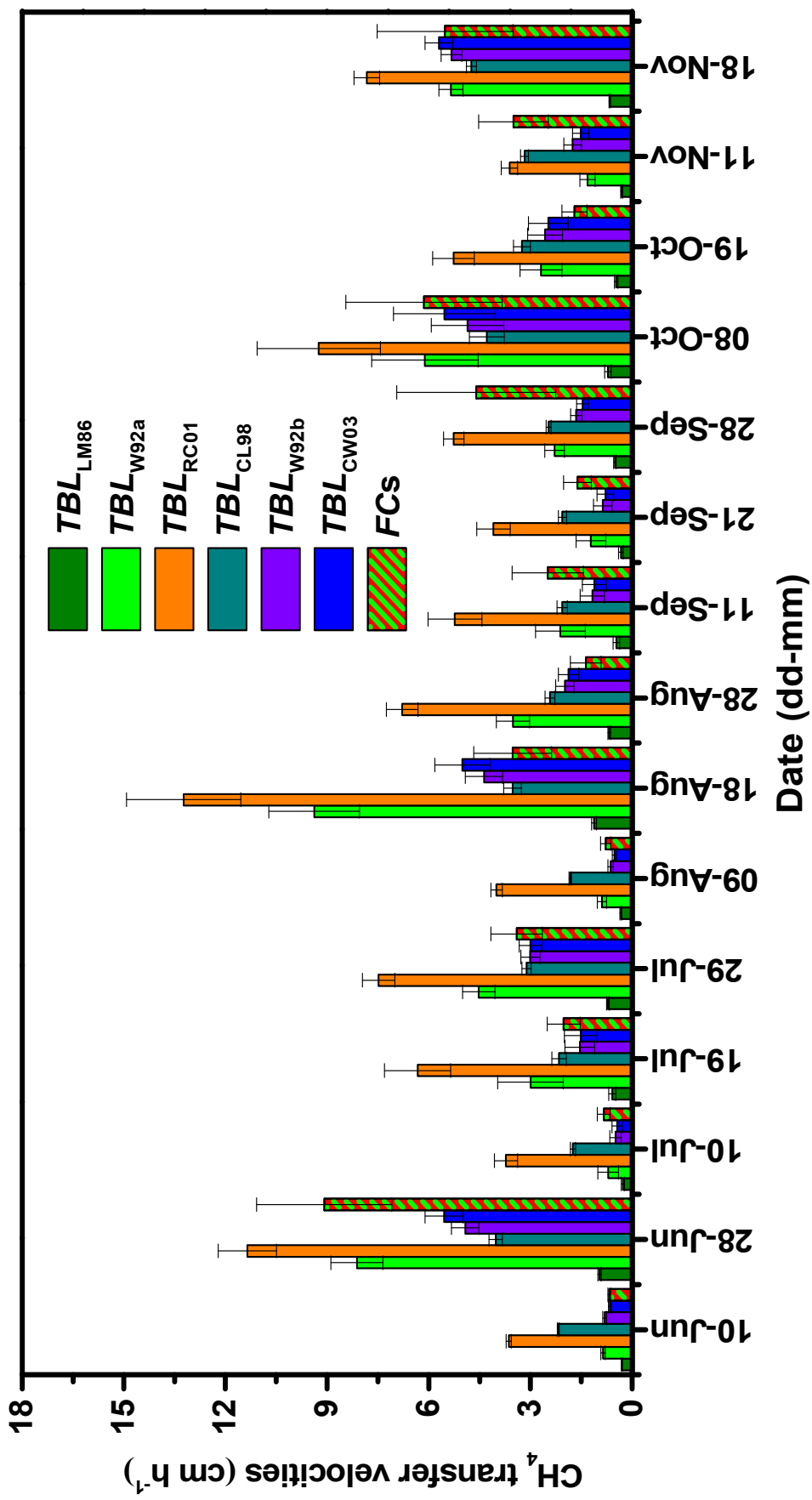
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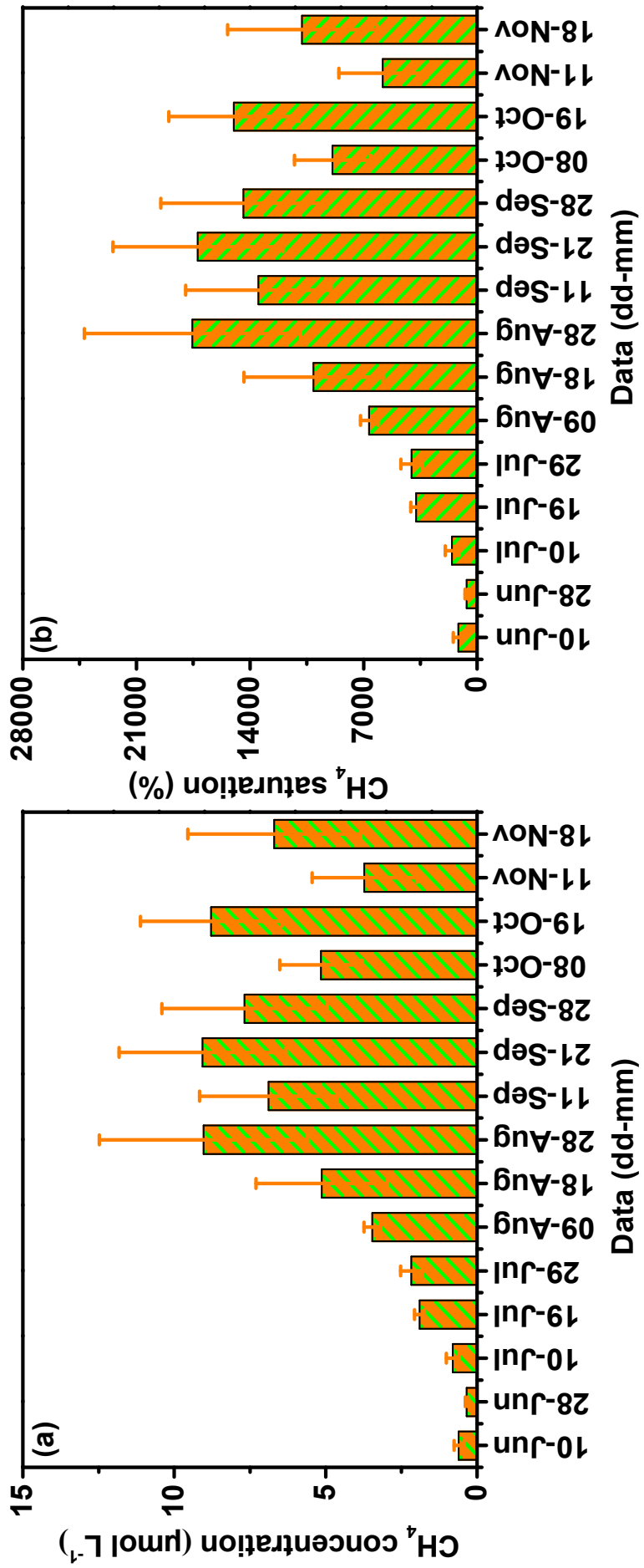
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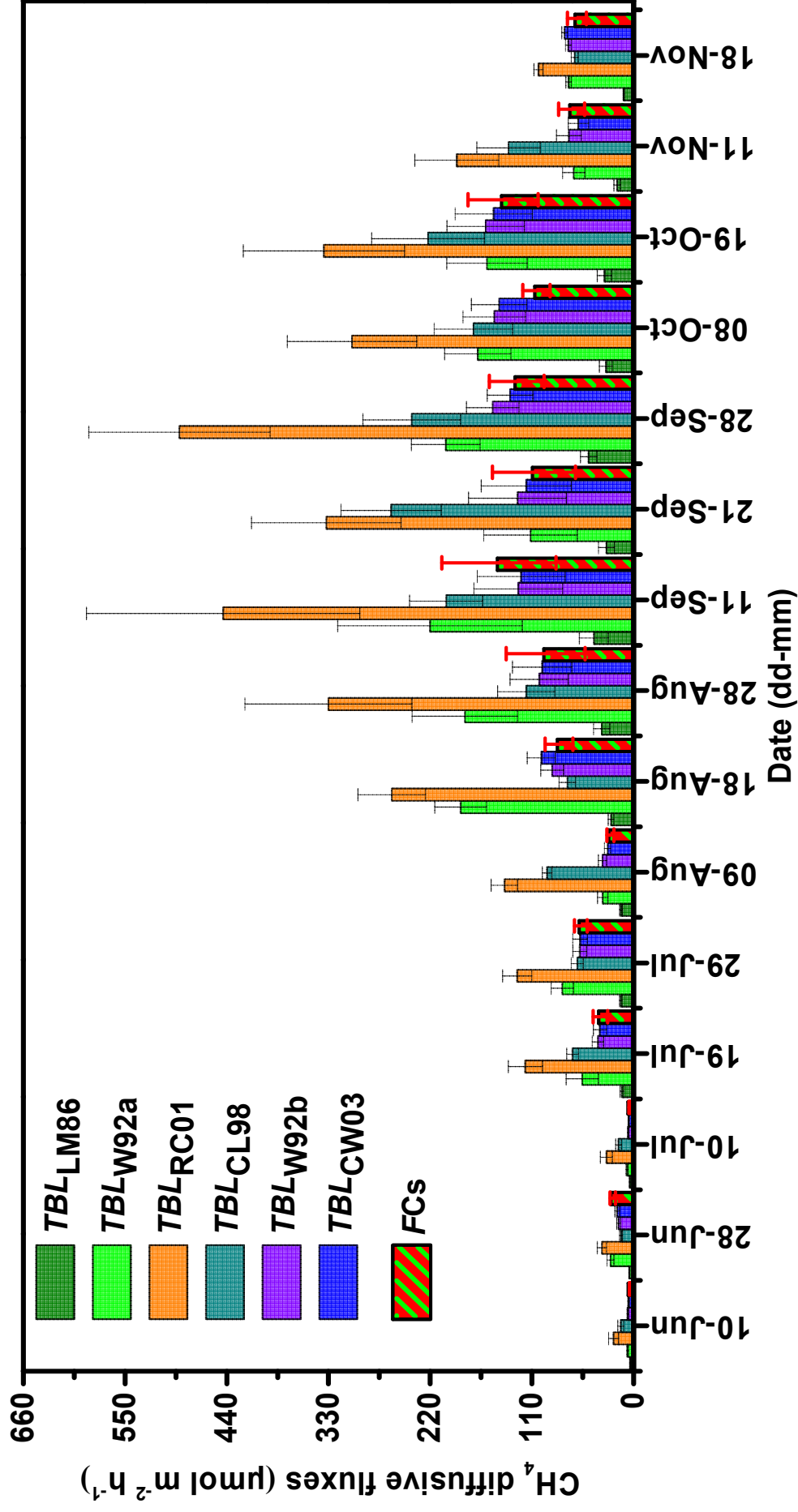
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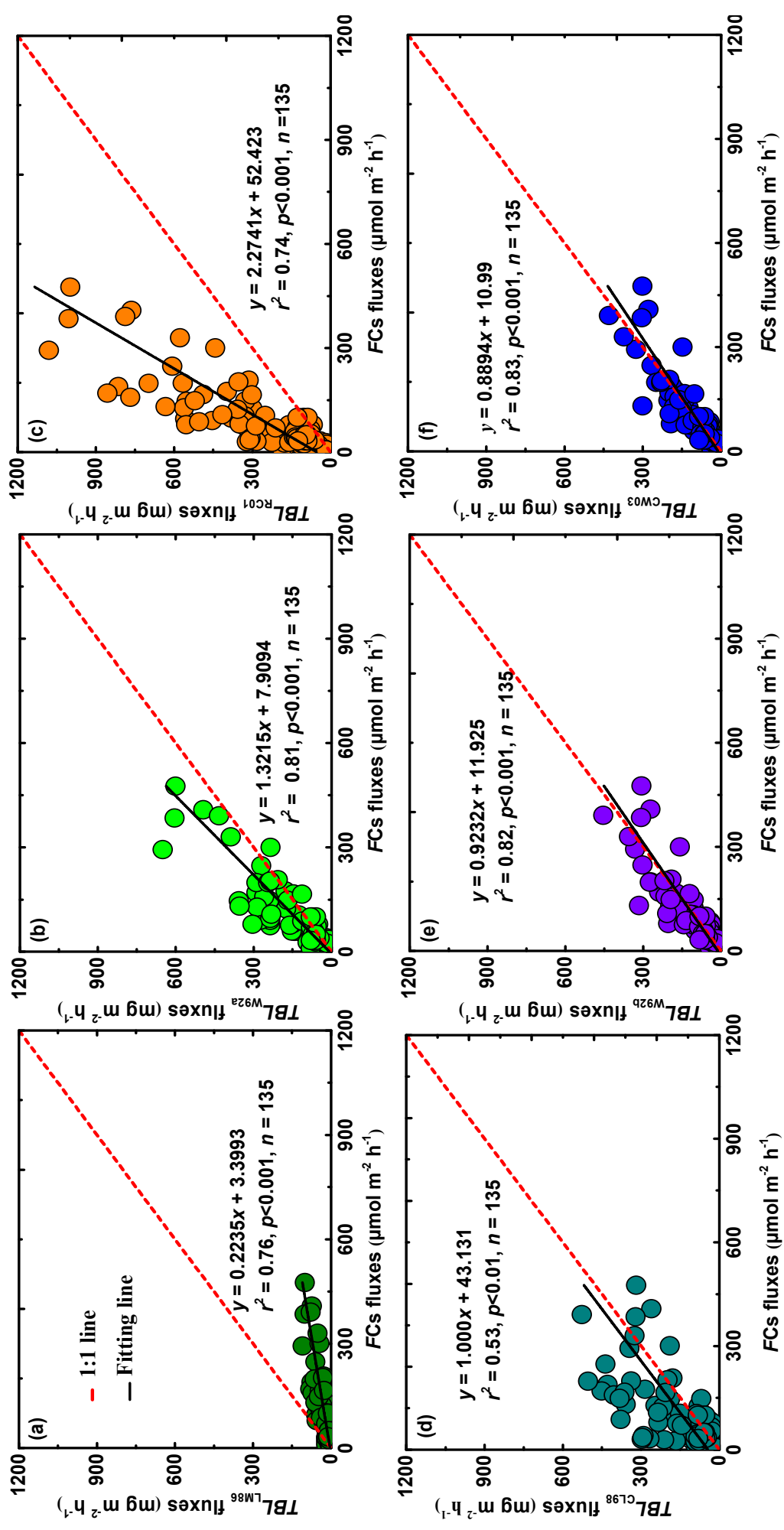
1  
 2 **Figure 1.** Temporal variation in CH<sub>4</sub> transfer velocities from the aquaculture ponds during the aquaculture period in the Min River  
 3 Estuary. Values represent the means of nine replicates samples, while the vertical lines indicate standard errors .



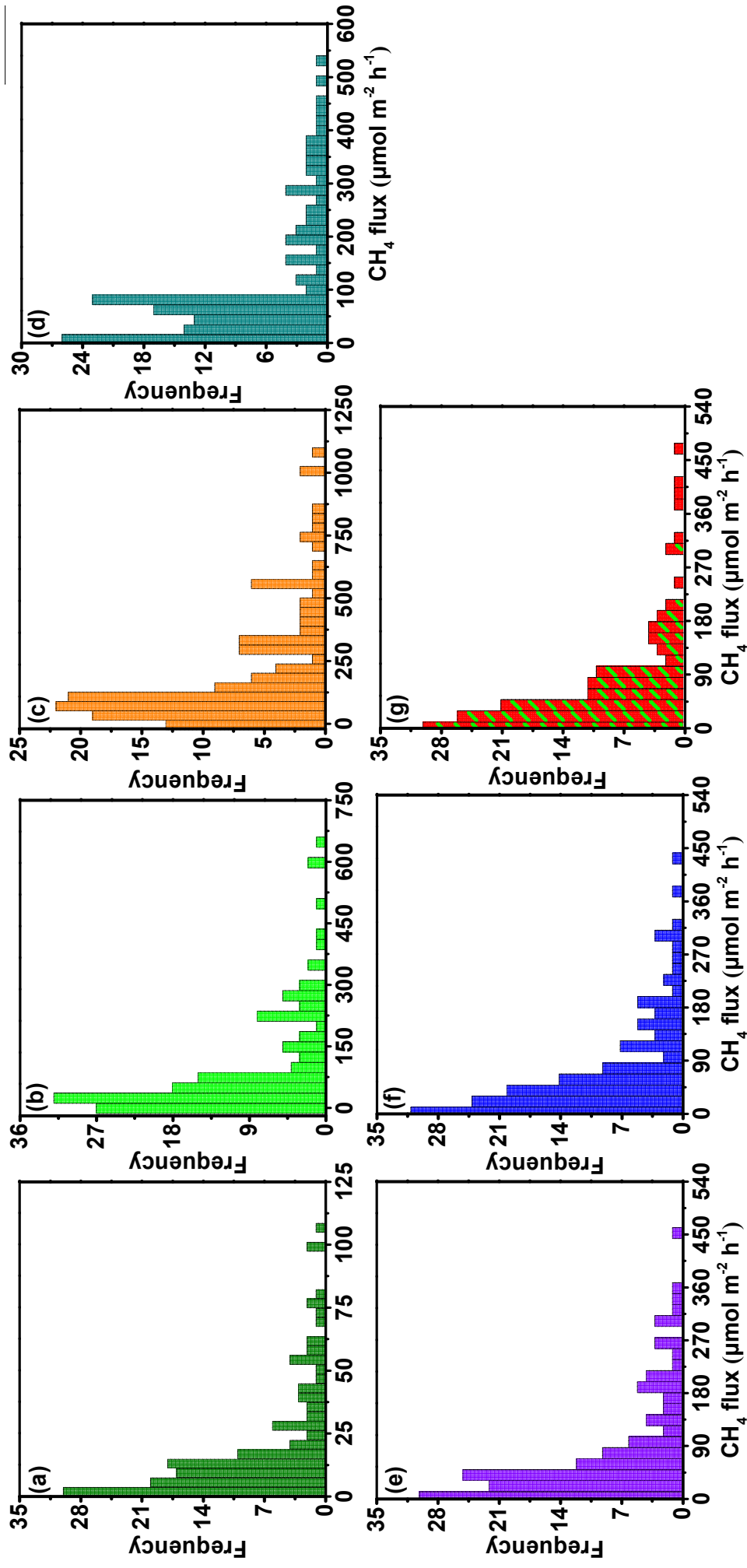
4  
5 **Figure 2.** Temporal variation in (a) CH<sub>4</sub> concentration and (b) CH<sub>4</sub> saturation in the surface water (20 cm depth) of the aquaculture ponds in the  
6 Min River Estuary during the aquaculture period. Values represent the means of nine replicates samples, while the vertical lines indicate standard  
7 errors.



8  
9 **Figure 3.** Temporal variation in CH<sub>4</sub> diffusive fluxes measured with the floating chamber method and the gas transfer velocity model methods  
10 during the aquaculture period from the aquaculture ponds in the Min River Estuary. Values represent the means of nine replicates samples, while  
11 the vertical lines indicate standard errors.



12  
 13 **Figure 4.** Comparison of CH<sub>4</sub> diffusive flux measured by using the FCs method and TBL models. Regression equation, linear correlation ( $r^2$ ) and  
 14 significance ( $p$ ) are also shown. Parameter bounds on the regression coefficients are 95% confidence intervals.



15

16 **Figure 5.** Frequency distribution of  $\text{CH}_4$  diffusive fluxes from (a)  $TBL_{LM86}$ , (b)  $TBL_{W92a}$ , (c)  $TBL_{RC01}$ , (d)  $TBL_{CL98}$ , (e)  $TBL_{W92b}$ , (f)  $TBL_{CW03}$ , and (g)

17 FCs measurements at the aquaculture ponds in the Min River Estuary during the aquaculture period.



1 **Table 1**

2 Summary of the *TBL* and *FCs* methods applied to measure  $\text{CH}_4$  diffusive fluxes from the aquaculture ponds in Min River Estuary during the  
 3 aquaculture period.

	<i>TBL</i> methods					<i>FCs</i> method		
	<i>TBL</i> <sub>LM86</sub>	<i>TBL</i> <sub>W92a</sub>	<i>TBL</i> <sub>RC01</sub>	<i>TBL</i> <sub>CL98</sub>	<i>TBL</i> <sub>W92b</sub>	<i>TBL</i> <sub>CW03</sub>		
<b>Minimum</b> ( $\mu\text{mol m}^{-2} \text{h}^{-1}$ )	0.6	1.3	5.6	1.9	1.9	1.3	1.3	1.3
<b>Maximum</b> ( $\mu\text{mol m}^{-2} \text{h}^{-1}$ )	108.8	650.0	1079.4	527.5	454.4	428.8	476.3	476.3
<b>Average</b> ( $\mu\text{mol m}^{-2} \text{h}^{-1}$ )	19.4	103.1	215.6	115.0	78.1	75.0	71.9	71.9
<b>Standard deviation</b>	23.1	130.6	236.3	122.5	91.3	86.9	88.8	88.8
<b>Coefficient of variation</b>	1.18	1.27	1.09	1.06	1.16	1.15	1.24	1.24

4 **Table 2**

5 Pearson correlation coefficients for dissolved CH<sub>4</sub> concentration, CH<sub>4</sub> diffusive fluxes  
 6 and environmental variables from the aquaculture ponds in Min River Estuary during  
 7 the aquaculture period<sup>a</sup>. Bold numbers denote correlation coefficients for significant  
 8 relationships.

Environmental variables	Dissolved CH <sub>4</sub> concentration	CH <sub>4</sub> diffusive fluxes
<b>Meteorological parameters</b>		
Air temperature	<b>0.214*</b>	<b>0.203*</b>
Wind speed ( <i>W</i> <sub>s</sub> )	NS	<b>0.281*</b>
Atmospheric pressure	NS	NS
<b>Water parameters</b>		
Water temperature	NS	NS
Dissolved oxygen (DO)	NS	NS
TOC concentration	<b>0.312**</b>	<b>0.296**</b>
NO <sub>3</sub> <sup>-</sup> -N concentration	<b>-0.401**</b>	<b>-0.392**</b>
Electrical conductivity (EC)	<b>-0.361**</b>	<b>-0.185*</b>

9 <sup>a</sup> The symbols \* and \*\* indicate significant correlations at the 0.05 and 0.01 levels, respectively. *n* = 135 for  
 10 environmental variables and CH<sub>4</sub> diffusive fluxes from the aquaculture ponds. CH<sub>4</sub> diffusive fluxes were directly  
 11 measured using floating chambers method.

# Supporting Information

## Title: Diffusive CH<sub>4</sub> fluxes from aquaculture ponds using floating chambers and thin boundary layer equations

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33 **Supporting Information Summary**

34 **No. of pages: 16**      **No. of figures: 4**      **No. of tables: 5**

35 **Page S5:** Figure S1. Location of the study area and sampling sites at aquaculture  
36 ponds in Min River Estuary, Southeast China.

37 **Page S6:** Figure S2. Schematic diagram for the gas sampling device of CH<sub>4</sub> diffusive  
38 flux across the water-air interface. Numbers 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, and 12  
39 represents chambers body, Neoprene floats, thin gauze, mooring anchor, sampling  
40 tube, 60-mL plastic syringes equipped with three-way stopcocks, fixed rope, valve  
41 body, valve body, gas collecting hole, ribbon, and handle, respectively.

42 **Page S7:** Figure S3. (a) Temporal variation in the wind speed ( $W_s$ ), and (b) frequency  
43 distribution of wind speed at the shrimp ponds in the Min River Estuary during the  
44 aquaculture period.

45 **Page S8:** Figure S4. Temporal variation in (a) water temperature, (b) dissolved  
46 oxygen, (c) TOC, and (b) N-NO<sub>x</sub><sup>-</sup> in the surface water (20 cm depth) of the  
47 aquaculture ponds in the Min River Estuary during the aquaculture period. *Error bars*  
48 represent standard error ( $n = 9$ ). Data are after Yang et al. [*unpublished data*] for  
49 reference and review only.

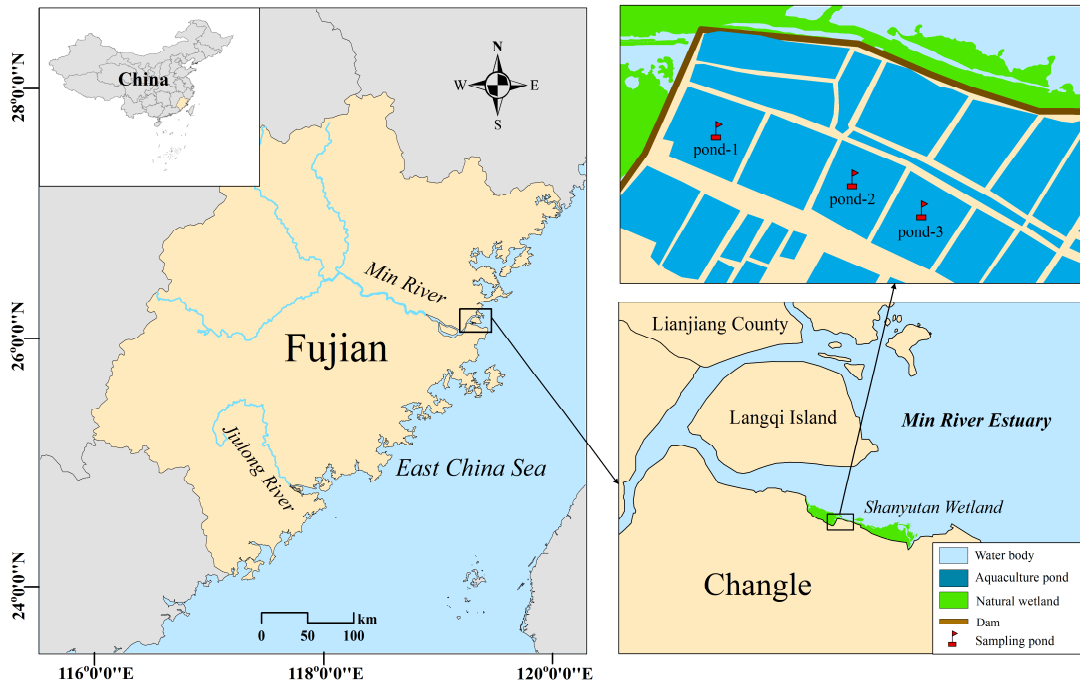
50 **Page S9:** Figure S5. Boxplots of CH<sub>4</sub> diffusive fluxes estimated using the *TBL* and  
51 *FCs* methods at the aquaculture ponds in Min River Estuary during the aquaculture  
52 period. The letters above the boxes represent the LSD (Least Significant Difference)  
53 test results, and different letters mean significant difference at 0.05 level. The centre  
54 line and square represent the median value and area-weighted average.

55 **Page S10:** Figure S6. Comparison of CH<sub>4</sub> fluxes measured using the *FCs* with gauze  
56 (*FCs-G*) and without gauze (*FCs-NG*) from the aquaculture ponds in the Min River  
57 Estuary during the aquaculture period. Data are after Yang et al. [*unpublished data*]  
58 for reference and review only. Values represent the means of nine replicates samples,  
59 while the vertical lines indicate standard errors.

60 **Page S11:** Table S1. Characteristics of the three aquaculture ponds in the Min River  
61 Estuary.

62 **Page S12:** Table S2. The best GLS model (lowest AIC values) with the CH<sub>4</sub> fluxes as  
63 functions of environmental values.

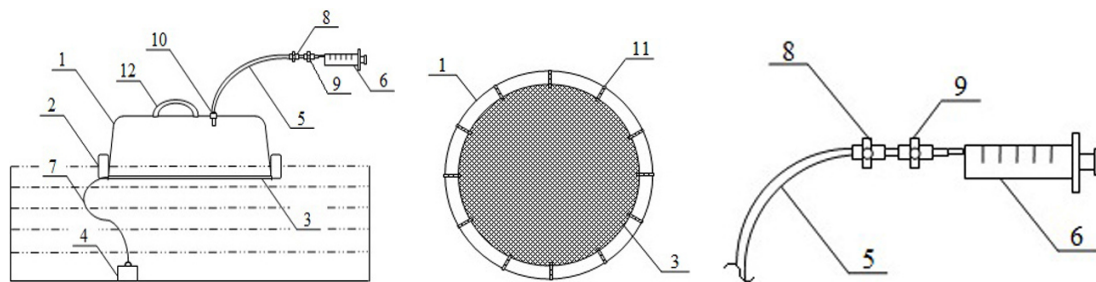
64 **Page S15:** Table S3. Pearson correlation coefficients for environmental variables,  
65 dissolved CH<sub>4</sub> concentration and CH<sub>4</sub> diffusive fluxes from the aquaculture ponds in  
66 Min River Estuary during the aquaculture period.



67

68 **Figure S1.** Location of the study area and sampling sites at aquaculture ponds in Min

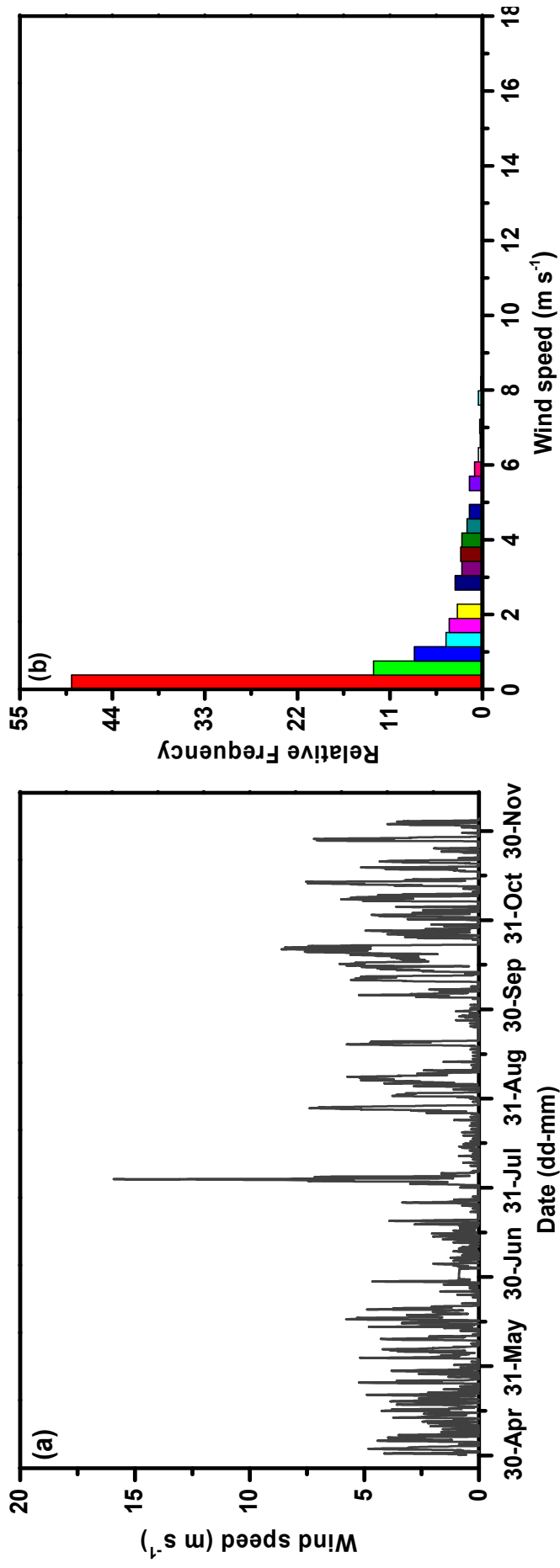
69 River Estuary, Southeast China.



70

71 **Figure S2.** Schematic diagram for the gas sampling device of CH<sub>4</sub> diffusive flux  
 72 across the water-air interface. Numbers 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, and 12  
 73 represents chambers body, Neoprene floats, thin gauze, mooring anchor, sampling  
 74 tube, 60-mL plastic syringes equipped with three-way stopcocks, fixed rope, valve  
 75 body, valve body, gas collecting hole, ribbon, and handle, respectively.

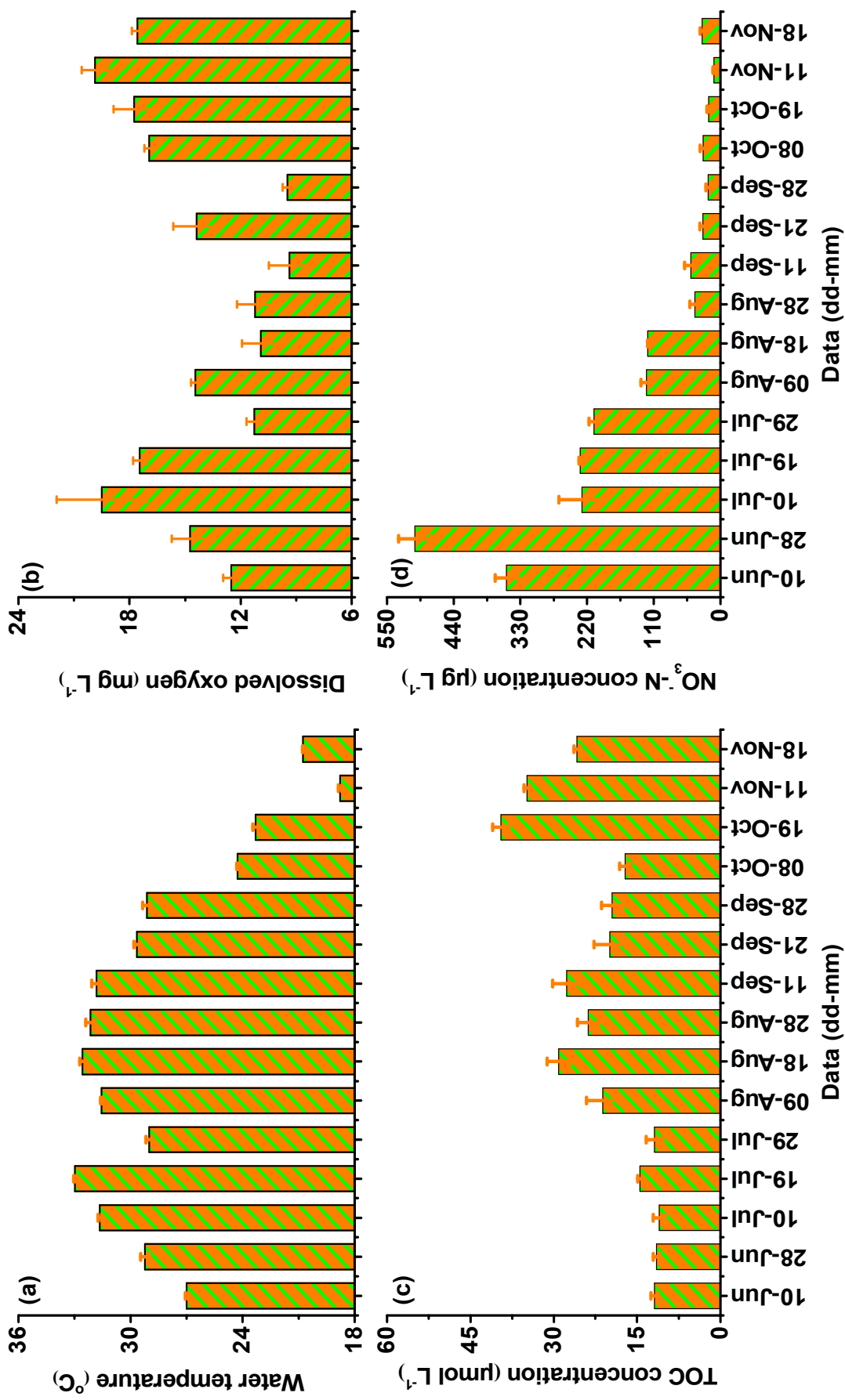




76

77 **Figure S3.** (a) Temporal variation in the wind speed ( $W_s$ ), and (b) frequency distribution of wind speed at the shrimp ponds in the Min River

78 Estuary during the aquaculture period.

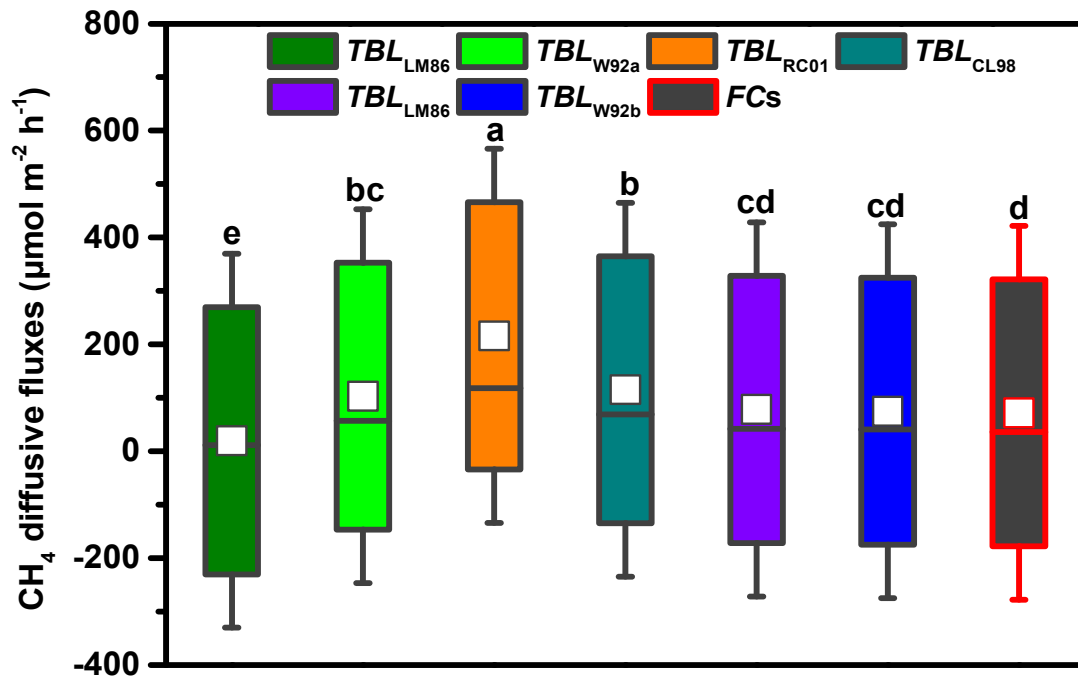


79

80 **Figure S4.** Temporal variation in (a) water temperature, (b) dissolved oxygen, (c) TOC, and (d)  $\text{NO}_3\text{-N}$  in the surface water (20 cm depth)

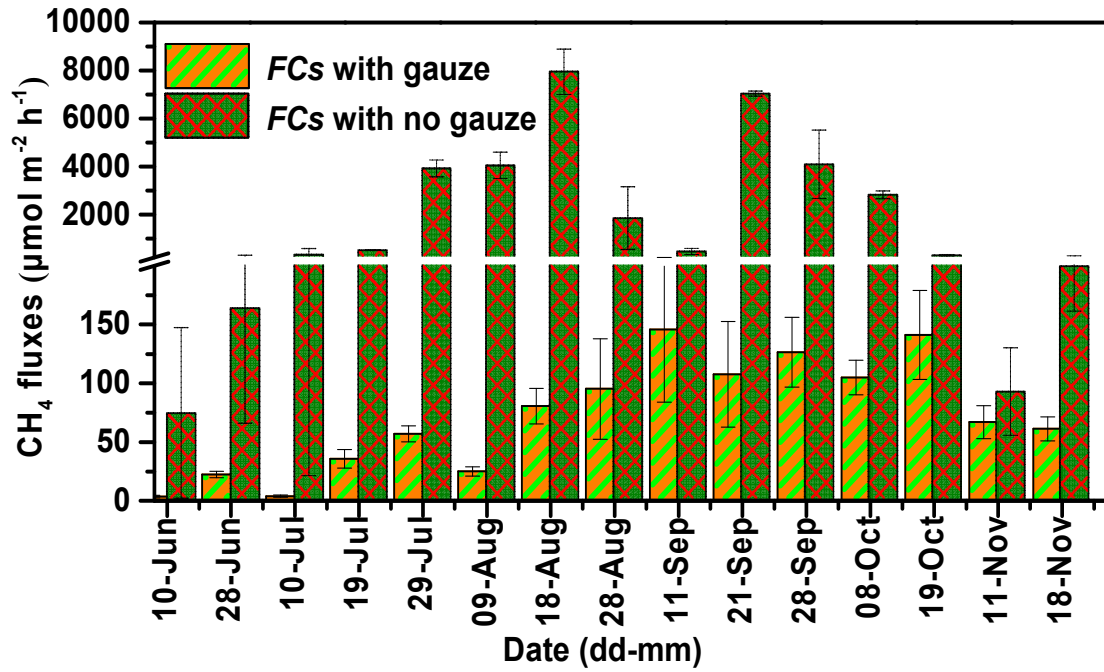
81 of the aquaculture ponds in the Min River Estuary during the aquaculture period. *Error bars* represent standard error ( $n = 9$ ). Data are after

82 Yang et al. [*unpublished data*] for reference and review only.



83

84 **Figure S5.** Boxplots of CH<sub>4</sub> diffusive fluxes estimated using the *TBL* and *FCs*  
 85 methods at the aquaculture ponds in Min River Estuary during the aquaculture period.  
 86 The letters above the boxes represent the LSD (Least Significant Difference) test  
 87 results, and different letters mean significant difference at 0.05 level. The centre line  
 88 and square represent the median value and area-weighted average.



89

90 **Figure S6.** Comparison of CH<sub>4</sub> fluxes measured using the *FCs* with gauze (*FCs-G*)  
 91 and without gauze (*FCs-NG*) from the aquaculture ponds in the Min River Estuary  
 92 during the aquaculture period. Data are after Yang et al. [*unpublished data*] for  
 93 reference and review only. Values represent the means of nine replicates samples,  
 94 while the vertical lines indicate standard errors.

95 **Table S1** Characteristics of the three aquaculture ponds in the Min River Estuary. \*.

Parameters	Pond-I	Pond-II	Pond-III
<b>Shrimp species</b>	<i>Litopenaeus vannamei</i>	<i>Litopenaeus vannamei</i>	<i>Litopenaeus vannamei</i>
<b>Water depth (m)</b>	1.3 (0.3 – 1.7)	1.7 (0.5 – 2.1)	1.5 (0.3 – 1.8)
<b>Water salinity (‰)</b>	3.6 (1.5 – 6.8)	2.2 (1.7 – 5.2)	2.8 (1.7 – 5.4)
<b>Surface area (m<sup>2</sup>)</b>	21426.94	18412.89	19112.71
<b>Stocking density (PL m<sup>-2</sup>)<sup>a</sup></b>	150	120	119
<b>Survival rate (%)<sup>a</sup></b>	45	62	59
<b>Feed conversion rate<sup>b</sup></b>	3.5	2.3	2.5

96 \* Based on Zhang et al. (2019).

97 <sup>a</sup> The data for the stocking density, survival rate, and yield were provided by the farmers; <sup>b</sup> Feed conversion rate =

98 dry weight of feeds added / wet weight of shrimps produced.

99 **Table S2** The best GLS model (lowest AIC values) with the CH<sub>4</sub> fluxes as functions of environmental values.

Variables	Model	Model $R^2$ and $P$ -value	Model independent factors statistics				
$TBL_{LM86}$	gls(CH4flux ~ disCH4 + U10 + watertemp + NO3N, data=dades, method ="REML")	$R^2=0.47$ $P<0.0001$	Coefficients: (Intercept) Estimate Std. Error t-value p-value disCH4 0.0300 0.00399 7.54 <0.0001 U10 0.0515 0.0167 3.08 0.0025 watertemp 0.0187 0.00553 3.39 0.00093 NO3N -0.786 0.178 -4.41 <0.0001				
$TBL_{W92a}$	gls(CH4flux ~ disCH4 + U10 + watertemp + NO3N, data=dades, method ="REML")	$R^2=0.50$ $P<0.0001$	Coefficients: (Intercept) Estimate Std. Error t-value p-value disCH4 0.141 0.0227 6.20 <0.0001 U10 0.614 0.0951 6.45 <0.0001 watertemp 0.122 0.0314 3.89 <0.0001 NO3N -4.64 1.014 -4.58 <0.0001				
$TBL_{RC01}$	gls(CH4flux ~ disCH4 + U10 + watertemp + NO3N, data=dades, method ="REML")	$R^2=0.54$ $P<0.0001$	Coefficients: (Intercept) Estimate Std. Error t-value p-value disCH4 0.331 0.0393 8.42 <0.0001 U10 0.464 0.165 2.81 0.0057 watertemp 0.178 0.0546 3.26 0.0014 NO3N -8.18 1.76 -4.65 <0.0001				
$TBL_{CL98}$	gls(CH4flux ~ disCH4 + NO3N, data=dades, method ="REML")	$R^2=0.53$ $P<0.0001$	Coefficients: (Intercept) Estimate Std. Error t-value p-value disCH4 0.331 0.0393 8.42 <0.0001 U10 0.464 0.165 2.81 0.0057 watertemp 0.178 0.0546 3.26 0.0014 NO3N -8.18 1.76 -4.65 <0.0001				

	method ="REML")	$P < 0.0001$	Estimate	Std. Error	t-value	p-value
<i>TBL<sub>W92b</sub></i>	gls(CH4flux ~ disCH4 + U10 + NO3N, data=dades, method ="REML")	$R^2=0.46$ $P < 0.0001$	(Intercept) 1.45 disCH4 0.181 NO3N -3.59	0.212 0.0203 0.862	6.86 8.95 -4.16	<0.0001 <0.0001 <0.0001
<i>TBL<sub>CW03</sub></i>	gls(CH4flux ~ disCH4 + U10 + NO3N, data=dades, method ="REML")	$R^2=0.48$ $P < 0.0001$	Coefficients: (Intercept) 0.348 disCH4 0.108 U10 0.287 NO3N -2.90	Std. Error 0.244 0.0163 0.0676 0.695	t-value 1.43 6.61 4.24 -4.18	p-value 0.16 <0.0001 <0.0001 0.0001
FCs	gls(CH4flux ~ disCH4 + U10 + NO3N, data=dades, method ="REML")	$R^2=0.35$ $P < 0.0001$	Coefficients: (Intercept) 0.178 disCH4 0.0988 U10 0.340 NO3N -2.79	Std. Error 0.230 0.0153 0.0634 0.652	t-value 0.774 6.46 5.35 -4.27	p-value 0.44 <0.00019 <0.0001 <0.0001
	gls(CH4flux ~ disCH4 + U10 + NO3N, data=dades, method ="REML")		Estimate 0.258 0.0784 0.324 -2.71	Std. Error 0.263 0.0175 0.0728 0.748	t-value 0.98 4.47 4.46 -3.63	p-value 0.33 <0.0001 <0.0001 0.00041

100 **Table S3** Pearson correlation coefficients for environmental variables, dissolved CH<sub>4</sub> concentration and CH<sub>4</sub> diffusive fluxes from the  
101 aquaculture ponds in Min River Estuary during the aquaculture period.

Environmental variables	CH <sub>4</sub> diffusive fluxes		
	<i>TBL<sub>LM86</sub></i>	<i>TBL<sub>W92a</sub></i>	<i>TBL<sub>RC01</sub></i>
		<i>TBL<sub>CL98</sub></i>	<i>TBL<sub>W92b</sub></i>
			<i>TBL<sub>CW03</sub></i>

<b>Meteorological parameters</b>								
Air temperature	NS	<b>0.176*</b>	NS	<b>0.224**</b>	<b>0.253**</b>	<b>0.254**</b>	NS	NS
$U_{10}$	NS	NS	NS	NS	NS	NS	NS	NS
Atmospheric pressure	NS	NS	NS	NS	NS	NS	NS	NS
<b>Water parameters</b>								
Water temperature	NS	NS	NS	NS	NS	NS	NS	NS
Dissolved oxygen (DO)	NS	NS	NS	NS	NS	NS	NS	NS
TOC concentration	<b>0.301**</b>	<b>0.328**</b>	<b>0.306**</b>	<b>0.263**</b>	<b>0.347**</b>	<b>0.355**</b>	NS	NS
NO <sub>3</sub> <sup>-</sup> -N concentration	<b>-0.439**</b>	<b>-0.384**</b>	<b>-0.464**</b>	<b>-0.502**</b>	<b>-0.455**</b>	<b>-0.444**</b>	NS	NS
CH <sub>4</sub> concentration	<b>0.619**</b>	<b>0.522**</b>	<b>0.659**</b>	<b>0.688**</b>	<b>0.572**</b>	<b>0.553**</b>	NS	NS
Gas transfer velocity ( $k_x$ )	NS	NS	NS	NS	NS	NS	NS	NS

102

103

The symbols \* and \*\* indicate significant correlations at the 0.05 and 0.01 levels, respectively. NS indicates non-significant.



104 **References**

- 105 Zhang, Y.F., Yang,P., Yang,H., Tan,L.S., Guo,Q.Q., Zhao,G.H., Li,L., Gao, Y.C., Tong, C., 2019.  
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107 at aquaculture shrimp ponds in a subtropical estuary. Environ. Sci. Pollut. R. 26, 5623-5637.  
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