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Paper:

Choi, J., Song, T., Kwon, J., Lee, S., Han, H., Roy, N., Terashima, C., Fujishima, A., Paik, U. et. al. (2018). WO 3 Nanofibrous Backbone Scaffolds for Enhanced Optical Absorbance and Charge Transport in Metal Oxide (Fe₂O₃, BiVO₄) Semiconductor Photoanodes towards Solar Fuel Generation. *Applied Surface Science*
<http://dx.doi.org/10.1016/j.apsusc.2018.03.167>

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Accepted Manuscript

Full Length Article

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PII: S0169-4332(18)30854-7
DOI: <https://doi.org/10.1016/j.apsusc.2018.03.167>
Reference: APSUSC 38915

To appear in: *Applied Surface Science*

Received Date: 13 September 2017
Revised Date: 28 January 2018
Accepted Date: 21 March 2018

Please cite this article as: J. Choi, T. Song, J. Kwon, S. Lee, H. Han, N. Roy, C. Terashima, A. Fujishima, U. Paik, S. Pitchaimuthu, WO₃ Nanofibrous Backbone Scaffolds for Enhanced Optical Absorbance and Charge Transport in Metal Oxide (Fe₂O₃, BiVO₄) Semiconductor Photoanodes towards Solar Fuel Generation, *Applied Surface Science* (2018), doi: <https://doi.org/10.1016/j.apsusc.2018.03.167>

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WO₃ Nanofibrous Backbone Scaffolds for Enhanced Optical Absorbance and Charge Transport in Metal Oxide (Fe₂O₃, BiVO₄) Semiconductor Photoanodes towards Solar Fuel Generation

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Abstract:

Producing clean fuel (O_2 and H_2) using semiconductors through solar driven water splitting process has been considered as a promising technology to mitigate the existing environmental issues. Unlike the conventional single photoabsorbers, heterostructured semiconductors exhibit the merits of improved solar light photon harvesting and rapid charge separation, which are anticipated to result in high quantum yield of solar fuel generation in photoelectrochemical (PEC) cells. In this report, we demonstrate the electrospun derived WO_3 backbone fibrous channel as heteropartner to the primary photoabsorber (Fe_2O_3 and $BiVO_4$) for promoting the electron transport from charge injection point to charge collector as well as photoholes to the electrolyte. We examine structure, optical, photoelectrochemical and charge transfer property of Fe_2O_3/WO_3 and $BiVO_4/WO_3$ electrodes. These results were compared with directly coated Fe_2O_3 and $BiVO_4$ photoabsorber onto conducting substrate without WO_3 backbone. The optical results showed that the absorbance and visible light activity of Fe_2O_3 and $BiVO_4$ is significantly improved by WO_3 backbone fibers due to high amount of photo absorber loading. In addition, one dimensional (1-D) WO_3 fibers beneficially enhance the optical path length to the photoanode through light scattering mechanism. The electrochemical impedance analysis exhibits WO_3 nanofiber backbone reduces charge transfer resistance at Fe_2O_3 and $BiVO_4$ by rapid charge collection and charge separation compare to backbone-free Fe_2O_3 and $BiVO_4$. As a result, Fe_2O_3/WO_3 and $BiVO_4/WO_3$ fibrous hetero interface structures showed fourfold higher photocurrent generation from PEC cell.

Keywords: Photoelectrocatalyst, WO_3 fiber, Fe_2O_3 , $BiVO_4$, Electrochemical impedance; Solar fuel

1. Introduction:

Photoelectrochemical (PEC) technique is promising and attracts a great deal of attention in several applications such as water splitting fuel generation, [1, 2], solar rechargeable battery, [3] chemical synthesis, [4, 5] organic pollutant degradation, [6, 7] and biosensing [8, 9]. Typically, a photoactive semiconductor or molecular sensitizer are utilized as oxidation catalyst in the PEC process. The photoirradiated semiconductor produces photocharge carriers such as electron (e^-) and holes (h^+), which are separated and transport to the respective terminals of cathode and electrolyte by applying a small electric potential from outside. The kinetic energy of photoholes at the valence band of semiconductor will drive oxidation process at electrode/electrolyte interfaces. In general, the valence band energy position of semiconductor in the photoanode should lie higher than that of water oxidation potential ($E_g > 1.2$ V vs RHE) to drive the PEC water oxidation process [10]. Titanium dioxide (TiO_2) is one of the well documented PEC oxidation materials as it possesses appropriate VB position towards water oxidation, excellent chemical stability and photocatalytic activity. However, their UV activated band gap energy ($E_g \sim 3.2$ eV) is inadequate to demonstrate the solar light driven PEC process as it absorbs only 5% light photons from visible light region [11]. As a consequence, narrow band gap energy semiconductor materials are proposed to replace the TiO_2 in solar light driven PEC oxidation process. Recently, hematite (Fe_2O_3) [12-16], and bismuth vanadate ($BiVO_4$) [17-19] perceived profound attention in visible light driven water oxidation process. It is well known that the thickness of the photoabsorber coating (photoanode) dictates light absorbance quantity. But, the short hole diffusion length characteristics of Fe_2O_3 (2- 4 nm) [20] and $BiVO_4$ (60-200 nm) [21, 22] at electrode/electrolyte interface could limit the thickness of the photoanode as well affect the surface-mediated and/or internal electron/hole recombination

and thus, could ultimately affect PEC efficiency.

To promote the charge separation rate at semiconductor/electrolyte interfaces, multiple routes were proposed on semiconductor modification through a) metal carriers doping [23, 24], b) co-catalyst decoration [25], c) hetero partner assembly [26] and d) inserting interfacial layer [27]. Mostly, these protocols facilitate the charge separation at Fe_2O_3 and BiVO_4 layers through passivating the surface states, and grain boundaries responsible for charge recombination at electrode/electrolyte interfaces [28-31]. However, overcoming the photoanode thickness dependent hole-transport still remains a great challenge. Instead of coating the photoelectrocatalyst layer (guest) directly onto the substrates, it is anticipated that assembling onto nanoscale, wide-pore structured backbone scaffold (host) that are spatially connected to the substrate could simultaneously improve the charge collection and charge transport at guest layer/electrolyte interfaces [32]. In this line, we are demonstrating tungsten oxide (WO_3) as an appropriate backbone scaffold choice for the Fe_2O_3 and BiVO_4 guest layers as its conduction band is lower than these materials. Sivula et al [33] demonstrated that Fe_2O_3 guest layer coated onto WO_3 nanostructured host scaffold showed effective charge separation than directly coated Fe_2O_3 onto substrate. On the other hand, BiVO_4 primary photoabsorber shell layer coated WO_3 nanowire backbone layer markedly enhanced the charge separation at BiVO_4 /electrolyte interfaces [34]. In literature, similar type of WO_3 backbone scaffold has been progressed in the form of nanotube, inverse opal, and nanowire [35, 36]. Mostly, the demonstrated WO_3 scaffold structures in the literature exhibits limited pore-size (5-10 nm), which may be inadequate in either producing homogenous guest photoabsorber coating or sufficient space availability for the electrolyte percolation at the electrode surface.

Here we propose highly interconnected, wide-pore structured electrospun WO_3 nanofibers as a backbone scaffold for Fe_2O_3 and BiVO_4 guest layers coating. Compared to

the vacuum based physical techniques, the electrospinning technique is simple, economic and displays large scale viability. For instance, 1-D nanostructures prepared through chemical vapor deposition, and VLS growth technique required sophisticated environment including high temperature processing, high purity of chemical precursor, and long processing duration. In the case of anodization technique, it has major limitation on assembling 1-D nanostructures on metal substrates only. In this view, electrospinning process is a simple route for assembling 1-D nanostructures. It can be operated at room temperature, less processing time and readily collected onto substrates.

To the best of our knowledge, for the first time, systematically prepared Fe_2O_3 and BiVO_4 coating onto WO_3 fibers as illustrated in **Figure 1** is demonstrated in solar fuel generation. Identical amount of precursor is loaded onto both FTO substrate (**Figure 1a**) and WO_3 fibers coated FTO substrates (**Figure 1b**). Subsequently, these samples were sintered at $500\text{ }^\circ\text{C}$, and thus formed thin conformal guest layer coating. In view of light penetration at different portion of the photoanode (top and bottom layer) the photocharge carrier separation from guest layer to electrolyte and substrate seems to be different. For instance, in the case of direct coating protocol (**Figure 1a**), the photoholes transport distance (indicated as x_2 in Figure 1a) from closer to the electrolyte position is efficient than underneath layer position (indicated as x_1 in **Figure 1a**). At WO_3 scaffold based protocol (**Figure 1b**), the photoelectrons from guest layer is radially transport to the charge collector, thus hole transport distance from guest layer to the electrolyte is approximately identical ($x_1 = x_2$). Therefore, regardless of the light penetration position the photoholes can passage to the electrolyte efficiently. The structure, optical, electrochemical and PEC properties are systematically examined, which reveals the advantage of WO_3 backbone scaffold towards improving the charge separation and charge transport at guest photoelectrocatalyst/electrolyte interface in water oxidation reactions.

2. Experimental

2a. WO₃ nanofibrous electrode fabrication: WO₃ nanofiber (NF) layer was fabricated by electrospinning method. 0.6 g of tungsten (VI) ethoxide (Alfa Aesar) was solved in 2.5 mL of N,N-dimethylformamide (99.8%, Sigma Aldrich) and sonicated for 30 min. And then, 0.15 g of acetic acid (99.5%, Samchun Chemical) and 0.5 g of poly(vinyl acetate) (M_w ~500,000, Sigma Aldrich) were mixed with the solution and stirred overnight. The NFs were electrospun onto the pre-cleaned fluorinated tin oxide (FTO) substrates using the resulted solution at a DC voltage of 20 kV with a flow of 0.2 ml/h. Finally, the as-spun nanofiber layer was annealed at 500 °C for 3 h in air. In order to overcome the fibrous films peel off from substrate during sintering process, the WO₃ fiber films were kept under hot press technique at 100 °C for 2 minutes. During the hot pressing pre-treatment, the PVA polymer binder was melted and facilitate the WO₃ fibers attached onto substrate.

2b. Synthesis of the BiVO₄ and Fe₂O₃ layer: The coating solution of BiVO₄ was prepared as follows: 0.1462 g of ammonium metavanadate (99%, Sigma Aldrich), 0.6061 g of bismuth nitrate pentahydrate (98%, Sigma Aldrich), 0.4803 g of citric acid (anhydrous, Sigma Aldrich), and 0.825 g of nitric acid (60%, Daejung Chemical) were mixed into 2.925 mL of distilled water. For the Fe₂O₃ solution, 0.505 g of iron nitrate nonahydrate (98%, Sigma Aldrich), 0.4803 g of citric acid (anhydrous, Sigma Aldrich), and 0.825 g of nitric acid (60%, Daejung Chemical) were dissolved in 2.925 mL of distilled water. The BiVO₄ layer and Fe₂O₃ layers were formed by spin-coating the solutions at 2000 rpm for 30s onto WO₃ nanofiber layer, followed by annealing at 500 °C for 3 h in air. To compare the photoanode performance of BiVO₄ and Fe₂O₃ in the presence and absence of WO₃ backbone fiber in the PEC cells, similar quantity of BiVO₄ and Fe₂O₃ precursor solution as is explained above was coated onto FTO directly. The resultant films were annealing at 500 °C for 3 h in air.

2c. Characterization: The crystalline structure of electrospun WO_3 film was studied using an X-ray diffractometer (XRD, New D8 Advance, Bruker). The surface morphology of the semiconductor films was analyzed by field emission scanning microscopy (FESEM, JSM-7600F, JEOL). The heterostructure formation at $\text{BiVO}_4/\text{WO}_3$ sample was analyzed by high resolution transmission electron microscopy (JEM-2100F, JEOL). The optical absorbance spectra of the resultant semiconductor films were recorded using a UV-Vis spectrophotometer (V670, JASCO) in the diffuse reflectance mode. Note that, the incident light beam was allowed through the substrate side.

2d. Photoelectrochemical studies: The photoelectrochemical measurements were carried out using an FRA-equipped potentiostat (PGSTAT 302N, Autolab). The standard three-electrode configuration encompass with semiconductor layers coated onto FTO glass as the working electrode, Ag/AgCl reference electrode and a platinum foil counter electrode were used in these experiments. The photoelectrochemical performance was evaluated using a 0.5 M Na_2SO_4 aqueous solution-based electrolyte. The pH of the solution was 6. Prior to the PEC experiments, the electrolyte was purged with nitrogen gas for 30 min. A 300 W Xe lamp (6258, Newport) with AM 1.5 and an IR cut filter was used as the light source. The current was recorded from PEC cell under chronoamperometric mode at constant applied potential 0.7 V Vs Ag/AgCl. At the consecutive light on/off cycles, respective output current was recorded. The electrochemical impedance spectroscopy (EIS) was used to examine the charge transfer characteristics of the electrode/electrolyte interfaces. The Nyquist plots were recorded in the frequency range between 0.1 Hz–100 kHz at an amplitude of 20 mV using a potentiostat (PGSTAT 302N, Autolab).

3. Results and discussion

The X-ray diffraction (XRD) pattern of as-synthesized electrospun WO_3 membrane onto FTO substrate is presented in **Figure 1a**. The significant peaks appearing at 23.1° , 23.6° , and 24.4° correspond to the (002), (020), and (200) planes of monoclinic WO_3 , respectively (JCPDS 01-083-0950). Other significant peaks at 33.6° , 37.5° and 51.1° represents (202), (103) and (114) crystalline planes of WO_3 [37, 38] [27]. The BiVO_4 film coated onto FTO substrate showed monoclinic scheelite crystalline structure (PDF 00-014-0688). In this sample, strong crystalline peaks observed at 28.4° , 30.8° and 50.5° corresponds to (-130), (040), and (202) crystalline planes of monoclinic scheelite structure[39]. The Fe_2O_3 thin film coated onto FTO conducting substrate, the crystalline peaks of FTO (indicated as *) is dominated than Fe_2O_3 . However, a small peak exhibits at 54.1° endorse the (116) crystalline phase of hematite structure (JCPDS- 01-086-0550). In the case of BiVO_4 and Fe_2O_3 layer coated onto WO_3 fibers, similar crystalline peaks were observed as is in the backbone-free films. These results ensure the post deposition of BiVO_4 and Fe_2O_3 layer and sustain their growth on WO_3 fiber surface.

The surface morphology of the Fe_2O_3 , BiVO_4 and WO_3 thin films coated onto FTO substrates are presented in **Figure 3 (a) – (c)**. From **Figure 3 (a)**, the Fe_2O_3 film exhibits thin layer growth and strong features of FTO quasi crystals is observed in the background. In the case of BiVO_4 thin films, it shows mosaic morphology with pores formation around grain boundaries. The similar porous BiVO_4 films is reported by other researchers [40-42]. The **Figure 3 (c)** SEM image of electrospun WO_3 sample reveals highly interconnected fibrous channels with $\sim 100\text{-}150$ nm diameter. In between the WO_3 fibrous channel ample room is available for growing Fe_2O_3 and BiVO_4 thin films. As is expected, the **Figure 3 (d) and (e)** shows that the WO_3 fibrous surface is completely filled with Fe_2O_3 and BiVO_4 . It implies that WO_3 backbone fiber scaffold enhance film integrity of Fe_2O_3 and BiVO_4 coating compared to their individual films coating directly onto FTO substrates. In order to ensure the

Fe_2O_3 and BiVO_4 formation onto WO_3 fibers, elemental mapping analysis is carried out and the corresponding results were presented in **Figure S1** and **S2** (See supporting information). The Fe and O constitutes from **Figure S1**, as well as Bi, V, and O species observed from **Figure S2** endorse the formation of Fe_2O_3 and BiVO_4 films onto WO_3 fibers. Further, ensure the heterostructure formation at primary photoabsorber coated WO_3 backbone fiber, we randomly analyze $\text{BiVO}_4/\text{WO}_3$ sample using high resolution transmission electron microscopy (HRTEM). The **Figure 4** shows the HRTEM image of $\text{BiVO}_4/\text{WO}_3$ sample. From **Figure 4 (a)**, the primary photoabsorber of BiVO_4 is completely covered as thin layer onto WO_3 fiber surface. Further examining at 10 nm scale (**Figure 4 (b)**), a thin layer of BiVO_4 is coated onto (002) crystalline phase of WO_3 fiber surface. These results are in line with XRD and elemental mapping results on $\text{BiVO}_4/\text{WO}_3$ composite.

The optical absorption spectrum of Fe_2O_3 and BiVO_4 films in the presence and absence of WO_3 backbone nanofiber is presented in **Figure 5a**. The onset light absorbance at backbone-free Fe_2O_3 and BiVO_4 films are found to be at around ~590 nm, and ~490 nm, respectively. Though, similar quantity of Fe_2O_3 and BiVO_4 precursor coated onto WO_3 fiber compared to direct coating onto substrate as depicted in **Scheme 1**, the resultant optical absorbance of Fe_2O_3 and BiVO_4 is strikingly enhanced (**Figure 5a**) at visible light wavelength. The optical absorbance of Fe_2O_3 is enhanced in the regions between ~340 and 535 nm by WO_3 fiber. In the case of BiVO_4 films, the onset light absorbance is shifted from 470 nm to ~480 nm. This might be attributed to the improvement of BiVO_4 crystal growth on WO_3 fiber surface compare to direct growth onto substrate. Overall, the backbone WO_3 fiber (host) promotes the light absorbance of Fe_2O_3 and BiVO_4 (guest) at visible light wavelength due to high quantity of loading. For instance, optical absorption quantity of single coating cycle of Fe_2O_3 and BiVO_4 films onto WO_3 fiber is slightly higher than that of three coating cycles of individual Fe_2O_3 and BiVO_4 films onto FTO substrate (**Figure S3**, see supporting

information). Similar optical absorbance enhancement is observed at WO_3 based heterostructured semiconductors by other researchers [34, 43, 44]. One of the other plausible reasons for light absorbance enhancement is multiple light scattering contributed from one dimensional WO_3 fibers. From optical reflectance spectra (**Figure 5b**), it is understood that ~55% of input light is reflected by WO_3 fibers which enhance the optical path length at photoanode. The light scattering from backbone fiber to the primary photoabsorber Fe_2O_3 and BiVO_4 may facilitate more photocharge carrier generation towards photoelectrochemical reaction.

To understand the role of WO_3 backbone fiber in charge separation of photocharge carriers at primary Fe_2O_3 and BiVO_4 photoabsorber layer/electrolyte interface, the charge transport characteristics are studied using electrochemical impedance spectroscopy (EIS). The EIS provides the insights into charge separation (photoelectron and photoholes) as well as charge transport (photoelectrons) at Fe_2O_3 or $\text{BiVO}_4/\text{WO}_3$ hetero interface as well as Fe_2O_3 or $\text{BiVO}_4/\text{WO}_3$ fiber/electrolyte interface [45, 46]. Typical Nyquist plots of PEC cell using Fe_2O_3 and BiVO_4 photoanode in the presence and absence of WO_3 backbone fiber layer studied at dark and light irradiation condition is summarized in **Figure 6 (a) and (b), respectively**. The obtained Nyquist plot is simulated with the equivalent circuit shown in the **Figure S4** (see supporting information). In the equivalent circuit, R_s indicates the sheet resistance of the charge collector (FTO), CPE indicates the constant phase element, and R_{ct} indicates the charge transfer resistance at the electrode/electrolyte interface. Under dark condition the diameter of semicircle in Nyquist plot indicates the charge transfer resistance (R_{ct}) i.e. electron transport from outside circuit to the electrolyte. This explain the capability of electron conduction through solid film. For instance, the film possessing high electron conduction results less charge transfer resistance. The estimated charge transfer resistance (R_{ct}) value of WO_3 back bone nanofiber from **Figure 6 (a)** is found to be ~8606 Ω . The

simulated R_{ct} values of Fe_2O_3 and BiVO_4 films are found to be $1.54 \times 10^5 \Omega$ and $1.37 \times 10^5 \Omega$, respectively. Surprisingly, the R_{ct} values of $\text{Fe}_2\text{O}_3/\text{WO}_3$ and $\text{BiVO}_4/\text{WO}_3$ films are markedly reduced to $\sim 64530 \Omega$ and $\sim 52367 \Omega$, respectively. It clearly implies that the weak electron transport at Fe_2O_3 and BiVO_4 films are improved markedly by highly conducting WO_3 backbone fibers, which in turn reduces the charge transfer resistance. The diameter of the semicircle in **Figure 6 (b)** obtained under applied potential in associate with light irradiation indicates the charge transfer resistance of photocharge carrier separation at electrode/electrolyte interfaces. Here the photoholes transport is also involved in addition to the photoelectrons. Briefly, the photocharge carrier electron and holes could reach the respective terminals. The photoelectrons from conduction band (CB) of photoabsorber (Fe_2O_3 or BiVO_4) will transfer to charge collector (FTO). Conversely, the photoholes from valence band (VB) of photoabsorber to the electrolyte (**Figure 7a**). In the case of n type photoabsorbers Fe_2O_3 or BiVO_4 they are effectively conducting photoelectrons from CB to charge collector but inadequate in photohole transfer to the electrolyte. From **Figure 6b**, estimated charge transfer resistance value of Fe_2O_3 and BiVO_4 is markedly reduced in the presence of WO_3 back bone nanofibers. In particular, R_{ct} value of BiVO_4 film is reduced three folds from 10^5 order to 10^2 (inset of **Figure 6 (b)**). This implies that highly interconnected WO_3 back bone nanofibers collect the photoelectrons from point of charge injection at primary photoabsorber Fe_2O_3 or BiVO_4 to the charge collector. This enhanced charge collection from Fe_2O_3 or $\text{BiVO}_4 / \text{WO}_3$ interfaces lead effective charge separation at electrode/electrolyte interfaces.

Further, the photoelectrochemical property of Fe_2O_3 and BiVO_4 in the presence and absence of WO_3 films were examined through chronoamperometry plots. Under dark and light condition (on/off), the current generation from PEC cell is recorded and summarized in **Figure 7b**. Comparing the overall photocurrent generation, the WO_3 fiber backbone assisted

Fe_2O_3 and BiVO_4 photoanodes showed two-fold higher photocurrent generation compare to that of backbone-free films. As discussed in the impedance analysis, it clearly advocates that backbone fibers facilitate the electron transport to the charge collector and charge separation at Fe_2O_3 and BiVO_4 /electrolyte interfaces. Thus, the effective charge separation through fiber backbone, and efficient light harvesting at visible light region results high photocurrent generation at Fe_2O_3 and BiVO_4 films compare to WO_3 backbone-free films. Based on the above experimental discussion, the schematic structure of Fe_2O_3 or BiVO_4 / WO_3 heterointerface is illustrated in **Figure 7a** [47, 48]. Though the observed photocurrent is lower than that of previous reports on heterostructure photoanodes such as $\text{Fe}_2\text{O}_3/\text{WO}_3$, [33, 49, 50] and $\text{BiVO}_4/\text{WO}_3$ [51] [52], which can be improved by increasing the electrode thickness of WO_3 fibrous electrode. It is anticipated that high filling of fibrous network will afford more room for loading primary photoabsorber (Fe_2O_3 or BiVO_4). Furthermore, three dimensionally filled WO_3 back bone fiber channels can facilitate the rapid photoelectron transport from the entire electrode to charge collector lead thickness independent photocurrent generation at Fe_2O_3 or BiVO_4 films. Therefore, photocurrent results on WO_3 scaffold based films encourages to revisit the photoanode architecture with fibrous backbones instead of directly coated flat type thin films. It is worthy to mention that due to the difference in thickness of the $\text{Fe}_2\text{O}_3/\text{WO}_3$ and $\text{BiVO}_4/\text{WO}_3$ electrodes, we couldn't compare the performance between them. However, the heterostrutured $\text{Fe}_2\text{O}_3/\text{WO}_3$ and $\text{BiVO}_4/\text{WO}_3$ films showed high photocurrent density and less charge transfer resistance than that of individual Fe_2O_3 and BiVO_4 thin films.

4. Conclusion

In conclusion, merits of WO_3 backbone fiber in charge separation and charge transport at Fe_2O_3 and BiVO_4 photoanodes were demonstrated. This proof-of-concept based backbone assisted photoanodes showed high light absorbance and charge separation at

electrode/electrolyte interfaces compare to directly coated backbone-free semiconductor films. Furthermore, these highly conductive back bone assisted photoanodes may overcome thickness dependence diffusion length, which need to be studied in detail. In addition, the wide-pore structured fiber-type backbones may support the effective electrolyte percolation at photoanode and can reduce the charge recombination at electrode/electrolyte interfaces. Yet, the conductivity of the WO_3 fibers can be improved by metal doping. It is anticipated that readily available WO_3 fibrous electrodes can be transformed to any other hetero photoabsorbers. The additional coating of co-catalyst [53] will enhance the performance of PEC oxidation reactions, which can be applied solar fuel generation, chemical synthesis and water pollutant treatment applications.

Acknowledgement

One of the lead authors S.P acknowledges Welsh Government and European Regional Development Fund (ERDF) for partial support of this project through Sêr Cymru II-Rising Star Fellowship program (80761-SU-102 (West)).

References

- [1] A. Fujishima, K. Honda, Electrochemical Photolysis of Water at a Semiconductor Electrode, *Nature*, 238 (1972) 37-38.
- [2] R. Raja, P. Sudhagar, A. Devadoss, C. Terashima, L.K. Shrestha, K. Nakata, R. Jayavel, K. Ariga, A. Fujishima, Pt-free solar driven photoelectrochemical hydrogen fuel generation using 1T MoS_2 co-catalyst assembled CdS QDs/ TiO_2 photoelectrode, *Chemical Communications*, 51 (2015) 522-525.
- [3] G. Kim, M. Oh, Y. Park, Solar-rechargeable battery based on photoelectrochemical water oxidation: Solar water battery, *6* (2016) 33400.
- [4] K. Fuku, K. Sayama, Efficient oxidative hydrogen peroxide production and accumulation in photoelectrochemical water splitting using a tungsten trioxide/bismuth vanadate

photoanode, *Chemical Communications*, 52 (2016) 5406-5409.

[5] K. Fuku, Y. Miyase, Y. Miseki, T. Funaki, T. Gunji, K. Sayama, Photoelectrochemical Hydrogen Peroxide Production from Water on a WO₃/BiVO₄ Photoanode and from O₂ on an Au Cathode Without External Bias, *Chemistry – An Asian Journal*, 12 (2017) 1111-1119.

[6] J.M. Kesselman, N.S. Lewis, M.R. Hoffmann, Photoelectrochemical Degradation of 4-Chlorocatechol at TiO₂ Electrodes: Comparison between Sorption and Photoreactivity, *Environmental Science & Technology*, 31 (1997) 2298-2302.

[7] M.S. Koo, K. Cho, J. Yoon, W. Choi, Photoelectrochemical Degradation of Organic Compounds Coupled with Molecular Hydrogen Generation Using Electrochromic TiO₂ Nanotube Arrays, *Environmental Science & Technology*, 51 (2017) 6590-6598.

[8] A. Devadoss, P. Sudhagar, S. Das, S.Y. Lee, C. Terashima, K. Nakata, A. Fujishima, W. Choi, Y.S. Kang, U. Paik, Synergistic Metal–Metal Oxide Nanoparticles Supported Electrocatalytic Graphene for Improved Photoelectrochemical Glucose Oxidation, *ACS Applied Materials & Interfaces*, 6 (2014) 4864-4871.

[9] A. Devadoss, P. Sudhagar, C. Terashima, K. Nakata, A. Fujishima, Photoelectrochemical biosensors: New insights into promising photoelectrodes and signal amplification strategies, *Journal of Photochemistry and Photobiology C: Photochemistry Reviews*, 24 (2015) 43-63.

[10] M.G. Walter, E.L. Warren, J.R. McKone, S.W. Boettcher, Q. Mi, E.A. Santori, N.S. Lewis, Solar Water Splitting Cells, *Chemical Reviews*, 110 (2010) 6446-6473.

[11] P. Sudhagar, T. Song, A. Devadoss, J.W. Lee, M. Haro, S. Gimenez, C. Terashima, V.V. Lysak, J. Bisquert, A. Fujishima, U. Paik, Modulating the interaction between gold and TiO₂ nanowires for enhanced solar driven photoelectrocatalytic hydrogen generation, *Physical Chemistry Chemical Physics*, (2015).

[12] D.K. Bora, A. Braun, E.C. Constable, "In rust we trust". Hematite - the prospective inorganic backbone for artificial photosynthesis, *Energy & Environmental Science*, 6 (2013)

- [13] M.J. Katz, S.C. Riha, N.C. Jeong, A.B.F. Martinson, O.K. Farha, J.T. Hupp, Toward solar fuels: Water splitting with sunlight and “rust”?, *Coordination Chemistry Reviews*, 256 (2012) 2521-2529.
- [14] Y. Qiu, S.-F. Leung, Q. Zhang, B. Hua, Q. Lin, Z. Wei, K.-H. Tsui, Y. Zhang, S. Yang, Z. Fan, Efficient Photoelectrochemical Water Splitting with Ultrathin films of Hematite on Three-Dimensional Nanophotonic Structures, *Nano Letters*, 14 (2014) 2123-2129.
- [15] A. Annamalai, P.S. Shinde, A. Subramanian, J.Y. Kim, J.H. Kim, S.H. Choi, J.S. Lee, J.S. Jang, Bifunctional TiO₂ underlayer for [small alpha]-Fe₂O₃ nanorod based photoelectrochemical cells: enhanced interface and Ti⁴⁺ doping, *Journal of Materials Chemistry A*, 3 (2015) 5007-5013.
- [16] A. Annamalai, A. Subramanian, U. Kang, H. Park, S.H. Choi, J.S. Jang, Activation of Hematite Photoanodes for Solar Water Splitting: Effect of FTO Deformation, *The Journal of Physical Chemistry C*, 119 (2015) 3810-3817.
- [17] T.W. Kim, K.-S. Choi, Nanoporous BiVO₄ Photoanodes with Dual-Layer Oxygen Evolution Catalysts for Solar Water Splitting, *Science*, 343 (2014) 990-994.
- [18] Q. Jia, K. Iwashina, A. Kudo, Facile fabrication of an efficient BiVO₄ thin film electrode for water splitting under visible light irradiation, *Proceedings of the National Academy of Sciences*, 109 (2012) 11564-11569.
- [19] C. Ravidhas, A. Juliat Josephine, P. Sudhagar, A. Devadoss, C. Terashima, K. Nakata, A. Fujishima, A. Moses Ezhil Raj, C. Sanjeeviraja, Facile synthesis of nanostructured monoclinic bismuth vanadate by a co-precipitation method: Structural, optical and photocatalytic properties, *Materials Science in Semiconductor Processing*, 30 (2015) 343-351.
- [20] Y. Yuan, J. Gu, K.-H. Ye, Z. Chai, X. Yu, X. Chen, C. Zhao, Y. Zhang, W. Mai,

Combining Bulk/Surface Engineering of Hematite To Synergistically Improve Its Photoelectrochemical Water Splitting Performance, *ACS Applied Materials & Interfaces*, 8 (2016) 16071-16077.

[21] R.P. Antony, P.S. Bassi, F.F. Abdi, S.Y. Chiam, Y. Ren, J. Barber, J.S.C. Loo, L.H. Wong, Electrospun Mo-BiVO₄ for Efficient Photoelectrochemical Water Oxidation: Direct Evidence of Improved Hole Diffusion Length and Charge separation, *Electrochimica Acta*, 211 (2016) 173-182.

[22] A.J.E. Rettie, H.C. Lee, L.G. Marshall, J.-F. Lin, C. Capan, J. Lindemuth, J.S. McCloy, J. Zhou, A.J. Bard, C.B. Mullins, Combined Charge Carrier Transport and Photoelectrochemical Characterization of BiVO₄ Single Crystals: Intrinsic Behavior of a Complex Metal Oxide, *Journal of the American Chemical Society*, 135 (2013) 11389-11396.

[23] H.S. Park, K.E. Kweon, H. Ye, E. Paek, G.S. Hwang, A.J. Bard, Factors in the Metal Doping of BiVO₄ for Improved Photoelectrocatalytic Activity as Studied by Scanning Electrochemical Microscopy and First-Principles Density-Functional Calculation, *The Journal of Physical Chemistry C*, 115 (2011) 17870-17879.

[24] Y. Ling, Y. Li, Review of Sn-Doped Hematite Nanostructures for Photoelectrochemical Water Splitting, *Particle & Particle Systems Characterization*, 31 (2014) 1113-1121.

[25] C. Du, J. Wang, X. Liu, J. Yang, K. Cao, Y. Wen, R. Chen, B. Shan, Ultrathin CoOx-modified hematite with low onset potential for solar water oxidation, *Physical Chemistry Chemical Physics*, 19 (2017) 14178-14184.

[26] J.H. Kim, G. Magesh, H.J. Kang, M. Banu, J.H. Kim, J. Lee, J.S. Lee, Carbonate-coordinated cobalt co-catalyzed BiVO₄/WO₃ composite photoanode tailored for CO₂ reduction to fuels, *Nano Energy*, 15 (2015) 153-163.

[27] J. Choi, P. Sudhagar, J.H. Kim, J. Kwon, J. Kim, C. Terashima, A. Fujishima, T. Song, U. Paik, WO₃/W:BiVO₄/BiVO₄ graded photoabsorber electrode for enhanced

photoelectrocatalytic solar light driven water oxidation, *Physical Chemistry Chemical Physics*, 19 (2017) 4648-4655.

[28] F. Le Formal, N. Tetreault, M. Cornuz, T. Moehl, M. Gratzel, K. Sivula, Passivating surface states on water splitting hematite photoanodes with alumina overlayers, *Chemical Science*, 2 (2011) 737-743.

[29] C. Du, M. Zhang, J.-W. Jang, Y. Liu, G.-Y. Liu, D. Wang, Observation and Alteration of Surface States of Hematite Photoelectrodes, *The Journal of Physical Chemistry C*, 118 (2014) 17054-17059.

[30] M.N. Shaddad, M.A. Ghanem, A.M. Al-Mayouf, S. Gimenez, J. Bisquert, I. Herraiz-Cardona, Cooperative Catalytic Effect of ZrO₂ and α -Fe₂O₃ Nanoparticles on BiVO₄ Photoanodes for Enhanced Photoelectrochemical Water Splitting, *ChemSusChem*, 9 (2016) 2779-2783.

[31] A. Malathi, V. Vasanthakumar, P. Arunachalam, J. Madhavan, M.A. Ghanem, A low cost additive-free facile synthesis of BiFeWO₆/BiVO₄ nanocomposite with enhanced visible-light induced photocatalytic activity, *Journal of Colloid and Interface Science*, 506 (2017) 553-563.

[32] S. Kment, F. Riboni, S. Pausova, L. Wang, L. Wang, H. Han, Z. Hubicka, J. Krysa, P. Schmuki, R. Zboril, Photoanodes based on TiO₂ and [small alpha]-Fe₂O₃ for solar water splitting - superior role of 1D nanoarchitectures and of combined heterostructures, *Chemical Society Reviews*, 46 (2017) 3716-3769.

[33] K. Sivula, F.L. Formal, M. Grätzel, WO₃-Fe₂O₃ Photoanodes for Water Splitting: A Host Scaffold, Guest Absorber Approach, *Chemistry of Materials*, 21 (2009) 2862-2867.

[34] P.M. Rao, L. Cai, C. Liu, I.S. Cho, C.H. Lee, J.M. Weisse, P. Yang, X. Zheng, Simultaneously Efficient Light Absorption and Charge Separation in WO₃/BiVO₄ Core/Shell Nanowire Photoanode for Photoelectrochemical Water Oxidation, *Nano Letters*,

14 (2014) 1099-1105.

[35] T. Zhang, J. Su, L. Guo, Morphology engineering of WO₃/BiVO₄ heterojunctions for efficient photocatalytic water oxidation, *CrystEngComm*, 18 (2016) 8961-8970.

[36] S.Y. Chae, C.S. Lee, H. Jung, O.-S. Joo, B.K. Min, J.H. Kim, Y.J. Hwang, Insight into Charge Separation in WO₃/BiVO₄ Heterojunction for Solar Water Splitting, *ACS Applied Materials & Interfaces*, 9 (2017) 19780-19790.

[37] T. Zhang, J. Su, L. Guo, Hierarchical architecture of WO₃ nanosheets by self-assembly of nanorods for photoelectrochemical applications, *CrystEngComm*, 18 (2016) 665-669.

[38] T. Zhang, Z. Zhu, H. Chen, Y. Bai, S. Xiao, X. Zheng, Q. Xue, S. Yang, Iron-doping-enhanced photoelectrochemical water splitting performance of nanostructured WO₃: a combined experimental and theoretical study, *Nanoscale*, 7 (2015) 2933-2940.

[39] O. Monfort, T. Roch, M. Gregor, L. Satrapinskyy, D. Raptis, P. Lianos, G. Plesch, Photooxidative properties of various BiVO₄/TiO₂ layered composite films and study of their photocatalytic mechanism in pollutant degradation, *Journal of Environmental Chemical Engineering*, 5 (2017) 5143-5149.

[40] F.M. Toma, J.K. Cooper, V. Kunzelmann, M.T. McDowell, J. Yu, D.M. Larson, N.J. Borys, C. Abelyan, J.W. Beeman, K.M. Yu, J. Yang, L. Chen, M.R. Shaner, J. Spurgeon, F.A. Houle, K.A. Persson, I.D. Sharp, Mechanistic insights into chemical and photochemical transformations of bismuth vanadate photoanodes, *Nature Communications*, 7 (2016) 12012.

[41] J.H. Kim, Y.H. Jo, J.H. Kim, J.S. Lee, Ultrafast fabrication of highly active BiVO₄ photoanodes by hybrid microwave annealing for unbiased solar water splitting, *Nanoscale*, 8 (2016) 17623-17631.

[42] X. Zhao, W. Luo, J. Feng, M. Li, Z. Li, T. Yu, Z. Zou, Quantitative Analysis and Visualized Evidence for High Charge Separation Efficiency in a Solid-Liquid Bulk Heterojunction, *Advanced Energy Materials*, 4 (2014) 1301785-n/a.

- [43] S.J. Hong, S. Lee, J.S. Jang, J.S. Lee, Heterojunction BiVO₄/WO₃ electrodes for enhanced photoactivity of water oxidation, *Energy & Environmental Science*, 4 (2011) 1781-1787.
- [44] M. Alexander, K. Ilina, F. Alena, F.-R. Dina, B. Thomas, S. Christina, Dual absorber Fe₂O₃ /WO₃ host-guest architectures for improved charge generation and transfer in photoelectrochemical applications, *Materials Research Express*, 4 (2017) 016409.
- [45] B. Klahr, S. Gimenez, F. Fabregat-Santiago, J. Bisquert, T.W. Hamann, Photoelectrochemical and Impedance Spectroscopic Investigation of Water Oxidation with “Co–Pi”-Coated Hematite Electrodes, *Journal of the American Chemical Society*, 134 (2012) 16693-16700.
- [46] L. Bertoluzzi, P. Lopez-Varo, J.A. Jimenez Tejada, J. Bisquert, Charge transfer processes at the semiconductor/electrolyte interface for solar fuel production: insight from impedance spectroscopy, *Journal of Materials Chemistry A*, 4 (2016) 2873-2879.
- [47] I. Grigioni, K.G. Stamplecoskie, E. Selli, P.V. Kamat, Dynamics of Photogenerated Charge Carriers in WO₃/BiVO₄ Heterojunction Photoanodes, *The Journal of Physical Chemistry C*, 119 (2015) 20792-20800.
- [48] S. Hosseini, E. Eftekhari, S.M. Soltani, F.E. Babadi, L.J. Minggu, M.H.S. Ismail, Synthesis, characterization and performance evaluation of three-layered photoanodes by introducing a blend of WO₃ and Fe₂O₃ for dye degradation, *Applied Surface Science*, 289 (2014) 53-61.
- [49] Enhanced Water Oxidation Photoactivity of Nano-Architected α -Fe₂O₃–WO₃ Composite Synthesized by Single-Step Hydrothermal Method, *Journal of Electronic Materials*, (2018).
- [50] Y. Li, L. Zhang, R. Liu, Z. Cao, X. Sun, X. Liu, J. Luo, WO₃@ α -Fe₂O₃ Heterojunction Arrays with Improved Photoelectrochemical Behavior for Neutral pH Water Splitting,

ChemCatChem, 8 (2016) 2765-2770.

[51] J. Su, L. Guo, N. Bao, C.A. Grimes, Nanostructured WO₃/BiVO₄ Heterojunction Films for Efficient Photoelectrochemical Water Splitting, Nano Letters, 11 (2011) 1928-1933.

[52] Y. Pihosh, I. Turkevych, K. Mawatari, J. Uemura, Y. Kazoe, S. Kosar, K. Makita, T. Sugaya, T. Matsui, D. Fujita, M. Tosa, M. Kondo, T. Kitamori, Photocatalytic generation of hydrogen by core-shell WO₃/BiVO₄ nanorods with ultimate water splitting efficiency, Scientific Reports, 5 (2015) 11141.

[53] S.K. Pilli, R. Janarthanan, T.G. Deutsch, T.E. Furtak, L.D. Brown, J.A. Turner, A.M. Herring, Efficient photoelectrochemical water oxidation over cobalt-phosphate (Co-Pi) catalyst modified BiVO₄/1D-WO₃ heterojunction electrodes, Physical Chemistry Chemical Physics, 15 (2013) 14723-14728.

Figures legends

Figure 1. Schematic illustration of Fe_2O_3 or BiVO_4 semiconductor film synthesising (a) directly onto substrate and (b) WO_3 fibrous pre-coated substrate by spin coating.

Figure 2. (a) XRD result of WO_3 fiber, Fe_2O_3 and BiVO_4 films in the presence and absence of WO_3 fibers. Note that the crystalline contribution from FTO substrate is indicated in * symbol.

Figure 3. SEM images of (a) Fe_2O_3 thin film, (b) BiVO_4 thin film, (c) WO_3 nanofibers, (d) Fe_2O_3 coated WO_3 nanofiber, and (e) BiVO_4 coated WO_3 nanofiber coated on FTO glass substrates.

Figure 4. HRTEM images of $\text{BiVO}_4/\text{WO}_3$ sample (a) at 50 nm scale and (b) at 10 nm scale.

Figure 5. (a) Optical absorbance spectra of Fe_2O_3 and BiVO_4 films in the presence and absence of WO_3 fiber, (b) Optical reflectance spectra of $\text{Fe}_2\text{O}_3/\text{WO}_3$ and $\text{BiVO}_4/\text{WO}_3$ composite fibers compared with WO_3 fiber film (note that films are coated onto FTO substrate).

Figure 6. Nyquist plots of PEC cells consisted with different photoanodes measured at (a) under dark condition and (b) Light irradiation condition (100 mWcm^{-2}). Note that the measurements were carried out at operating potential 0.7 V vs Ag/AgCl

Figure 7. (a) Proposed energetic structure of $\text{Fe}_2\text{O}_3/\text{WO}_3$ and $\text{BiVO}_4/\text{WO}_3$ heterointerfaces at photoelectrocatalytic water oxidation performance and (b) Chronoamperometry plots of PEC cell with different photoanodes (measurements carried out at applied potential 0.7 V vs Ag/AgCl). Note that 0.5 M of aqueous Na_2SO_4 is used as electrolyte. The photocurrent measured under light irradiation is obtained at light intensity AM 1.5 (100 mWcm^{-2}).

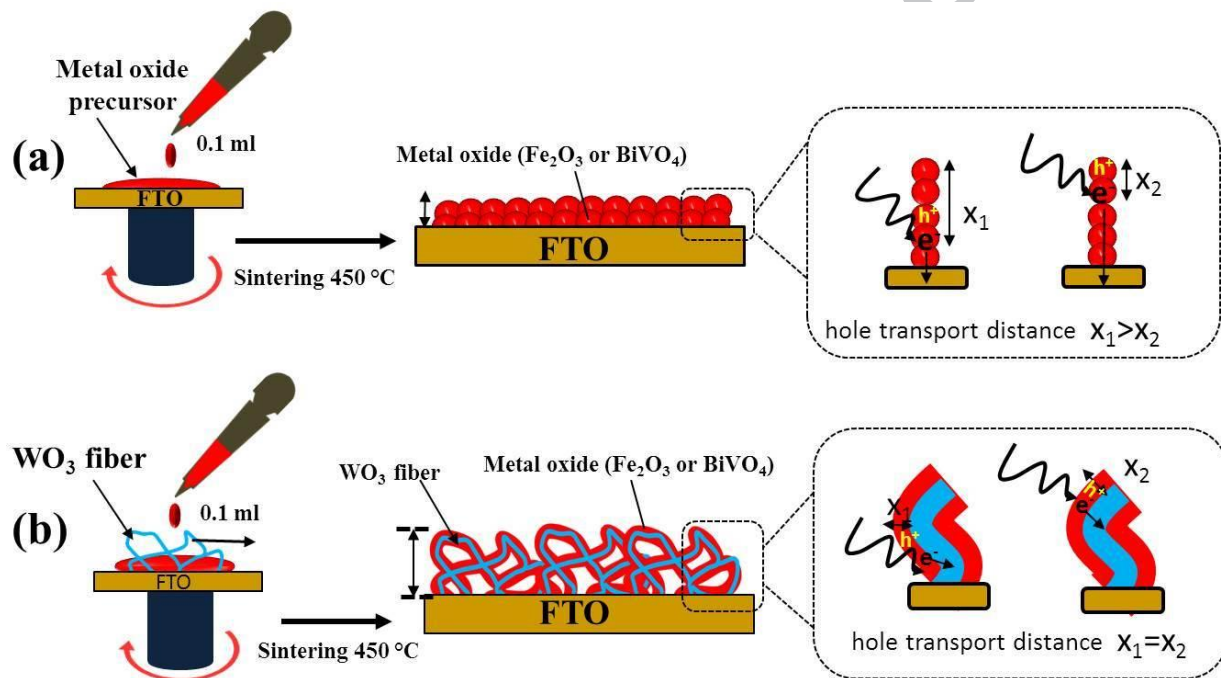


Figure 1.

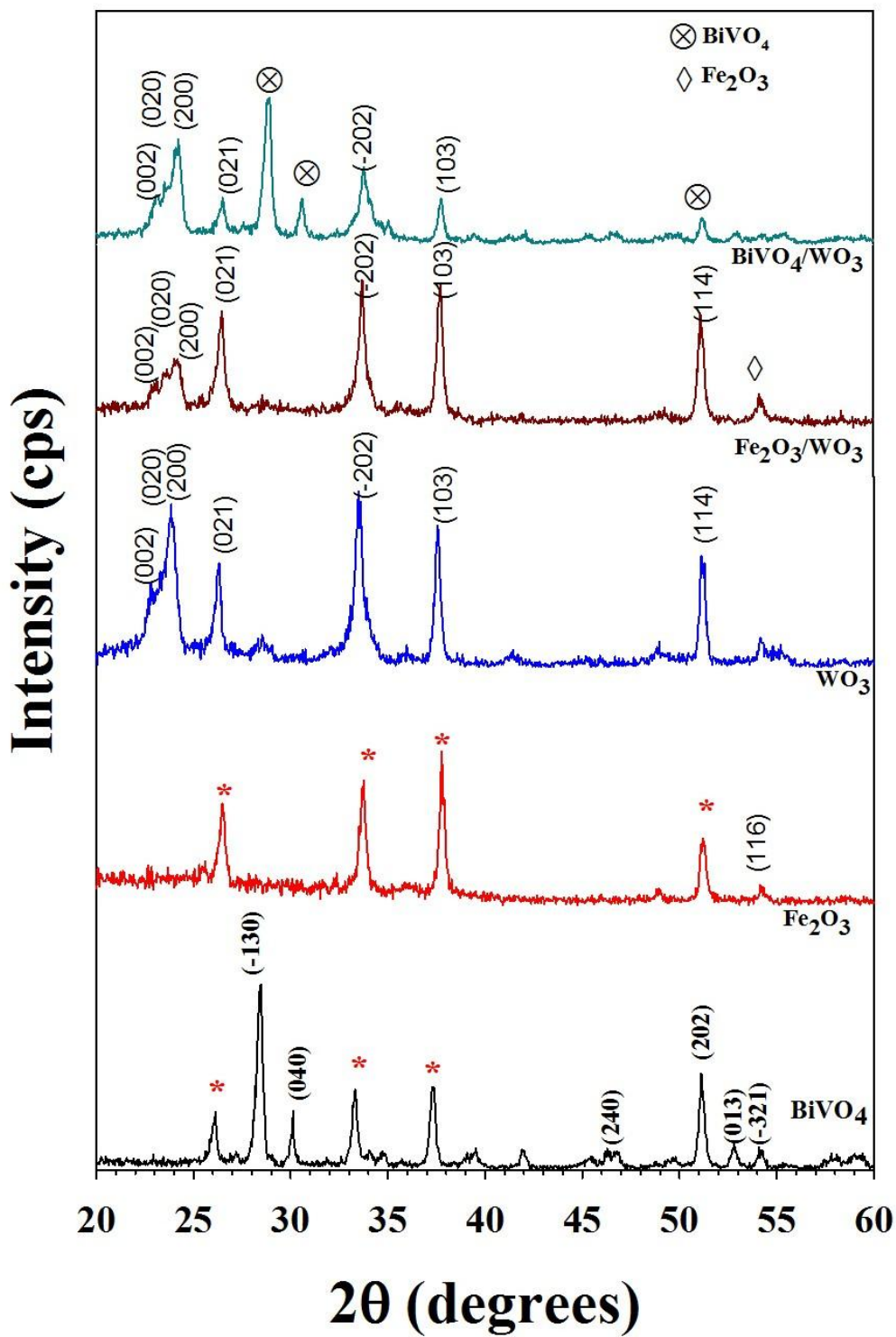


Figure 2.

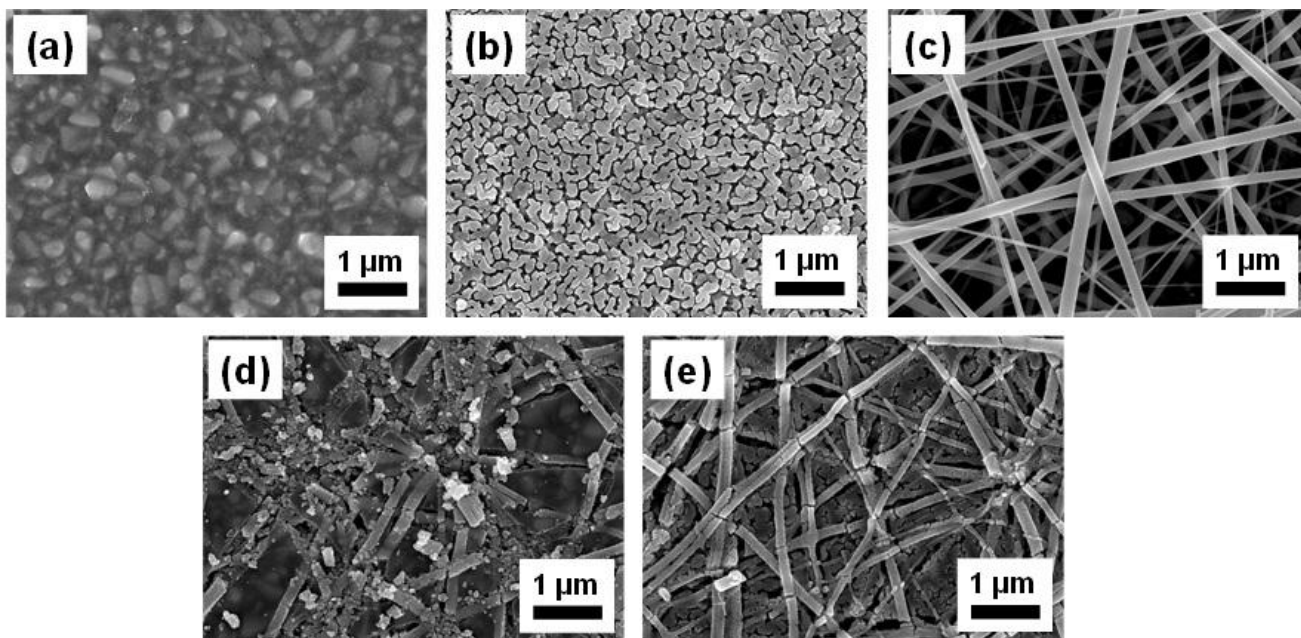


Figure 3.

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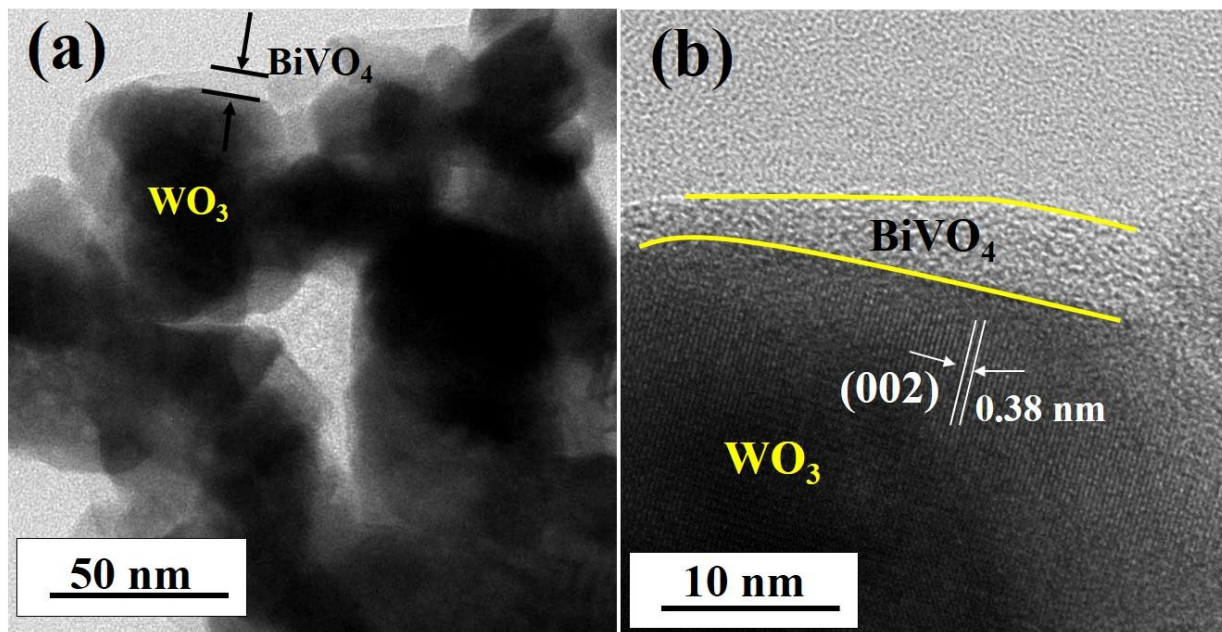


Figure 4

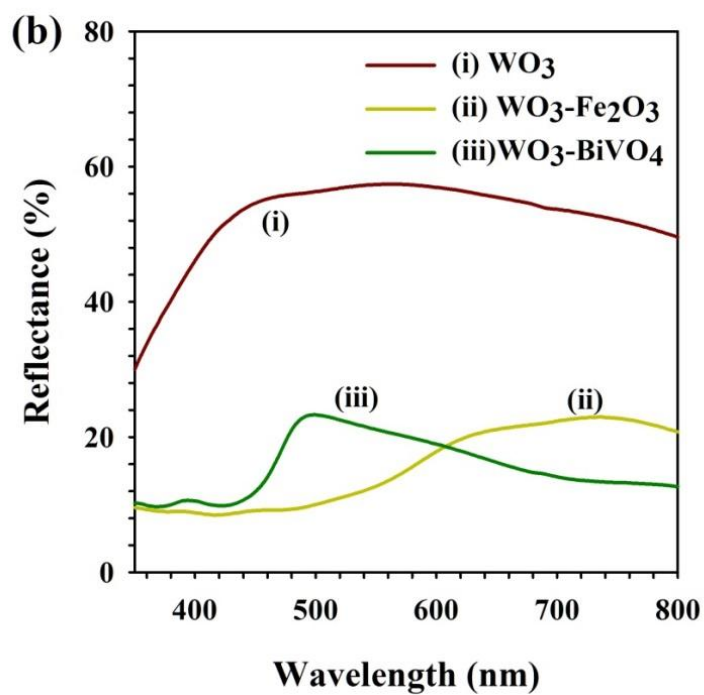
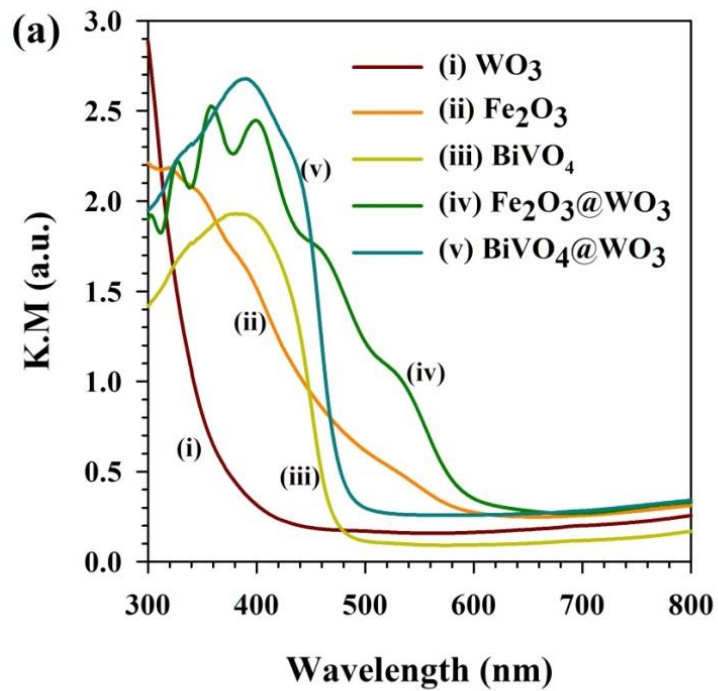


Figure 5.

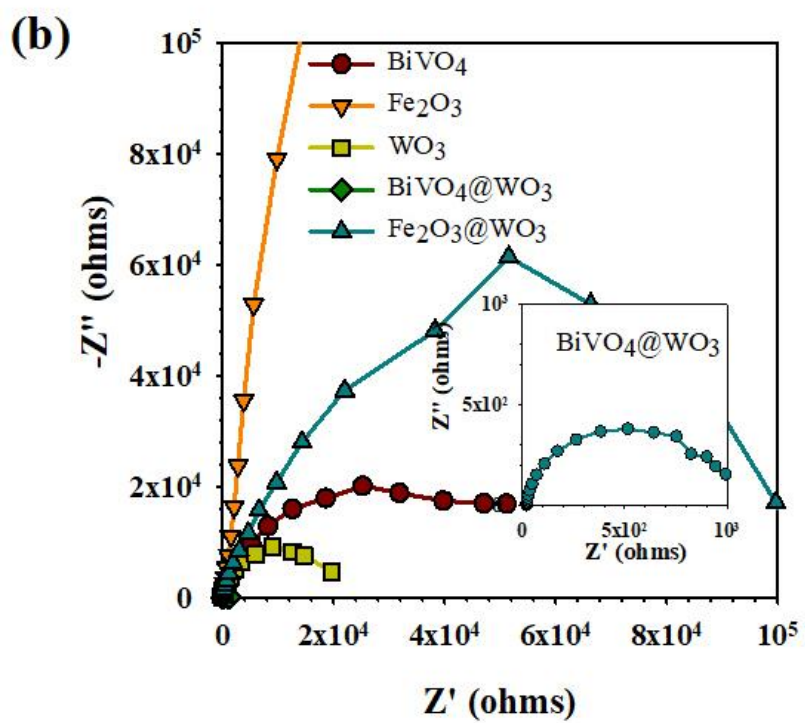
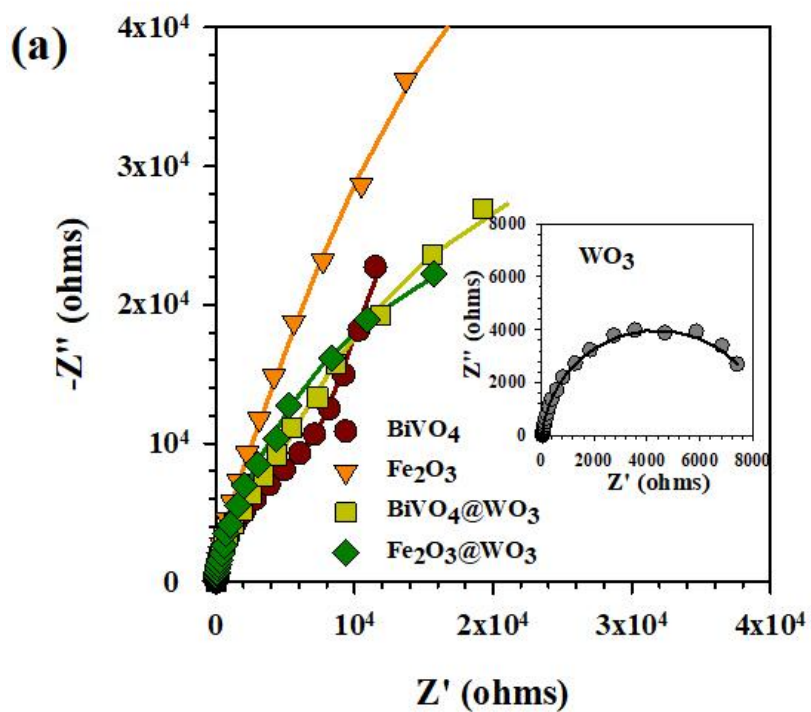


Figure 6

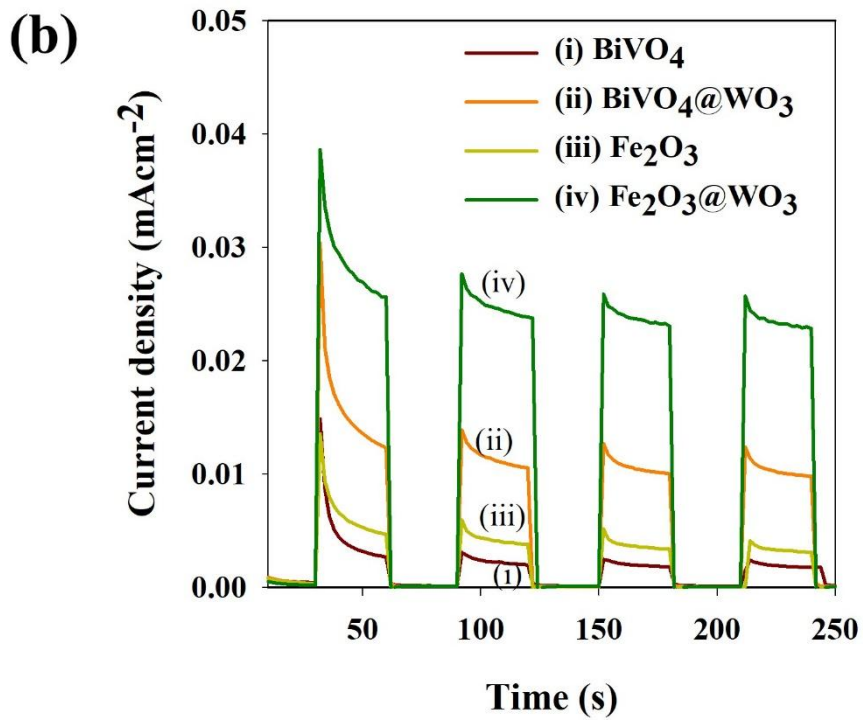
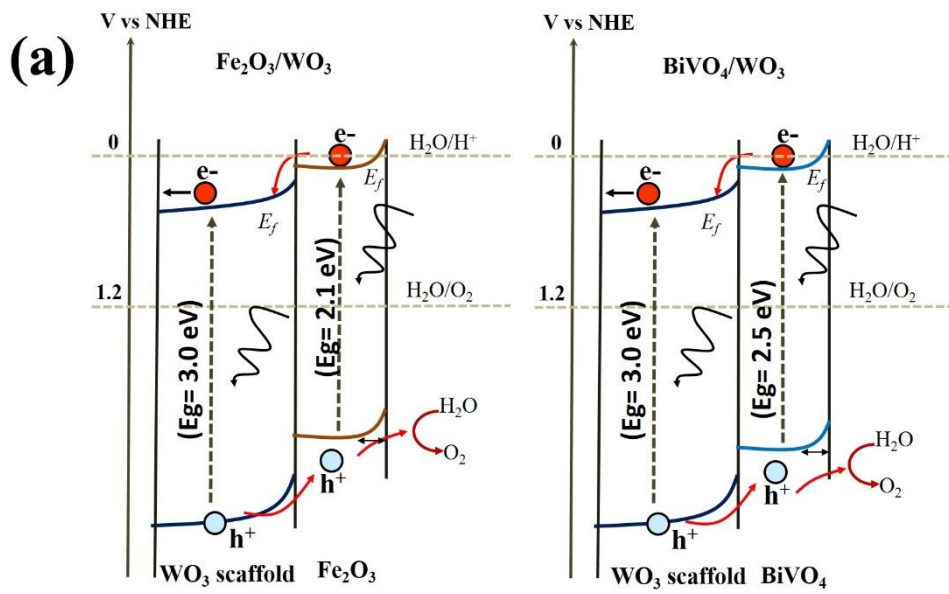
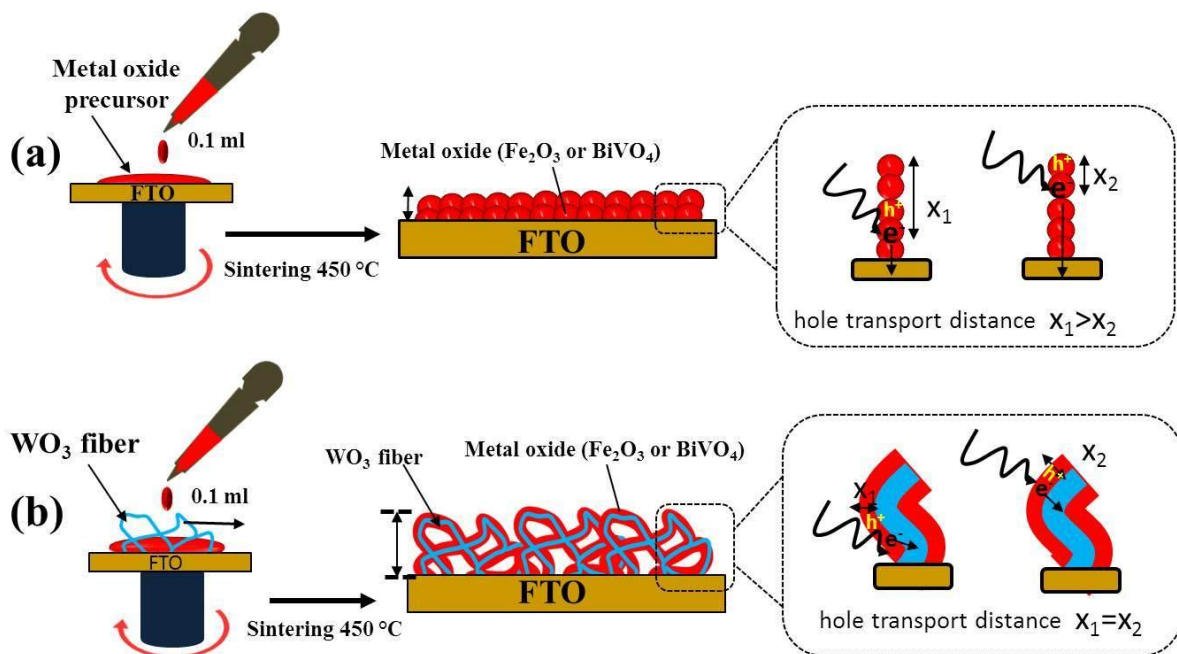


Figure 7

Highlights

- Electrospinning derived WO_3 fibrous is demonstrated as backbone scaffold in light driven photoanode.
- Synthesis route of sol-gel assisted spin coated Fe_2O_3 and BiVO_4 thin films onto WO_3 fibers is demonstrated.
- High visible light harvesting is achieved at Fe_2O_3 and BiVO_4 through introducing WO_3 nanofibrous backbone scaffold.
- Charge transport and charge separation at $\text{Fe}_2\text{O}_3/\text{WO}_3$ and $\text{BiVO}_4/\text{WO}_3$ based heterostructured photoanode/electrolyte interfaces is promoted.
- Underlying mechanism of one dimensional WO_3 fibrous backbone scaffold assisted photoanode in solar fuel generation is examined.

Graphical abstract



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